

**Isolation and Characterization of Certain Biomolecules
from *Euphorbia caducifolia* Haines**



**Thesis submitted to the University of Hyderabad for the degree of
DOCTOR OF PHILOSOPHY**

By

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DECLARATION

I, **Kusuma Venumadhav**, hereby declare that the thesis entitled “**Isolation and Characterization of Certain Biomolecules from *Euphorbia caducifolia* Haines**” submitted to University of Hyderabad has been carried out by me under the supervision of **Prof. Kottapalli Seshagirirao**, Department of Plant Sciences, School of Life Sciences, University of Hyderabad, Hyderabad – 500046, India. I also declare that the work has not been submitted for any other degree or diploma to any University or Institution and is free from plagiarism.

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CERTIFICATE

This is to certify that this thesis entitled “**Isolation and Characterization of Certain Biomolecules from *Euphorbia caducifolia* Haines**” is a record of Bonafide work done by **Kusuma Venumadhav**, a research scholar for Ph. D. programme under the Department of Plant Sciences, School of Life Sciences, University of Hyderabad under my guidance and supervision. This thesis is free from plagiarism and has not been submitted in part or in full to this or any other University or Institution for the award of any degree or diploma.

Parts of the thesis have been:

A. Published in the following publications:

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Further, the student has passed the following courses towards the fulfillment of the coursework requirement for Ph.D.

| Sl.No. | Course Code | Name | Credits | Pass/Fail |
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| 2. | PL 802 | Research Ethics & Management | 2 | Pass |
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| 4. | PL 801 | Biostatistics | 2 | Pass |

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Kusuma Venuamdahv

ABBREVIATIONS

| | |
|--------|---|
| µg | micro gram |
| µl | microliter |
| ATR-IR | Attenuated transmission resonance Infra-red |
| BHT | Butylated Hydroxy Toulene |
| BSA | Bovine serum albumin |
| CD | Circular dichroism |
| CLGG | Cross linked guar gum |
| CLLSG | Cross linked leucaena seed gum |
| cm | Centimeter |
| db | Dry basis |
| DEAE | Diethylaminoethyl |
| DMSO | Dimethyl sulfoxide |
| DPPH | 1,1-Diphenyl-2-picryldydrazyl |
| DSC | Differential scanning calorimetry |
| ECP | Euphorbia caducifolia polysaccharide |
| g | Gravitational force |
| Gal | Galactose |
| GC-MS | Gas chromatography coupled with mass spectroscopy |
| GlcNAc | N-acetyl Glucosamine |
| Glu | D-glucose |
| gm | Gram |
| HCl | Hydrogen chloride |
| HRBC | Human red blood cells |
| IL-6 | Interleukin-6 |
| kDa | kilo Dalton |
| M | Molarity |
| Man | Mannose |
| MIC | Minimum inhibitory concentration |
| Min | Minute |
| ml | milliliter |
| mM | milli molar |
| MS | Mass spectrometry |

| | |
|--------------|---|
| MTT | 3-5,5-dimethylthiazol-2-yl-2,5-diphenyltetrazolium bromide |
| MW | Molecular weight |
| N | Normality |
| ng | Nano gram |
| NaCl | Sodium chloride |
| NaOH | Sodium Hydroxide |
| PAGE | Polyacrylamide gel electrophoresis |
| PBS | Phosphate buffered saline |
| pH | -log(H ⁺) concentration |
| PMF | Peptide mass fingerprinting |
| ROS | Reactive oxygen species |
| RPM | Rotations per minute |
| RT | Room temperature |
| SDS | Sodium dodecyl sulphate |
| SEM | Scanning electron microscope |
| TCA | Trichloroacetic acid |
| TEMED | N’N’N’N’ tetra methyl ethylene diamine |
| TFA | Trifluoro Acetic acid |
| TEM | Transmission Electron Microscopy |
| TG | Thermo gravity |
| TNF α | Tumor necrosis factor alpha |
| Tris | tris(hydroxymethyl)aminoethane |
| UV | Ultra violet |
| W/V | Weight/Volume |
| XRD | X-ray diffraction |

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CHAPTER I

INTRODUCTION

Natural products are the substances produced by living organisms. Secondary metabolites gained attention as natural products where primary and secondary metabolites are considered to be natural products by organic chemists. Secondary metabolites have been leading the drug discovery in early stages of Natural product chemistry (Harvey, 2008). From the discovery of an antibiotic “Penicillin” in 1928, the “Golden Age of Antibiotics” started but later on the interest in the drug discovery from natural products has been reduced and the novel lead compounds are producing with the help of combinational chemistry. However, Combinational Chemistry gives low molecular structural analogues of the natural products (Butler, 2004; Mishra & Tiwari, 2011; Lewis, 2012). David J. Newmann reported 1328 only New chemical entities from 1981 to 2014 in various categories (Fig. 1.1) (Newman & Cragg, 2016). Natural products are still a source of novel drugs and plants have been a major contributor of such products.

These are products derived from nature includes micro-organisms, plants, and animals part of terrestrial and marine sources. The microbial world produced active compounds like antibiotics, anticancer, antidiabetic and other after the exploration of penicillin. Major natural products from the animal sources are “venoms and toxins” apart from other small contributions. Marine environment places the second role after the plant in producing Novel Natural Products. Marine region covers 70% of the earth with unique biodiversity (Haefner, 2003). Algae, sponges, tunicates, ascidians, and bryozoans

are the marine organisms explored with potential drugs (Mayer et al., 2010). Plants are the greater source of natural products.

Life has relied on nature for its basic needs mostly on plants, majorly for food and medicine. The plants are the core source of the creation of the traditional medicine system all over the world (Cragg & Newman, 2013). History of Medicine says the role of plants in treating infections and diseases. Natural products from plants always remain as a valuable source contributing a revolutionary green approach to the discovery of novel compounds (Table 1.1). Around 70,000 plants were screened for their potential activity till now which are using in different medical systems across the world. In the global market, Europe contributed a significant number of natural products (Fig. 1.2) (Dev, 1999).

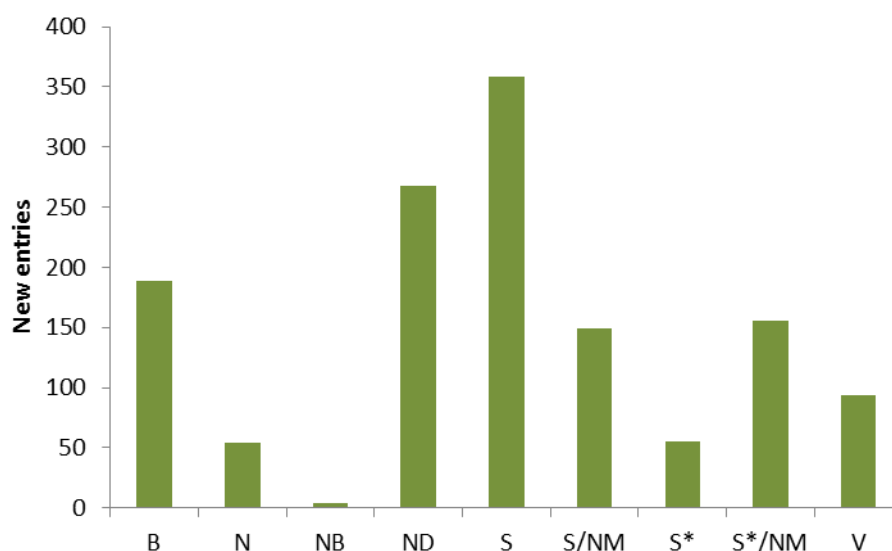


Fig. 1.1. New chemical entries of drugs from 1981 to 2014

B-Biological; N-Natural products; NB-Botanical Natural product; ND-Derived Natural product; S-Synthetic; S/NM-Natural product mimic; S*-Synthetic with natural pharmacore; V-Vaccine

Table 1.1. Some examples of commonly available Plant product

| Category | Plant name | Therapeutic uses |
|---|--|-----------------------------------|
| Diosgenin, Hecogenin, Stigmasterol | <i>Dioscorea spp.</i> | Oral contraceptives |
| Digoxin, Digltoxin | <i>Digitalis purpurea</i> L, <i>D. lanata</i> | Cardiotonic |
| Tropane Alkaloids Atropine Ahyoscyamine Scopolamine | <i>Atropa belladonna</i> L, <i>Datura spp.</i> <i>Hyoscyamus niger</i> L, <i>Mandragora officinarum</i> L | Anticholinergics |
| Opium Alkaloids Codeine Morphine | <i>Papaver somniferum</i> L | Analgesics |
| | | Antitussive |
| | | |
| Reserpine | <i>Rauwolfia serpentina</i> (L.) Bentham ex Kurz | psychotropic, Antihypertensive |
| Vinca Alkaloids (Catharanthus) Vinblastine Vincristine | <i>Catharanthus roseus</i> (L) G. Don | Anticancer |
| Physostigmine | <i>Physostigma venenosum</i> Balfour | Cholinergic |
| Pilocarpine | <i>Pilocarpus jaborandi</i> Holmes | |
| Cinchona alkaloids Quinine Quinidine | <i>Chincona spp.</i> | Antimalarial, antiarrhythmic |
| Colchicine | <i>Colchicum autumnale</i> L. | Antigout |
| Cocaine | <i>Erythroxylum coca</i> Lamarck | Local anesthetic |
| d-Tubocurarine | <i>Strychnos toxifera</i> Bentham | Skeletal muscle relaxant |
| Taxol | <i>Taxus brevifolia</i> Nutt. | Anticancer |

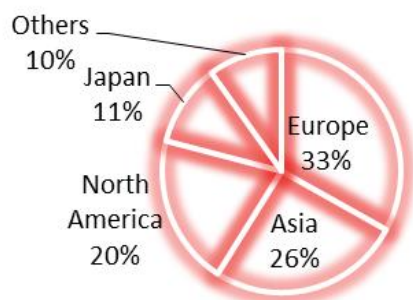


Fig. 1.2. Contribution of natural products in the market

In India, plants have been used for many therapeutic applications by our traditional system of medicines such as Ayurveda, Unani, Homeopathy, and Siddha. Traditional medicine system gave birth to many active compounds in early years of natural chemistry, but reports of natural products from different sources has been decreased rapidly in last decade and looking for new active compounds. According to National Medicinal Plant Board (NMPB), 6000 species are using in different medical systems and a maximum number of plants using in the folk medicine compared to other documented medicine systems (Fig. 1.3)(FRLTH INDIA. 2017). Now it needs to explore new natural compounds from potential folk medicinal plants and ethnomedicinal plants for better therapeutic applications. Detailed use plants in the medical systems documented for the use of future reference along with identification. In Ayurveda, Charaka Samhita is the guide for the plant formulations. Like, modern medicine attained documentation regarding the plant materials and their identifications along with medical importance called Pharmacognosy.

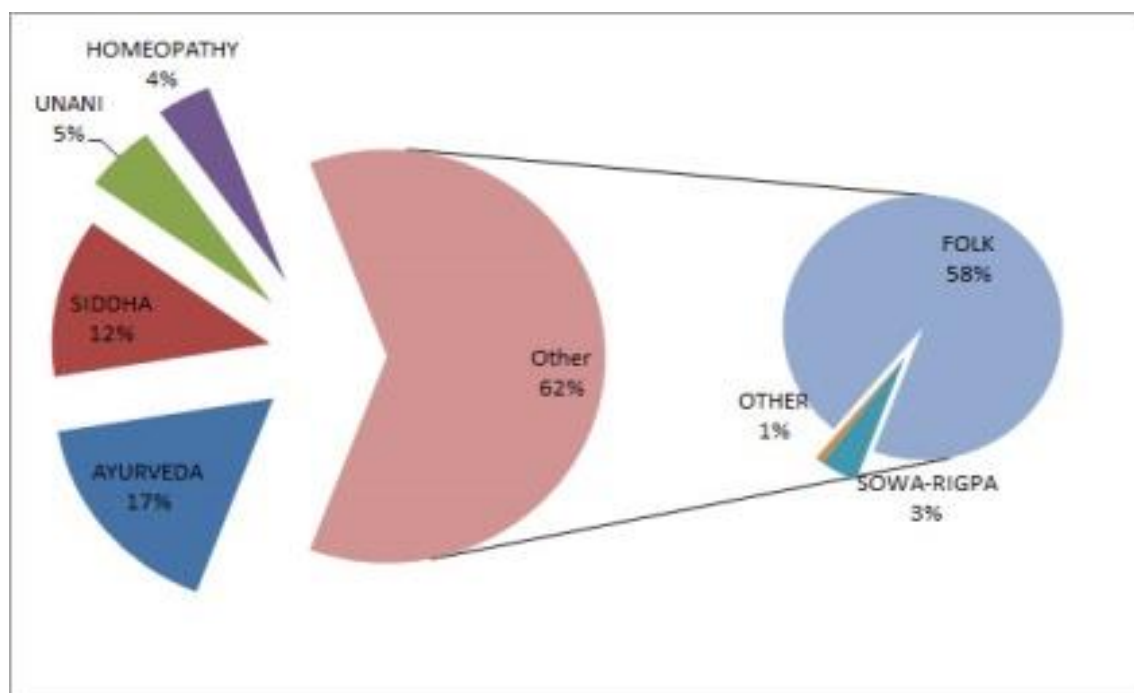


Fig. 1.3. Distribution of medicinal plants in Indian Medicine System

Pharmacognosy is a study of crude drugs or knowledge of crude drugs on chemical as well as structural characterization and also metabolic pathways leading to pharmacologically active compounds. The horizon of pharmacognosy has enlarged from classic morphological characters to the modern study of bioactive compounds with their mechanism of action and therapeutic uses (Kato & Pezzuto, 2017). Pharmacognosy gained attention due to, search for new leads for drug development, biotechnology for the production of pharmaceuticals and nutraceuticals, increased interest on phytotherapy and validation of traditional medicines. Apart from the study of secondary metabolites, Primary metabolites and their small molecules advanced for the therapeutics. Primary metabolites include polysaccharides, proteins, nucleic acids, fatty acids and secondary metabolites majorly includes alkaloids, flavonoids, terpenoids, glycosides, saponins, phenolic compounds, anthraquinones, tannins and others.

Polysaccharides

Polysaccharides are biological macromolecules with the repetitive arrangement of monosaccharides as basic units. These are one of 4 major biological macromolecules along with proteins, nucleic acids, and fatty acids. Polysaccharides can serve as an energy source and structural components to the cell with its highly complex structure (Z. Wang, Luo, & Ena, 2007). Cellulose, Chitin, and starch are the abundant polysaccharides in the earth. Applications of polysaccharides on industrial orientation are enormous. Majorly they are as thickening agents, chelating agents, emulsifying agents, stabilizing agents, suspending agents, flocculating agents and also used to form gels and membranes. Polysaccharides are biocompatible and biodegradable in nature along with bioavailability. Gums and their derivatives are considered as a major class of the polysaccharides as part of industrial applications (Lapasin & Pricl, 1995). Marine origin polysaccharides are gaining interests along with microbial polysaccharides. Agar,

Agarose, alginates, and carrageenan are the seaweed polysaccharides employing in the various fields (Laurienzo, 2010). Exopolysaccharides of the microorganisms like xanthan, dextrans are currently using the market in addition to many novel immunomodulatory polysaccharides (Sutherland, 1998).

Botanical polysaccharides

Polysaccharides from higher plants show significant therapeutic properties by its non-toxic nature and ability to not cause any side effects compared to bacterial and fungal polysaccharides. Thus, botanical polysaccharides are ideal therapeutics for immunomodulatory, antitumor and wound healing activities (Schepetkin & Quinn, 2006). Apart from other activities, most polysaccharides are considering for immunomodulatory activity as they are the non-immune origin. Botanical polysaccharides affect the macrophages and stimulate the production of cytokines, ROS, chemokines and along with cell proliferation (Fig. 1.4). Plant polysaccharides from the different plant species have shown immunomodulatory activity on various types of macrophages.

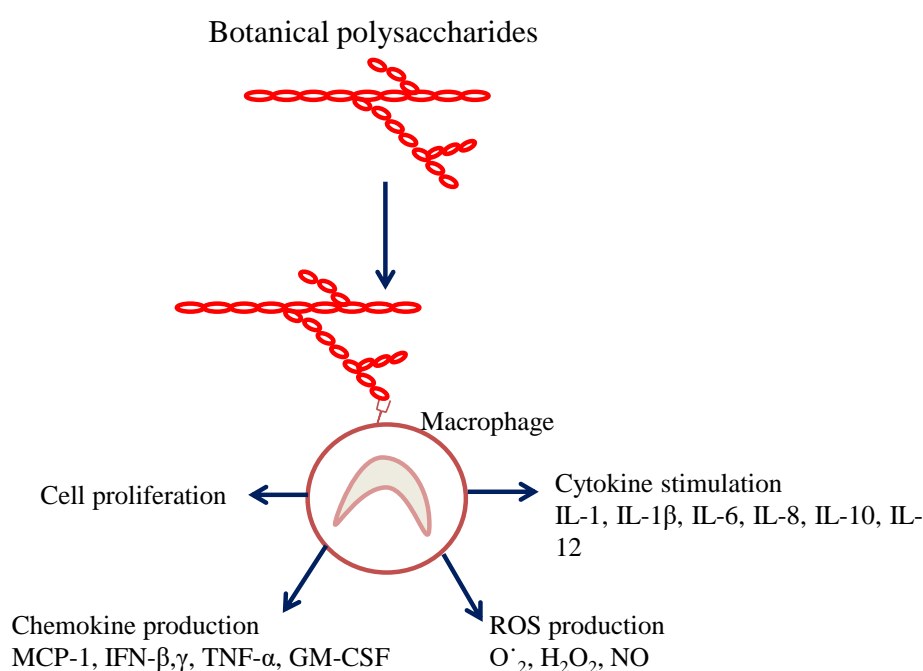


Fig. 1.4. Action of botanical polysaccharides on Macrophages

Lectins

“Lectin is a sugar-binding protein or glycoprotein of non-immune origin which agglutinates cells and/or precipitates glycoconjugates” (Goldstein et al., 1980). In 1888 Stillmark demonstrated a protein which agglutinates the RBC called “Ricin” from *Ricinus communis*. Later noticed the “Abrin” in *Abrus precatorius* by H.Hellin in 1891. Karl Landsteiner in 1900 discovered the ABO blood group using lectin. Lectin “Concanavalin A” crystallized and revealed the sugar specificity of lectins by inhibiting the haemagglutination.

Lectin Classes

Lectins were classified on sugar specific nature by Goldstein and poretz in 1986. Classes are based binding of the lectin to the sugars mannose or glucose, galactose or N-acetyl galactosamine, N-acetyl glucosamine, fucose and sialic acid (Table 1.2). Further, based on their overall structure and subunits, lectins are classified as Merolectins, Hololectins, Super lectins and Chimerolectins. They are grouped on evolutionary relations into different families, i.e., Chitin binding lectins, Curcubitaceae phloem lectins, Amaranthin lectins, Monocot mannose binding lectins, Legume lectins, Jacalin related lectins and Ribosome inactivating protein type 2 lectins (Hamid et al., 2013) (Table.1.2)

Table 1.2 Classification of lectins on sugar basis with some examples

| Class/Binding Lectins | Examples | Predominant plant families |
|------------------------------------|---|---------------------------------------|
| Mannose | Concanavalin A, Lentil lectin, Snowdrop lectin | Fabaceae |
| Galactose / N-Acetyl Galactosamine | Ricin, Peanut agglutinin, Jacalin | Euphorbiaceae, Fabaceae, Loranthaceae |
| N-acetylglucosamine | Wheat germ agglutinin | Poaceae, Solanaceae, Fabaceae |
| N-acetylneuraminic acid | Elderberry lectin, Maackia amurensis hemoagglutinin and leucoagglutinin | Nelumbonaceae, Fabaceae |
| Fucose | Ulex europaeus agglutinin, Aleuria aurantia lectin | Fabaceae |

Table 1.3 Lectins from *Euphorbia* genus

| Plant Name | Part | Sugar Specificity | Authors | Year |
|--------------------------------------|-----------------|-------------------------|-------------------------------------|------|
| <i>Euphorbia heterophylla</i> | Seeds | N-acetyl galactosamine | NsimbaLubaki et.al | 1983 |
| <i>Euphorbia charcias</i> | Latex | Galactose | Babieri et al. | 1983 |
| <i>Euphorbia calcina L</i> | Latex | N-acetyl galactosamine | NsimbaLubaki et al. | 1986 |
| <i>Euphorbia dalberi L</i> | Latex | N- acetyl galactosamine | NsimbaLubaki et al. | 1986 |
| <i>Euphorbia coerulscens</i> | Latex | Galactose | Lynn & Clevette-Radford | 1986 |
| <i>Euphorbia hermentiana</i> | Latex | Galactose | Lynn & Clevette-Radford | 1986 |
| <i>Euphorbia escula</i> | Latex | Galactose | Lynn & Clevette-Radford | 1986 |
| <i>Euphorbia globose</i> | Latex | Galactose | Lynn & Clevette-Radford Lynn et al. | 1986 |
| <i>Euphorbia lactea</i> | Latex | Galactose | Lynn & Clevette-Radford Lynn et al. | 1986 |
| <i>Euphorbia lactea cristata</i> | Latex | Galactose | Lynn & Clevette-Radford Lynn et al. | 1986 |
| <i>Euphorbia lathyris</i> | Latex | Galactose | Lynn & Clevette-Radford et al. | 1986 |
| <i>Euphorbia mammillaris</i> | Latex | Galactose | Lynn & Clevette-Radford et al. | 1986 |
| <i>Euphorbia splendens</i> | Latex | Galactose | Lynn & Clevette-Radford et al. | 1986 |
| <i>Euphorbia stapilioides</i> | Latex | Galactose | Radford et al. | 1986 |
| <i>Euphorbia marginata</i> | Latex | Galactose | Stirpe et al. | 1993 |
| <i>Euphorbia neriifolia</i> | Latex | Galatose | Seshagirirao & Prasad | 1995 |
| <i>Euphorbia milli</i> | Latex | Galactose | Seshagirirao | 1999 |
| <i>Euphorbia milii, var. milii</i> | Latex | GalNac | M. Dias-Baruffi et al. | 2000 |
| <i>Euphorbia antiquorum</i> | Latex | Galactose | Seshagirirao & Prasad | 2001 |
| <i>Euphorbia nivulia</i> | Latex | Galactose | Seshagirirao | 2001 |
| <i>Euphorbia milii var. hislopii</i> | Latex | Galactose | K.C. Fonseca et al. | 2010 |
| <i>Euphorbia helioscopia</i> | Leaves Whole | Fructose | Rafiq et al. | 2014 |
| <i>Euphorbia tirucalli</i> | plant | Mannose | Santana et al. | 2014 |
| <i>Euphorbia trigona</i> | Latex | Galactose | Villanueva J et al. | 2015 |
| <i>Euphorbia tithymaloides</i> | Leaves Whole | Galactose | Jawade et al. | 2016 |
| <i>Euphorbia pulcherima</i> | plant | Galactose | Torky et al. | 2016 |

Lectins from *Euphorbia*

Latex, a colloidal suspension of laticifer cells flows with cellular components, nuclei, mitochondria, amyloplast, rubber, sterols, alkaloid, carbohydrates, proteins, free amino acids, phenolics (K. Seshagirirao & Prasad, 2001). Euphorbiaceae is also one among the 20 families of latex-producing plants. Biochemical as well as biosystematics studies have been investigated using Euphorbiaceae lattices. Apart from screening of metabolites, proteins also present in the latex. Lennox and Ellis (1945) reported a protease “Euphorbin” in the latex of *Euphorbia lathyris*. 18 proteases were isolated from different *Euphorbia* species by Lynn and Clevette-Radford (1984, 1985, 1988). In 1983, Barbieri et al., purified galactose-specific lectins from the latex of *Euphorbia characias* and *Hura crepitans*. Most of the lectin from the latex of the genus *Euphorbia* has been reported galactose-specific in nature (Table. 1.3). Lectins are using for therapeutics, pharmaceutical and food industries.

Euphorbia L. Genus

Euphorbia genus belongs to Euphorbiaceae family and cosmopolitan distribution. This family is one of the largest family in angiosperms and consist of over 300 genera, 37 tribes, three subfamilies and 7500 species (Webster, 1994; Mwine & Van Damme, 2011). The diversity of unusual secondary metabolites in the family is accountable for the poison nature of some members. Abdel-Fattah in 1987 attempted for classification of family poisons. Listed as:

| Action of Poison on | Examples |
|---------------------|--|
| Fish | <i>E. scheffleri</i> Pax, <i>E. tirucalli</i> L., <i>E. inaequilatera</i> Sond <i>E. ledienii</i> A.Berger, <i>E. heterophylla</i> L., <i>E. cooperi</i> N.E.Br. ex A. Berger, |
| Human | <i>E. candelabrum</i> Kotschy, <i>E. virosa</i> Willd., <i>E. poissonii</i> Pax, <i>E. unispina</i> N.E.Br. <i>E. venenifica</i> Tremaux ex Kotschy |
| Domestic animals | <i>E. caput-medusae</i> L., <i>E. silenifolia</i> Sweet <i>E. ingens</i> E.Mey. ex Boiss |

Euphorbia L is the largest genus of the family with almost 2000 species. Members of the genus distributed in temperate, tropical and subtropical regions and range from herbs to trees. Despite poison nature and irritant ability, plants are greatly used in traditional medicine. For example, *E. hirta*, *E. heterophylla* as antibacterial (Chika C et.al., 2007; Falodun et.al., 2008); *E. kansui*, *E. hyberna*, *E. tirucalli*, *E. thymifolia* as antiviral (Bedoya et al., 2009; Betancur-Galvis et al., 2002; Lin et al., 2016.; Zheng et al., 1998); *E. nivulia* Buch.-Ham. as an antifungal (Annapurna et al., 2004); *E. helioscopia*, *E. pulcherrima* as nematocidal (Devi & Pamila, 2000); *E. splendens*, *E. conspicua* N.E. Br.as Molluscicidal (De Oliveira Dias, Dos Santos Machado, Migliolo, & Franco, 2015; Mauricio Carvalho de Vasconcellos, 2003); *E. hirta* as insecticidal (Murakami et al., 2007); *Euphorbia* genus limited to *E. ficheriana* and *E. kansui* for polysaccharide extraction among a large number of species.

***Euphorbia caducifolia* Haines**

Euphorbia caducifolia Haines succulent milk hedge (Fig. 1.5), a folk medicinal plant belongs to Euphorbiaceae. The term “caducifolia” means falling leaf, leaves of the plants are caducus in nature. *E. caducifolia* is a latex-producing plant in *Euphorbia* genus and found in rocky areas of the tropical region. The plant latex is used for wound healing, leucoderma, and skin eruptions. Latex and root of the plant used to treat cancer. Root also used for snakebite in Maharashtra. Leaf extracts of the plant were reported for its antimicrobial activity. Apart, there are not many reports on phylloclade of the plant which is a rich source of biofuel and other compounds. Though the plant has been using for many medical applications but limited to little exploitation (Table.1.4). Looking the medicinal importance of the plant, it was considered worthwhile to evaluate the plant for the biological activity scientifically.



Fig. 1.5. Distribution of *E. caducifolia* Haines in the wild (Bhongiri area, Yadadri dist. Telangana)

Table 1.4. List of publications on *E. caducifolia*

| Part | Study | Authors | Year |
|-------------|------------------------------|-------------------------|-------------|
| Plant | Autecological Study | Hussain et al. | 1970 |
| Plant | CAM metabolism | Sen et al. | 1970 |
| Plant | Ecology and Seedlings | Sen et al. | 1970 |
| Latex | Triterpenes isolation | Govardhan et al. | 1984 |
| Latex | Latex Biochemistry | Lynn & Clevette-Radford | 1986 |
| Latex | Antiparasitic Activity | Maqbool et al. | 1987 |
| Phylloclade | Milk clotting Activity | Umar et al. | 1990 |
| Latex | Latex Biochemistry | Seshagirirao & Prasad | 2001 |
| Plant | Ethnobotany | Mohammed et al. | 2004 |
| Latex | Tumor promoting Terpene | Baloch et al. | 2005 |
| Plant | Ethnobotany | Meena et al. | 2009 |
| Latex | Co-carcinogenic Diterpene | Baloch et al. | 2010 |
| Latex | Antimicrobial Activity | Goyal et al. | 2012 |
| Leaves | Antimicrobial Activity | Kapoor, B.B.S et al. | 2013 |
| Latex | Snakebite treatment | Chavre | 2013 |
| Plant | Biofuel | Rajeswari et al. | 2013 |
| Latex | Phytochemical Study | Bhattacharyya et al. | 2013 |
| Phylloclade | Ultrastructure of Laticifers | Rajeswari et al. | 2014 |
| Aerial part | Anticancer Activity | Bano et al. | 2017 |
| Plant | Ethnomedicine | Gupta et al. | 2017 |
| Latex | Case report: Ocular injury | Taviad et al. | 2017 |

Based on the literature survey the present investigation we framed the following objectives:

Objectives

1. Pharmacognosy and Phytochemical screening of *E. caducifolia*
2. Isolation and characterization of polysaccharide from phylloclade of *E. caducifolia*
3. Isolation and characterization of Lectin from latex serum of *E. caducifolia*
4. Isolation and characterization of unique starch granules from latex of *E. caducifolia*

Bioprospecting study

Sample collection

(Based on Ethnobotany or Chemotaxonomy or Random selection)

Place:

Jangati vari guttalu
(Bandasomaram)
Bhongiri, Telangana, IN

Coordinates:

17°30'19"N 78°54'35"E



Collected Time:
Flowering Season
(Feb-March)

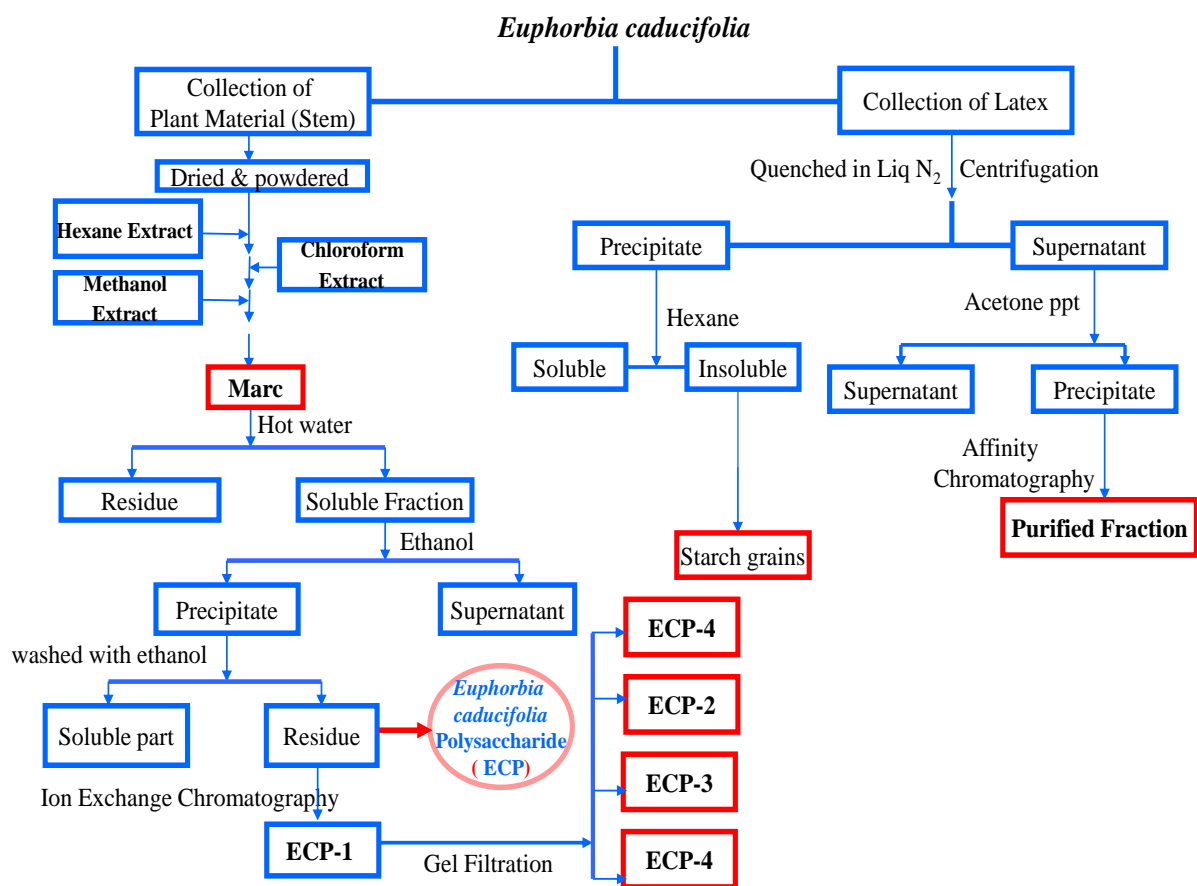
Taxonomic Identification of Plant by Curator Prof. K. Seshagirao
University of Hyderabad Herbarium (World Code – UH)

Germ plasm
Ex situ or /and Invitro



Herbarium
Vouche Specimen Nos. Venu-2027 & 2028





Isolation and purification Process of Biomolecules from *Euphorbia caducifolia*.

CHAPTER II

PHARMACOGNOSY

&

PHYTOCHEMICAL SCREENING OF *E.caducifolia*

Pharmacognosy is a study of crude drugs or knowledge of crude drugs on chemical as well as structural characterization and also metabolic pathways leading to pharmacologically active compounds. Pharmacognosy gained attention due to, search for new leads for drug development, biotechnology for the production of pharmaceuticals and nutraceuticals, increased interest on phytotherapy and validation of traditional medicines (Verpoorte, 2000).

Phytochemical screening is necessary to know the presence of metabolites and others in the extracts to produce the evidence for the ethnobotanical or ethnomedical surveys. High throughout screenings gives birth to the new moieties. Now it needs to explore new natural compounds from potential folk medicinal plants and ethnomedicinal plants for better therapeutic applications.

All living organisms contain an antioxidant defensive mechanism to counter the free radicals, reactive oxygen species (ROS), reactive nitrogen species (RNS) and other oxidants produced as byproducts of the metabolism. Most of the diseases caused by overproduction and reactive mechanism of free radicals (Alam et al., 2013; Valko et al., 2007). Many medicinal plants, spices, aromatic plants using as food supplements and medicines are natural antioxidants (Veeru et al., 2009). It has been an upsurge of interest

for potential antioxidants from plants sources. Therefore, this study was conducted to establish pharmacognosy of the plant and to reveal the phytochemical compounds exist in the phylloclade and to explore the ability of free radical scavenging activity as an antioxidant.

A. PHARMACOGNOSY OF *Euphorbia caducifolia*: *Euphorbia caducifolia*, collected from hills of the bongiri village, Yadadri District, Telangana State, India. The plant specimen was authenticated by Prof. K. Seshagirirao, Curator, University of Hyderabad Herbarium and deposited herbarium with voucher numbers Venu - 2027 and Venu-2028.

| | |
|---|--|
| Plant Name (Scientific Name) | : <i>Euphorbia caducifolia</i> Haines |
| (Common Names) | : Short time leaf succulent plant Danda Thor (Hindi) Kada jemudu or Katti jemudu (Telugu), |
| Family | : Euphorbiaceae |
| Distribution | : India, Pakistan & Afghanistan India (Rajasthan, Gujarat, Madya Pradesh, Bihar, Jharkhand, Chattisgarh, Odisha, Telangana, Andhra Pradesh, Karnataka, Maharastra) (Fig. 2.1) |
| Morphological Characters (Fig. 2.2) | : Plants are terete branching shrubs with height up to 2-3 m and 5-10 m broad, it can form Large Landscape |
| Phylloclade | : Modified stem, alternative spines |
| Leaves | : Caducus, Obovate, Fleshy, 3 to 6 cm long |
| Flower | : Cyathia inflorescence, Red & Yellow colour, Very dense, 1 cm across, Triad/single, central sessile |
| Capsule (Fruit) | : 3 lobed, 1 -1.2 cm, glabrous, dark pink on mature |
| Seeds | : Rounded, 0.2 -0.3 mm diameter, greyish, smooth |

- Microscopical Characters** : (Fig. 2.3)
- Phylloclades : T.S and L.S Phylloclades showed Laticifers
 - Leaves : Paracytic stomata, Vein islets: 5-7,
Vein termination:7-10
 - Seeds : Shows oil globules and starch granules,
Papillate surface
 - Latex : contain starch granules and calcium oxalate crystals.
- Powder Microscopy** : Phylloclade powder showed starch grains, Laticifers,
Calcium oxalate crystals
- Chemical constituents** : Phylloclades and latex contains Caducinol,
Cyclo caducinol, Epicyclo caducinol (Afza et al., 1989)
Acid polysaccharides (Uronic acid polysaccharides)
- Similar look like species/Substitutions/Adulteration possibilities:**
: *Euphorbia nivulia*, *Euphorbia gokakensis*

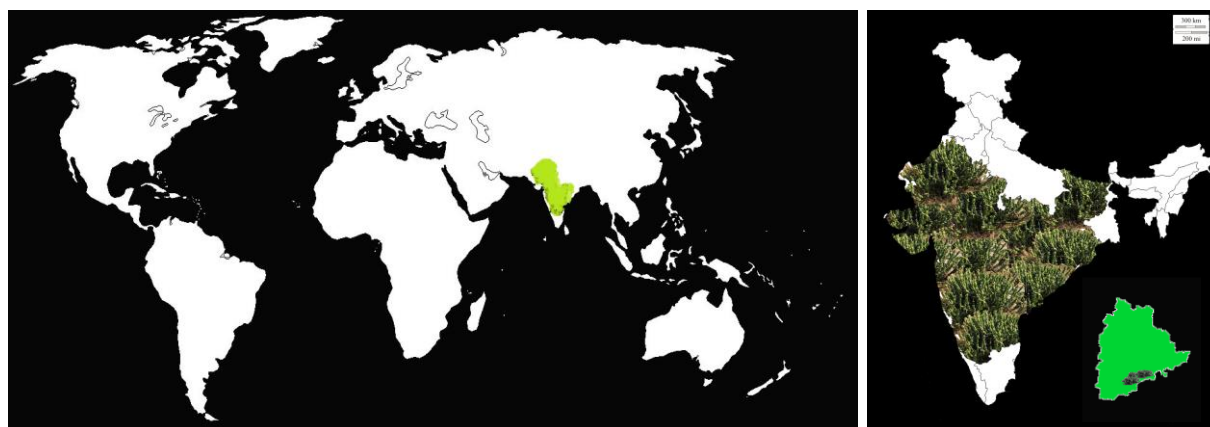


Fig. 2.1. Geographical distribution of *Euphorbia caducifolia*

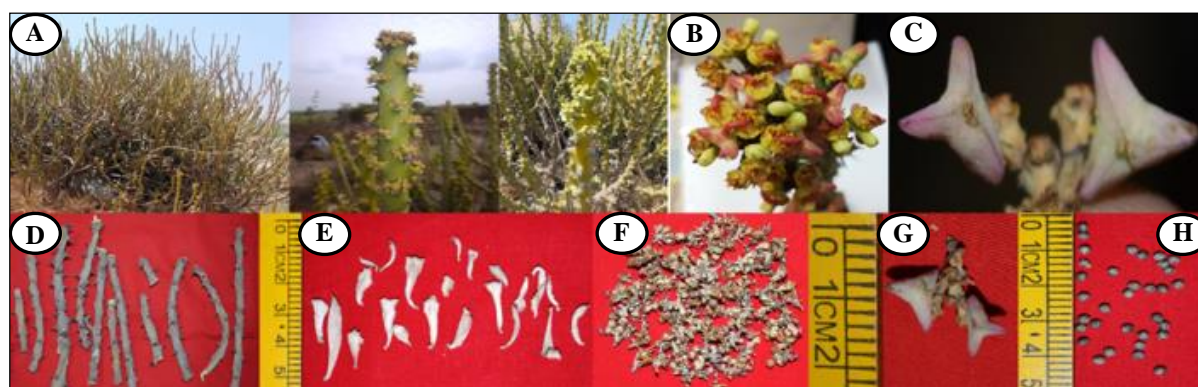


Fig. 2.2. Morphological characters of *E. caducifolia*:
A) Wild distribution, B) Cyathia with capsule, C & G) Capsule,
D) Dried phylloclades, E) Dried leaves, F) Dried flowers, H) Seeds

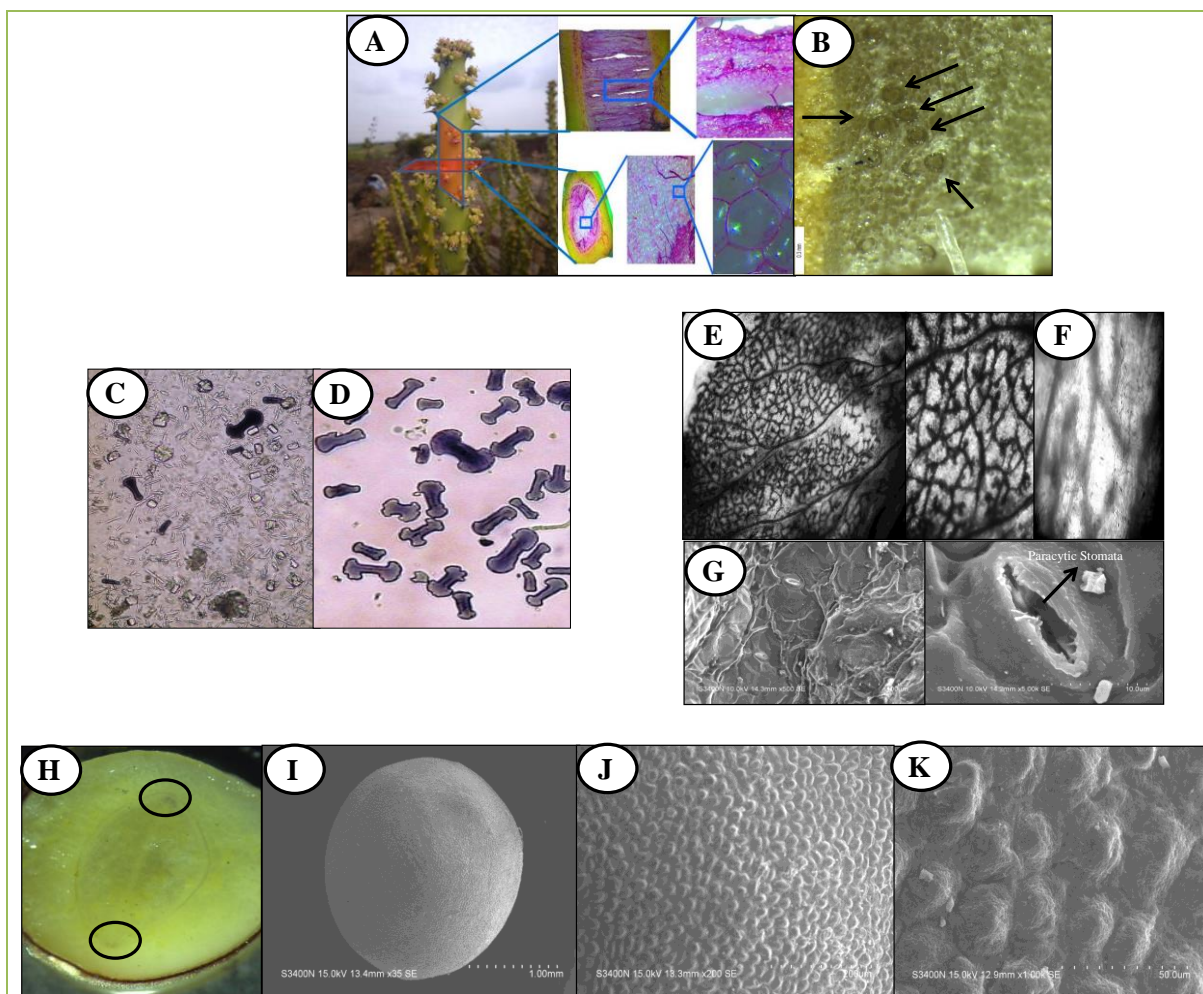


Fig. 2.3. Microscopic characterization of plant specimens: A) T.S & L.S of Phylloclade, B) Laticifers; Latex: C) Calcium oxalate crystals, D) Starch granules; Leaf: E) Venation, F & G) Paracytic stomata; Seed : Cross section with oil globules and starch (circles), SEM of seed: I) seed, J) Lower magnification (Papillate), K) Higher magnification (Undulate network)

B. PHYTOCHEMICAL SCREENING OF *E. caducifolia*

Materials and Methods

Preparation of Extracts: The dried plant material was made into powder and used for extraction. The powder was extracted with hexane, methanol, and water in successive order. The extracts were concentrated at temperature 40°C under vacuum and used for further analysis.

Phytochemical screening: Extracts were analyzed for alkaloids, carbohydrates, glycosides, flavonoids, phenolic compounds, quinones, tannins, saponins, and terpenoids (Santhi & Sengottuvel, 2016).

Estimation of Total Phenolic Content (TPC): The TPC of the extracts was determined using the Folin-Ciocalteu colorimetric method. 1 ml of extract was mixed with 1 ml of 1 N Folin's reagents and well shaken. After 5 min, 1 ml of 10% sodium carbonate was added and incubated for 1 h at room temperature. The absorbance was taken after incubation at 760 nm against blank. Total phenolic content was given as Gallic acid equivalents (GAE) (Basma et al., 2011).

Estimation of Flavonoids: 0.1 ml of extract was added to 0.3 ml 5% sodium nitrite and mixed with 3 ml of 1% aluminum chloride. After 5 min incubation, 2 ml of 1 M NaOH was added and made up the volume to 10 ml with water. The solution was mixed well and measured absorbance at 510 nm against blank. The flavonoid content was calculated using rutin as standard (Basma et al., 2011).

Antioxidant Activity

DPPH Radical Scavenging Assay: DPPH (2,2-diphenyl 1-picrylhydrazyl) free radical used to determine the scavenging activity of extracts (Molyneux, 2004). DPPH (1 ml, 0.1mM) added to various concentrations of extracts and BHT (3 ml) and 0.5 h incubated in the dark and measured absorbance at 517 nm against the blank (n=3). The activity was calculated by the following equation (Kumar et al., 2005):

$$\text{Percentage inhibition} = [(\text{Abs control} - \text{Abs sample}) / \text{Abs control}] \times 100$$

Nitric oxide radical scavenging Activity: Nitric oxide (NO) produced by Sodium Nitroprusside at pH 7.2-7.4 was determined by Griess reagent (Griess diazotization reaction). The reaction mixture (3 ml) was incubated at 25°C for 2 h containing Sodium Nitroprusside (10 mM) and extracts with different concentrations. An Aliquot (0.5 ml)

was added to Griess reagent (0.5 ml) and measured at 546nm (n=3). The activity was calculated by comparing control and test sample. Ascorbic acid used as a reference compound (Boora, Chirisa, & Mukanganyama, 2014).

Results and Discussion

Phytochemical analysis: Phytochemical studies of extracts showed the presence of various classes of metabolites. *E. caducifolia* showed the high quantity of terpenoids in methanol and hexane extracts as *Euphorbia* species are predominant with cyclic and acyclic terpenoids (Piazza, & Calvin, 1983). Test for the presence of alkaloids shown no significant colouration. Water extract was rich in carbohydrates, glycosides, tannins, phenols, flavonoids, and saponins. Further experiments revealed that the presence of acidic polysaccharides and saponin glycosides. Methanol extract contained reducing sugars, glycosides and some phenolic compounds whereas in hexane extract showed positive for reducing sugars and glycosides along with terpenoids (Table 2.1).

Table 2.1. Phytochemical Screening of *E. caducifolia* extracts

| Fraction | Water | Methanol | Hexane |
|---------------|-------|----------|--------|
| Alkaloids | -- | -- | -- |
| Carbohydrates | ++ | ++ | ++ |
| Glycosides | ++ | ++ | ++ |
| Flavonoids | ++ | + | -- |
| Phenolics | ++ | + | -- |
| Tannins | ++ | -- | -- |
| Saponins | ++ | -- | -- |
| Quinones | -- | ++ | ++ |
| Terpenoids | -- | ++ | ++ |

Estimation of total phenolic and flavonoid content: Phenolic compounds such as flavonoids and tannins are excellent free radical scavengers and protect the plant from different types of oxidative damages (Tadesse et al., 2016). Water extract showed the high phenolic and flavonoid contents whereas no significant amounts observed in hexane

extract. Methanol extract showed good amounts of flavonoids compared to phenolics (Table 2.2). Antioxidant activity of the extracts might be due to the presence of phenols and flavonoids present (Procházková et al., 2011).

Table 2.2. Total Phenolic and Flavonoid contents in *E. caducifolia* plant extracts

| Fraction | Phenolics | Flavonoids |
|----------|------------------------|-----------------------|
| | mg of GAE/g of Extract | mg of RE/g of Extract |
| Water | 68.740 ± 0.025 | 50.04 ± 0.125 |
| Methanol | 00.926 ± 0.06 | 13.534 ± 0.754 |
| Hexane | 00.080 ± 0.014 | 00.004 ± 0.0002 |

Antioxidant activity: Free radicals are the most reactive molecules and damage the cells. Aerobic organisms protect themselves from free radical by the defensive antioxidant mechanism, and it is necessary to provide external antioxidants in failure defensive system (Aksoy et al., 2013). The extracts were analyzed for their antioxidant capacity to use as a free radical scavenger. Most of the members of the castor family are a good source of natural antioxidants such as *Euphorbia hirta* (Basma et al., 2011), *E. heterophylla* (Okeniyi, Adedoyin, & Garba, 2012). *E. caducifolia* extracts also exhibited antioxidant activity. The potentiality of the extract to scavenge the free radical produced was considered for antioxidant activity (Fig. 2.4). DPPH and NO are the two radicals used for the experiments. All extracts were showed activity on concentration dependency. Water extract exhibited more scavenging activity towards DPPH (86±1.5%) as well NO (39.23±4.7%) to hexane extract. More free scavenging activity of water extract might be due to the presence of the high phenolic and flavonoid contents. Methanol extract also exhibited similar (DPPH-77.81±1.3% and NOSA-36.03±2.71%) to water but dropped activity frequency at high concentration of the extract. Higher antioxidant activity was

observed at higher concentrations. Extracts showed less ability to scavenge the NO free radical as compared to DPPH.

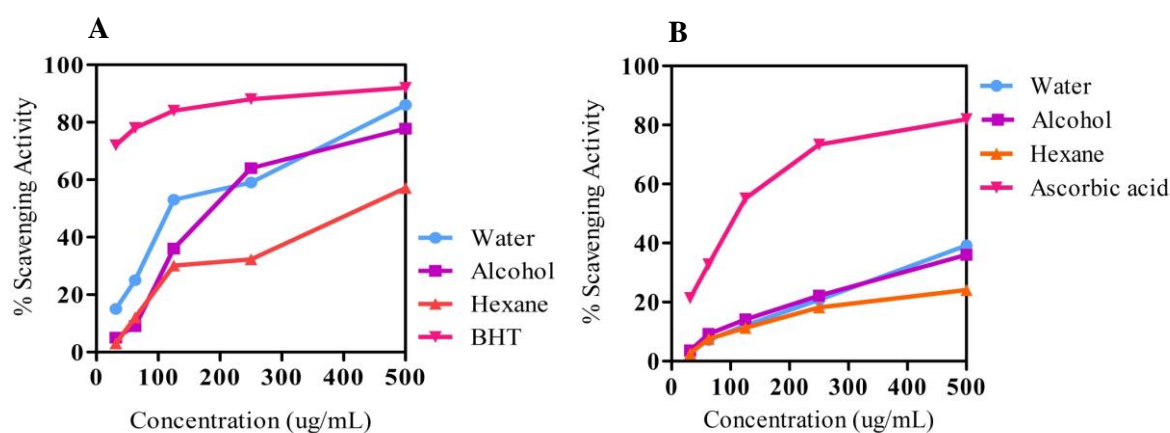


Fig. 2.4. Antioxidant activity of *E. caducifolia* plant extracts

A) DPPH Scavenging activity, B) Nitric Oxide Scavenging activity

Conclusions

Euphorbia caducifolia is a promising plant for the various therapeutic applications with rich contents of terpenoids, polysaccharides, tannins, flavonoids and phenolic compounds. The presence of significant amount of total phenolics and flavonoids contents in water extract might be responsible for strong antioxidant activity as compared with methanol and hexane extract. Hexane and methanol extract can be used for the terpenoid sources and water extract for phenolics and flavonoids. High contents of terpenoids in methanol and hexane extracts are useful in various applications. A further study of isolation and characterization of bioactive molecules from the plant is required for therapeutic and multiple applications.

CHAPTER III

ISOLATION

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CHARACTERIZATION OF POLYSACCHARIDE FROM PHYLLOCLADE OF *E.caducifolia*

1. Optimization of extraction process, Structural and Biological characterization of Polysaccharide

Polysaccharides can serve as an energy source and structural components to the cell with its highly complex structure (Z. Wang et al., 2007). Polysaccharides from higher plants show significant therapeutic properties by its non-toxic nature and ability to not cause any side effects compared to bacterial and fungal polysaccharides. Thus, botanical polysaccharides are ideal therapeutics for immunomodulatory, antitumor and wound healing activities (Schepetkin & Quinn, 2006).

Euphorbia caducifolia Haines, a Euphorbiaceae species grows extensively on rocky areas of arid and semi-arid regions with extensive branching and clumps of phylloclades (Shaik et al., 2013). Phylloclades are fleshy, succulent with spiny stipules. Milky and sticky natured latex flows through laticiferous cells along with proteins, phytosterols, and terpenoids (Afza et al., 1989). Plant leaves were consumed as Food in arid areas. Ethnomedical uses of the plant were reported, latex for wound healing activity and other skin infections (Goyal, Nagori, & Sasmal, 2012), root extracts for snakebite

(Chavre, 2013), leaves for antimicrobial activity (Kapoor et al., 2013) and phylloclades as biofuels (Shaik et al., 2013).

Response Surface Methodology (RSM) is known to be the best technique to evaluate the relationship between parameters and the yield (D. Ye et al., 2015). Extraction temperature, solid to liquid ratio and time are the common variables which affect the yield. In RSM, Box-Behnken experimental design is more efficient and easier method to analyze and interpret the variables affect the yield (Chaiklahan et al., 2013).

The objective of the study was to evaluate and optimize the variables affecting the polysaccharide yield using a single factorial (RSM), three level-three variable design (BBD). In addition, reporting its structural and biological studies as therapeutic interests.

Materials and Methods

Plant material: The collected stems were dried at room temperature until it gains constant weight. Dried stems were ground to pass 2 mm sieve and proceeded for the extraction.

Extraction process: The powder was freed from oils, fats, fatty acids, and other phytochemicals by sequential extraction using the soxhlet apparatus with hexane, chloroform, and methanol. The residual marc was dried and used for polysaccharides extraction (Huang et al., 2010).

Water extraction: Response surface methodology was used to extract crude polysaccharides in coercion with hot water method. In brief, dried marc was suspended in established volume of distilled water at designed temperature and time. The suspension was centrifuged, and collected supernatant was concentrated under vacuum at 45°C. The polysaccharides from the supernatant was precipitated with absolute ethanol and garnered

the pellet by centrifugation. The precipitated pellet was washed with different percentages of ethanol to remove soluble material and deproteinized using sevag method (Zou et al., 2015). A brown coloured crude polysaccharide was obtained (C.-L. Ye & Lai, 2015) and the concentration was determined by using Phenol-Sulphuric acid method (DuBois et al., 1956). The yield was calculated using the following equation:

$$\text{Yield \% (w/w)} = [\text{content of polysaccharide (g)/powder weight (g)}] \times 100$$

Response Surface Methodology (RSM) Experimental Design: Box-Behnken design (BBD), a three-level three-factor design was used to optimize the polysaccharides extraction. Parameters like temperature (A), solid to liquid ratio (B) and time (C) were the independent variables, and the polysaccharide yield was taken as response function (R) to the independent variables. The BBD matrix consisted of total 17 experimental points in random order with five replications of center points (D. Ye et al., 2015). Each variable was set at three levels and coded -1, 0, +1 for low, intermediate and high values respectively (Table 3.1). The fitted polynomial equation was expressed as surface and contour plots for visualizing the relationship between three independent variables and the response variable (Yuan et al., 2015).

Table 3.1 Factors and levels in Box-Behnken Design used to analyze extraction yield of polysaccharides from *Euphorbia caducifolia*.

| Variables for Extraction | Factor Level | | |
|--------------------------------|--------------|----|-----|
| | -1 | 0 | 1 |
| A: Temperature (°C) | 40 | 80 | 120 |
| B: Solid to liquid ratio (g/L) | 25 | 50 | 75 |
| C: Time (h) | 1 | 2 | 3 |

Purification and Monosaccharide analysis: Crude polysaccharide obtained was redissolved in water, and applied on DEAE Cellulose-52 column (2.5×30) and equilibrated with distilled water. The polysaccharides were eluted with stepwise NaCl solution (0, 0.1, 0.5 and 1M). The eluates containing polysaccharides were pooled, concentrated and dialyzed against distilled water. Dialyzed fraction was lyophilized (considered as EC polysaccharide or ECP) and stored at -20°C (Ye et al., 2008). EC polysaccharide was analyzed for monosaccharide composition using GC-MS. The polysaccharide was hydrolyzed in 2.5 M TFA for 3 h at 121°C in a sealed tube, and residual acid was removed with methanol under N₂ stream. The residue was dissolved in pyridine and derivatized with silylating reagent (BSTFA and 1% TMCS) in 1:2 ratio at 40°C for 2h. Monosaccharide derivatives were extracted with dichloromethane and analyzed for composition using GC-MS (Pitthard & Finch, 2001).

Spectral Analysis: Spectra of EC polysaccharide was taken by using Chemito UV2100, UV-Visible spectrometer and Bruker Alpha FTIR, Fourier transform infrared spectrometer. For UV-Visible spectrum, polysaccharides dissolved in distilled water and spectrum recorded at a frequency range 1100-190 nm. The FTIR spectrum was measured at a range of 4000-500 cm⁻¹ (Yuan et al., 2015).

Thermal Analysis (TG-DTA): Mettler TGA/DSC1 was used to determine the thermal properties of polysaccharides. EC polysaccharide was weighed to 4.1 mg and heated at a rate of 10°C/min from 40°C - 600°C under the flow of N₂ atmosphere (Janarthanan, Zin Wan Yunus, & Ahmad, 2003).

Antioxidant activity

a) DPPH-free radical scavenging activity: Blois method was used to determine the free radical scavenging activity of EC polysaccharide using DPPH radical with minor modifications (Molyneux, 2004). 1 ml of DPPH (0.1mM) was added to 3ml of various concentrations of polysaccharides (0.125-2 mg/mL) and incubated for 30 min in the dark and absorbance was measured at 517 nm. Ascorbic acid was used as a standard. Three separate experiments were performed, and the percentage inhibition of the polysaccharides was calculated by the following equation: (Kumar et al., 2005)

$$\text{Percentage inhibition} = [(A \text{ control} - A \text{ sample})/A \text{ control}] \times 100$$

Where A- Absorbance.

b) Nitric Oxide radical scavenging activity: Sodium Nitroprusside gives nitric oxide (NO[•]) in aqueous solution at pH 7.2-7.4, which reacts with oxygen to produce nitrite ions and the formed nitrite ions determined by Griess reagent (Griess diazotization reaction). 3 ml reaction mixture containing 10mM Sodium Nitroprusside in phosphate buffered saline pH 7.2 and EC polysaccharide with different concentrations (0.125-2 mg/mL) were incubated at 25°C for 2 h. 0.5 ml aliquot of reaction mixture was added to 0.5 ml of Griess reagent (equal parts of 1% sulphanilamide in 5% H₃PO₄ and 0.1% naphthyl ethylenediamine dihydrochloride), and the absorbance was measured at 546nm. BHT (Butylated hydroxytoluene) used as reference compound. All three experiments were done separately, and the percentage inhibition was calculated by comparing control and test samples (Green et al., 1982).

c) Reducing power: 1 ml of EC polysaccharide (0.125-2 mg/mL) was added to 2.5 mL of phosphate buffer (pH 6.6, 0.2 M) and 2.5 ml of 1% potassium ferricyanide

[K₃Fe(CN)₆] and incubated at 50°C for 30 min. 2.5 ml of 10% TCA was added to the mixture and centrifuged for 10 min at 3000 rpm. To the 2.5 ml of supernatant, 2.5 ml of distilled water and 0.5 ml of 0.1% FeCl₃ added. BHT used as reference compound. The absorbance was measured at 700 nm. The increase in the absorbance indicates the high reducing power (Yildirim et.al., 2001).

Emulsifying activity: The emulsifying activity of EC polysaccharide was performed by turbidimetry method (Benhura & Chidewe, 2004). 3ml of different concentrations of EC polysaccharide in water was added to the 1 ml of vegetable oil (castor oil) and homogenized for 10 min. A 50 µl aliquot was diluted with 5ml of 0.1% Sodium dodecyl sulfate, and the absorbance was measured at 500 nm. The ability of the emulsion was determined by recording the absorbance at 0 h, and the stability of the emulsion was determined by allowing the emulsion at room temperature for 24 h. The turbidity of the emulsion was calculated by:

$$T = 2.303A/l;$$

where T- Turbidity; A- Absorbance; *l*- Path length of the cuvette;

MTT Assay: Murine macrophages (RAW 264.7) was used for the experiment. The cells (1×10⁴) were inoculated in 96 well plate and incubated for 24 h in an incubator (5% CO₂, 37°C). The cells were treated with different concentrations EC polysaccharide and incubated for 24 h. After incubation, 3(4,5-dimethyl thiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) was added as 5 mg/mL and incubated further for 4 h. 100 µl of DMSO added to the wells and absorbance taken at 590nm. Percentage of cell viability was calculated with untreated cells as a control (Khan et al., 2006).

Cytokine Measurement: RAW 264.7 cells (5×10^5 cells/mL) were stimulated with EC polysaccharide and the levels of cytokines, Tumor necrotic factor α (TNF- α) and Interleukin-6 (IL-6) were measured after 24 h of incubation (Schepetkin et al., 2008; Xu et al., 2006). TNF- α and IL-6 concentrations were determined by extrapolation from the standard curve, according to the manufacturer's protocol (BD Bioscience).

Results and Discussions

Polysaccharides extraction: Response surface methodology, a statistical technique used to analyze the relationship between parameters and the yield (D. Ye et al., 2015). The polysaccharide yield obtained with an experimental matrix of the Box-Behnken design was calculated and showed the yield in the range of 0.9% to 2.4% (Table 3.2). As single factor analysis, temperature played a significant role in the percentage yield of the polysaccharides compared with solid to liquid ratio and time by enhancing diffusion coefficient of the polysaccharide at high temperatures (Ros et al., 2004). High temperatures ($>90^\circ\text{C}$) were used to extract polysaccharides from algae and plant sources (Chaiklahan et al., 2013). The response surface curves (3D) and contour plots (2D) of the design gives a visual relation between parameters to the percentage of yield (Fig. 3.1.1). Although the rise in temperature increases the polysaccharides yield, its interaction with solid to liquid ratio and time also affect the yield percentage. The influence of temperature on the yield with solid to liquid ratio shows statistical significance but not with time, albeit longer time affect the yield (Table 3.2). At 120°C , 2.36% of yield was obtained with 50 g/L of solid to liquid ratio and 3 h of time whereas 2.30% of yield was obtained with 75 g/L of solid to liquid ratio and 2 h of time, and the insignificant difference of the yield (0.06%) shows the interchangeable effect of the other two parameters on the yield. From RSM results it observed that yield saturation reached to 1.5% & 2.4 % at below and above 90°C respectively. The interactions between the

variables and yields were related with the help of multiple regression analysis by following second-order polynomial equation:

$$Yield(Y) = 1.44 + 0.57A + 0.062B + 0.055C + 0.15AB + 0.070AC + 0.020BC + 0.13A^2 + 0.071B^2 + 0.005C^2$$

Table 3.2: The Box-Behnken Design matrix and results of extraction yield (%) of polysaccharides from *Euphorbia caducifolia*

| Run | A: Temperature (°C) | B: Solid to liquid ratio (g/L) | C: Time (h) | Yield (%) |
|-----|---------------------------|--------------------------------------|-------------------|--------------|
| 1 | 80 | 25 | 1 | 1.39 |
| 2 | 80 | 75 | 1 | 1.62 |
| 3 | 80 | 75 | 3 | 1.69 |
| 4 | 40 | 50 | 1 | 0.93 |
| 5 | 80 | 25 | 3 | 1.38 |
| 6 | 40 | 75 | 2 | 0.96 |
| 7 | 80 | 50 | 2 | 1.46 |
| 8 | 80 | 50 | 2 | 1.53 |
| 9 | 80 | 50 | 2 | 1.40 |
| 10 | 40 | 25 | 2 | 1.29 |
| 11 | 120 | 75 | 2 | 2.30 |
| 12 | 40 | 50 | 3 | 0.98 |
| 13 | 80 | 50 | 2 | 1.40 |
| 14 | 80 | 50 | 2 | 1.43 |
| 15 | 120 | 50 | 3 | 2.36 |
| 16 | 120 | 25 | 2 | 2.01 |
| 17 | 120 | 50 | 1 | 2.03 |

From the regression analysis, the high value of the determination coefficient R^2 (0.97) and the adjusted R^2 (0.97) explains the high fitness of the regression model and correlation between the values. The Analysis of Variance (ANOVA) of the model suggests that the temperature variable affects the yield significantly as well as its interaction with solid to liquid ratio variable influence the yield ($P < 0.05$). The optimum conditions for the better extraction of the polysaccharides are 120°C of temperature, 75 g/L of solid to liquid ratio and 2 hr of time which correlates with predicted yield (2.425%) and experimental yield (2.30%).

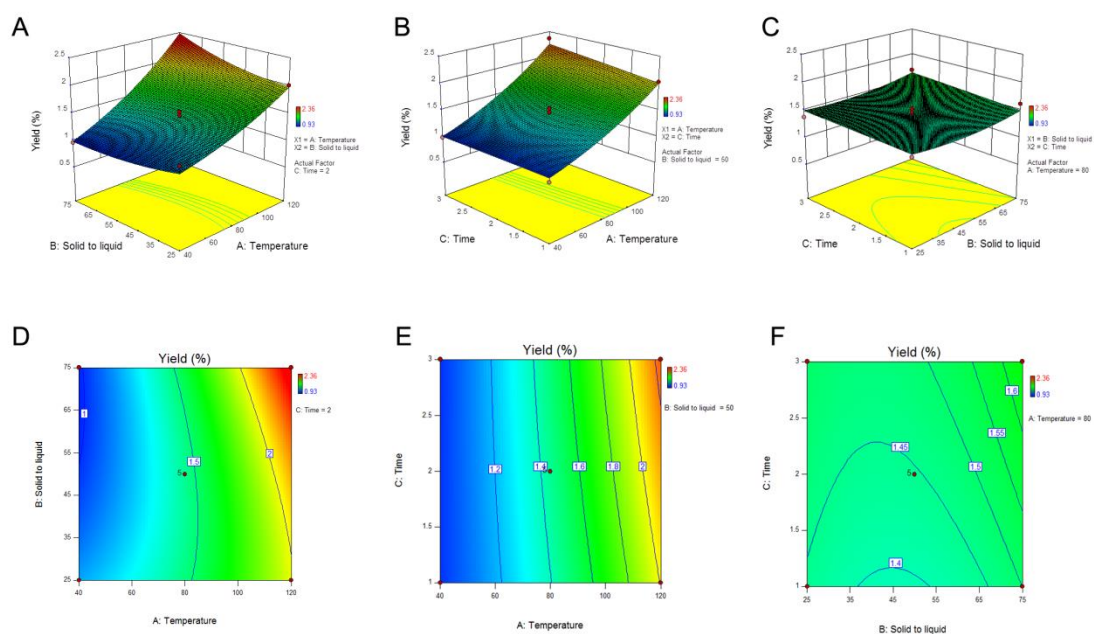
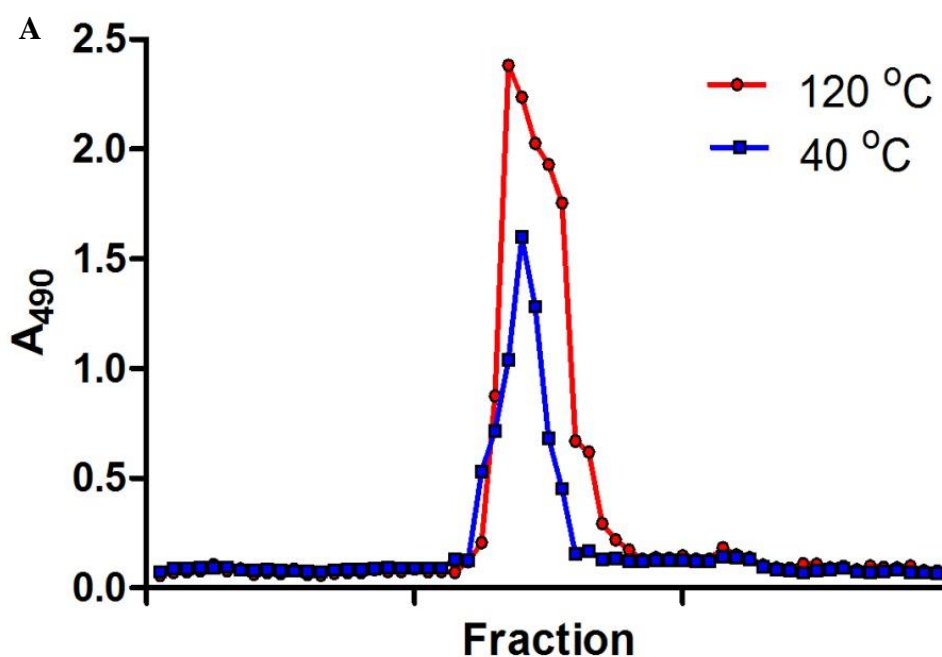


Fig. 3.1.1: Response surface curve (3D) showing the relation between parameters and polysaccharide yield; A) temperature and solid to liquid ratio, B) temperature and time, C) solid to liquid ratio and time. Contour plots (2D) showing the relation between parameters and polysaccharide yield; D) temperature and solid to liquid ratio, E) temperature and time, F) solid to liquid ratio and time.

Purification and Monosaccharide analysis: The lyophilized crude polysaccharide was chromatographed on an anion exchange column to get acidic polysaccharide fraction. Use of an anion exchange chromatography to elute the acidic polysaccharides has been

reported from wheat, sugar beet by Neukom et al. (1960) (Jermyn, 1962). From crude polysaccharide, the only fraction eluted with 1 M NaCl contains mannose, rhamnose, galactose, glucose, galacturonic acid, galactaric acid, arabinose, ribose, talose and xylose as monosaccharide constituents (Fig. 3.1.2). Previously, it has been reported that the presence of acidic polysaccharides with uronic acids and mucic acids in the leaves of *A. rosea*, *H. syriacus*, *C. olitorous* (Ohtani, Okai, Yamashita, Yuasa, & Misaki, 1995). Interestingly, similar compositions were observed with *E. caducifolia* as reported in roots of its family member *E. fisheriana* (Liu et al., 2011). Acidic polysaccharides extracted from plant sources have been reported for their extensive biological activities like hypoglycemic, anticomplementary, immunomodulatory (Ohtani et al., 1995). Similar to that of other acidic polysaccharides, EC polysaccharides was also expected to show biological activities like antioxidant and immunomodulatory activities.



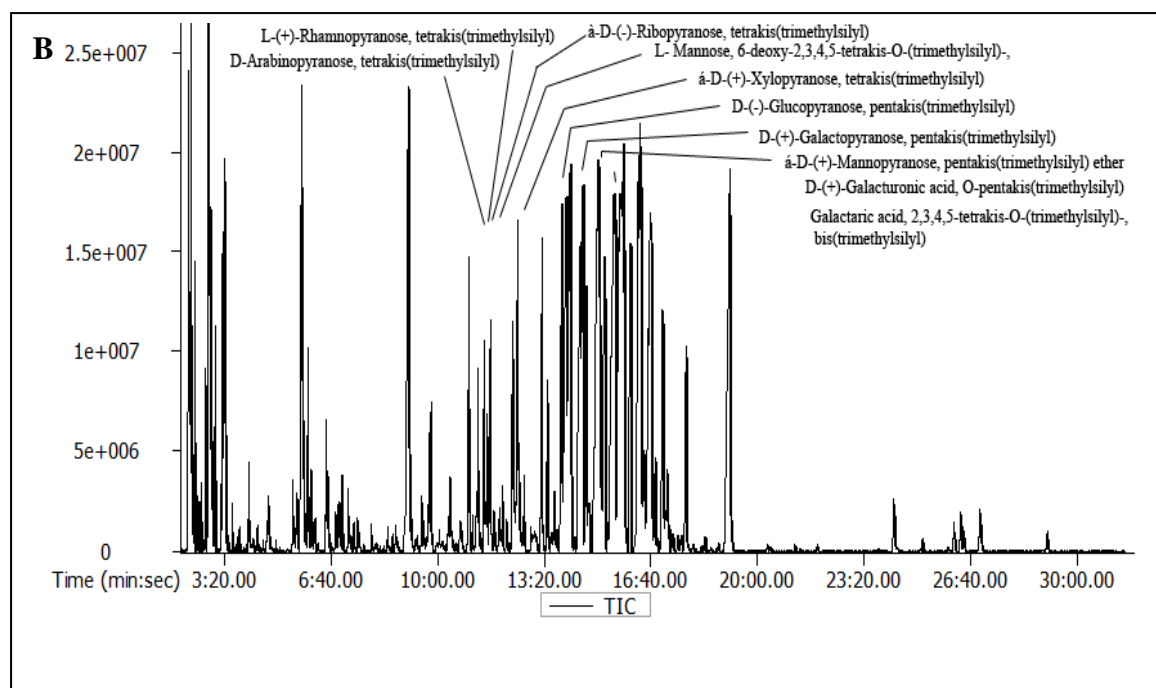


Fig. 3.1.2 Chromatograms of the EC polysaccharide;

A) Elution profile of EC polysaccharides purification on DEAE cellulose column at 120°C and 40°C, B) GC-MS profile of the EC polysaccharide for monosaccharide analysis: Trimethylsilyl derivatives of sugars

Spectral analysis: UV-Visible and infrared spectra of the EC polysaccharides exhibited typical characters of the polysaccharide. The UV-visible spectrum (Fig. 3.1.3A) of the polysaccharide did not show any absorption peaks at 260 nm and 280 nm which correlates with the absence of protein and nucleic acid respectively (Bian, Xie, & Chen, 2010). Forbye the lack of sharp peaks at amine stretch in IR spectrum of the EC polysaccharides (Fig. 3.1.3B) mount the evidence that no contamination in purified polysaccharides. A characteristic broad peak at around 3330 cm^{-1} assigned to the hydroxyl group(-OH) present in the polysaccharide. An aliphatic C-H stretching vibrations at 2921 cm^{-1} and weak absorption at 1418 cm^{-1} is from aliphatic C-H bending of CH_2 (Parikh & Madamwar, 2006). The two peaks towards the 1738 cm^{-1} and 1605 cm^{-1} in the spectra resulted from the COO^- deprotonated carboxylic group and two medium

broad peaks at 1075 and 1018 cm^{-1} suggest the presence of C-O bonds. β -configuration of the monomers present in EC polysaccharides displayed with a peak around 800 cm^{-1} (Zhou et al., 2000). Merged peaks in the fingerprint region indicate the linkages between monosaccharides.

Thermal analysis; Thermal nature of the polysaccharides was analyzed using TG-DTA. The thermogravimetric curve (Fig. 3.1.3C) of the EC Polysaccharides showed decomposition with three differentiated steps. The first mass loss event occurred from 50°C to 120°C with the loss of mass 10-15 % due to the diminution of adsorbed water which represents the hydrophilic nature of the polysaccharide (Zohuriaan & Shokrolahi, 2004). The Second stage starts at 230°C retains till 350°C and observed a loss of weight ca.36% at 285°C because of thermal depolymerization. The carbonization of polysaccharide causes sustained weight loss in the last stage occurred above 350°C to 600°C. The presence of charged cations like Na^+ , K^+ , Ca^{2+} and complex configuration of the polysaccharide resist the degradation process and left the solid residue around 35% after 600°C (Cerqueira et al., 2011; De, Ruiz-bermejo, Menor-salván, & Osuna-esteban, 2011). DTA thermogram of the EC polysaccharide showed early endotherm peak at 78°C with an onset temperature 48°C and end set point 114°C. Enthalpy change (ΔH) of 135.15 J/g was noticed at early stages. No glass transition (T_g) temperature was exhibited in the thermogram which attributed to interference of early endothermic peak (Parikh & Madamwar, 2006).

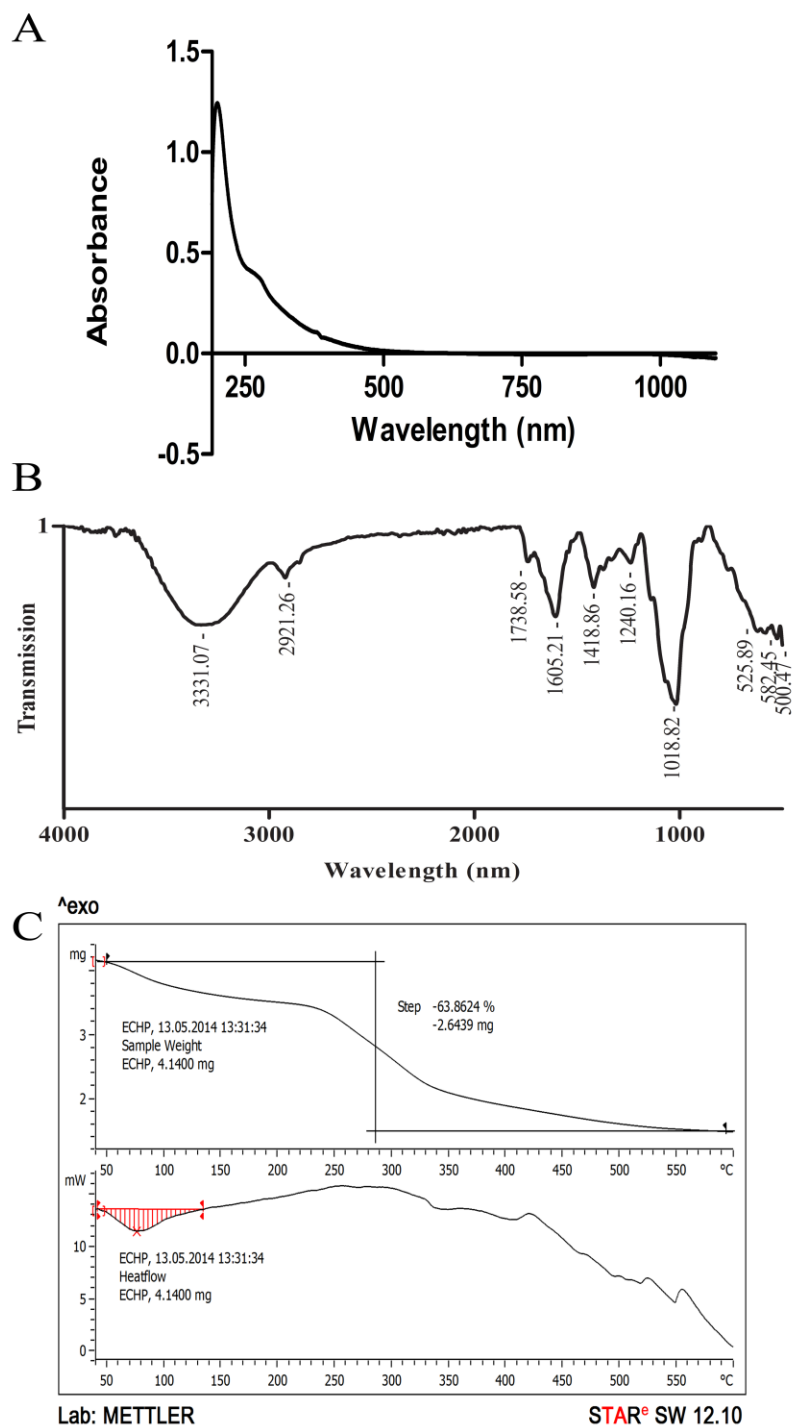


Fig. 3.1.3 Spectra and thermogram of EC polysaccharides;

A) UV-Visible Spectrum, B) ATR-IR Spectrum, C) TGDTA thermogram at heating rate $10^{\circ}\text{C min}^{-1}$ under a nitrogen atmosphere.

Antioxidant Activity: A simple and sensitive, DPPH free radical scavenging activity was performed to observe the antioxidant activity of polysaccharides. Antioxidant activity was

increased as the concentration of the polysaccharides increases and Ascorbic acid used as a standard to compare EC polysaccharides. Free radical scavenging activity of the polysaccharide gained momentum of activity after 0.5 mg/ml concentration to greater 35% to 65% (Fig. 3.1.4A). Besides DPPH, Nitric Oxide (NO[•]) free radical scavenging activity was also measured (Fig. 3.1.4B). NO[•] react with oxygen to give free radicals nitrite and proxy nitrite, the reason for many inflammatory processes and cause of several diseases (Luyen et al., 2014). EC Polysaccharides, a competitor for oxygen to react with NO[•] and thus inhibits the production of nitrates and nitrites. EC polysaccharides at 2 mg/ml concentration showed the free radical scavenging activity 65% with DPPH and 55% with Nitric Oxide. The ability of the polysaccharides to scavenge the free radicals in the reaction mixture more than 50% suggests as prospective antioxidants.

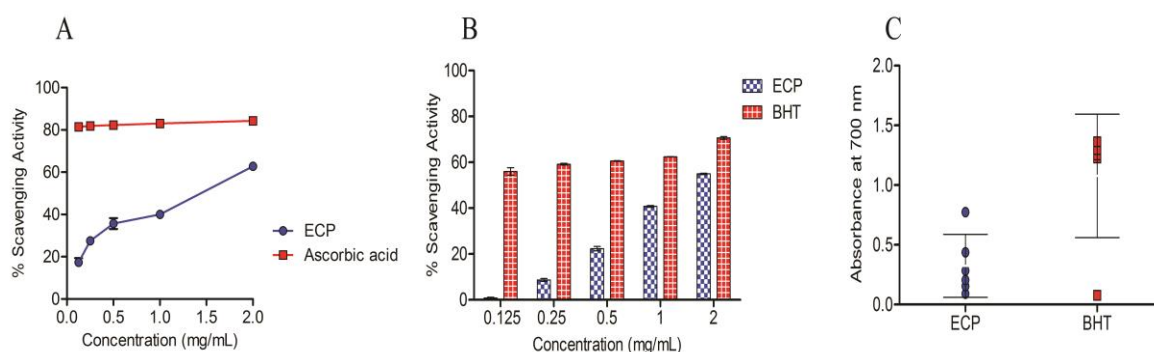


Fig. 3.1.4 Antioxidant activity of EC polysaccharides;

A) DPPH free radical scavenging activity, B) Nitric oxide radical scavenging activity, and C) Reducing power.

(values are mean ± SE, n=3)

The reducing ability of a compound to donate electron can serve as a potential antioxidant. Apart from free radical scavenging activity, the antioxidant activity of a compound can also be evaluated by its capacity to reduce the ferric cyanide complex (Fe³⁺) to ferrous cyanide form (Fe²⁺). Thereby the colour change in the reaction mixture

to different shades of green to blue indicates the antioxidant activity (Irshad, Zafaryab, Singh, & Rizvi, 2012). Similarly, green to blue shades were observed with EC polysaccharides, and prussian blue with BHT, where prussian blue indicates strong reducing activity. The absorbance was measured, as it proportionately correlates with the reducing power. Even though polysaccharide showed less degree of absorbance compared to BHT (Fig. 3.1.4C), its reducing ability as an electron donor evinced with increasing polysaccharide concentration.

Emulsifying Activity; Emulsions are the mixture of immiscible liquids where droplets of one liquid dispersed in other with the help of an emulsifier. Most of the acidic polysaccharides like gum arabic, gum karaya, gum tragacanth are using as emulsifiers in the pharmaceutical and food industries (Benhura & Chidewe, 2004). In an interest to that, the efficacy of the EC polysaccharide as an emulsifying agent was tested by turbidimetry method. The emulsification ability of the EC polysaccharide was observed under a microscope, further measured the turbidity with a spectrophotometer (Fig. 3.1.5). The undiluted emulsion showed flocculation of droplets into various sizes. Further dilution with 0.1% sodium dodecyl sulfate, resulted in more dispersed globules and aggregates of floccules on standing for 24 h. In turbidimetry, increasing the concentration of the polysaccharide to oil shown linearity of the emulsification, and observed 36% of turbidity shift at high concentration and 88% of shift at lower concentrations of polysaccharide on long standing. Stabilization of dispersed phase in continuous phase even after 24 h suggests that the EC polysaccharide as a good emulsifier. Phase separation of the liquids occurred within 1 hour in the control where no polysaccharide was added.

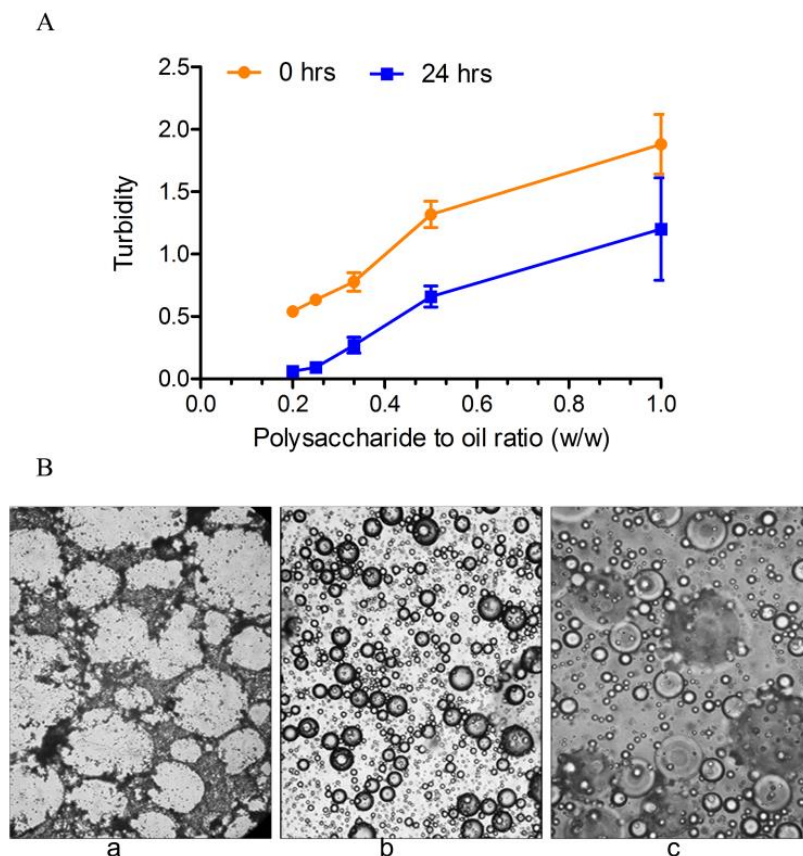


Fig. 3.1.5 Emulsion ability of the EC polysaccharides;

A) The turbidity of emulsion measured at 500 nm at 0 h and 24 h, B) Photomicrograph of Emulsion: a) Control (without polysaccharide), b) Emulsion at 0 h, c) Emulsion at 24 h.

(Values are mean±SE, n=3)

MTT Assay and Cytokine measurement: Polysaccharides from most of the plant sources are non-toxic in nature and modulate cytokine and chemokine production as an immunomodulator (Schepetkin & Quinn, 2006). Non-toxicity and immunomodulatory activities place the polysaccharides as ideal therapeutic candidates. The EC polysaccharides showed non-toxic nature and also directly stimulate the proliferation of RAW 264.7 cells (Fig. 3.1.6A). The proliferation effect was observed in the range of 50 ng/mL-1 µg/mL, i.e., from 107% to 150%, a further increase in the concentration depicted no significant cytotoxicity on cells compared to untreated cells.

Most of the botanical polysaccharides activate several signal transduction pathways of macrophages and stimulate Nitric oxide (NO), Reactive oxygen species (ROS), and production of inflammatory (IL-1, IL-6, IL-12) as well as anti-inflammatory cytokines (IL-10) along with chemokines (TNF- α) (Schepetkin & Quinn, 2006). Treatment of macrophages (RAW 264.7 cells) with various concentrations of EC polysaccharides significantly enhanced the production of the pro-inflammatory cytokines such as IL-6 and TNF- α whereas untreated macrophages produced negligible amounts (Fig. 3.1.6B, C). The maximum IL-6 production observed with dose effective from 500 ng/mL to 100 μ g/mL. Cell proliferation observed with EC polysaccharides at lower concentrations may be due to the initial production of IL-6, which stimulates the extracellular signal-regulated kinases and Mitogen-activated protein kinase pathways (Ogata et al., 1997). Increased levels of TNF- α and IL-6 at higher concentrations might be responsible for the certain level of cell death (<40%). These results evidence that the EC polysaccharides at lower concentrations can be used as a therapeutic agent for promoting wound healing and regeneration activities.

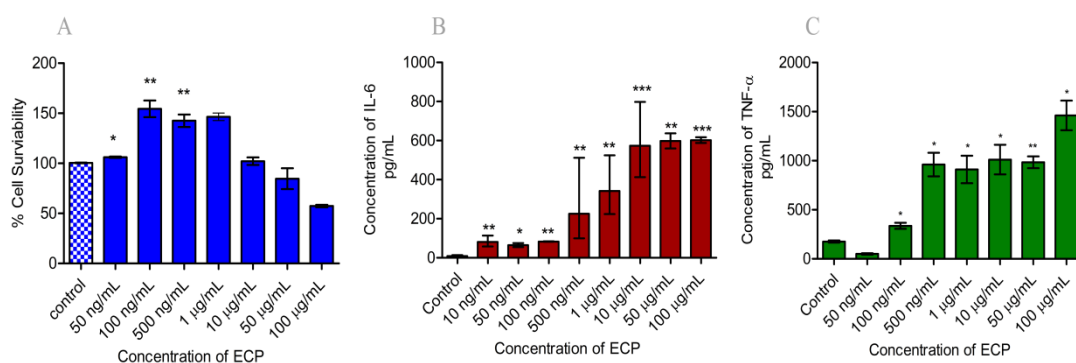


Fig. 3.1.6. Effect of the EC polysaccharides on the RAW 264.7 at a different concentration; A) Cell viability, and B) Concentration of the IL-6, C) TNF- α measured by ELISA in the culture supernatant.

(n=3, values= mean \pm SE, * =P<0.05, **=P<0.01, ***=P<0.001)

Conclusions

Altogether this study suggests that manipulation of the temperature influence the polysaccharide yield along with solid to liquid ratio and time. Polysaccharides extracted to a maximum yield of 2.3% at 120°C, and the optimum conditions were identified using the response surface methodology. Purified acidic polysaccharides with uronic acids shown typical nature of spectral as well as thermal characteristics. EC Polysaccharides act as potential antioxidants and as a good emulsifier. Moreover, polysaccharides exhibited immunomodulatory activity with non-toxic nature on RAW 264.7 cells.

2. Chemical modifications of isolated polysaccharide and its characterization

Polysaccharides are the carbohydrates composed of same or different types of saccharides in an organized manner with glycosidic bonds. These are used for various industrial applications mainly in food and pharmaceutical companies. Modifications of the polysaccharides change its structure and lead to the physical as well as functional differences (Cumpstey, 2013). Mostly, chemical methods are used to modify its structure to study the structure-activity relationship in interest towards its physical properties like solubility and biological activities among various types of molecular modifications. Polysaccharide hydroxyl groups are major contributors for chemical modifications. Apart, Alginates and pectins are composed of carboxylic groups called uronic acids. In modifications, Saccharide hydroxyl oxygen act as nucleophile and carbon serve as an electrophile. Nucleophile oxygen can undergo chemical reactions like Etherification (carboxymethyl ether, hydroxyethyl ethers), Esterification (acetyl, carboxyl esters, sulphonate esters) and electrophile carbon can undergo introducing the groups like sulfate and halide groups (Cumpstey, 2013). Additionally, oxidation of sugars, nitrogen group reactions of amino sugars, esterification and other substitutions on carboxylic groups of uronic acid sugars are also considered. All together gives change in chemical, physical and biological changes.

Materials and Methods

Sulfonation: 50 mg of polysaccharide dissolved in 10 ml of formamide and 10 ml of pyridine mixture. After, added 4 ml of chlorosulfonic acid on ice bath about 2 hrs. Placed the reaction mixture for 15 hrs at 4°C and ended the reaction by adding cold water. Sodium hydrogen carbonate used to neutralize the reaction mixture until it ceases the

effervescence. Then, dialyzed the solution about 7 days against distilled water and concentrated under reduced pressure and lyophilized (O'Neill, 1956).

Carboxymethylation: Yang et al. 2011 method was followed with minor modifications (Yang et al., 2011). 10 mg of polysaccharide suspended in 4 mL of isopropanol and kept under stirring at RT for 15 min, followed by the slow addition of 1.5 mL of 20% NaOH with 1 h stirring. Then the solution of 150 mg/ml of chloroacetic acid was added to the reaction mixture with stepwise under constant stirring. The reaction was continued for 4 h at 60°C. Cooled down to RT, and 0.5 M of hydrochloric acid was added to neutralize the mixture. The derived polysaccharide was dialyzed against normal water for 6 h and then with distilled water for 12 h. Dialysate was precipitated with ethanol and lyophilized.

Acetylation: 50 mg of ECP dissolved in 7 ml of formamide and further added 12.5 ml of pyridine. After, added 10 ml of acetic anhydride at 4°C about 2 h at frequent intervals. The reaction was continued for 20 h at RT and ceased the reaction by adding cold water. Sodium hydrogen carbonate used to neutralize the reaction mixture until it ceases the effervescence. Then, dialyzed the solution about 7 days against distilled water and concentrated under reduced pressure and lyophilized (Mellerowicz, 2013)

Phosphorylation: 50 mg of polysaccharide was suspended in 10 ml of N, N-Dimethylformamide (DMF) along with 4 g of urea and stirred for 1 h to homogenize the mixture. 50 mg of phosphoric acid was added and stirred for 4 h at 130°C. Reaction mixture was brought to RT, filtered, and washed with a mixture of 1-propanol, and distilled water followed by 0.1mM hydrochloric acid, and with distilled water. The obtained filtrate was lyophilized (Oshima et al., 2008).

Morphological and elemental changes: Morphological changes of the modified polysaccharides were studied under scanning electron microscopy (SEM), and the change

in the elemental composition was investigated with energy dispersive x-ray spectroscopy (EDS).

Functional group studies and PCA analysis: Chemical modifications of the polysaccharide further confirmed with functional group analysis by IR spectroscopy. Minor changes in the chemical modifications and major distributions of the different functional groups of the modified polysaccharides were studied under the statistical analysis of principal components. PC1 and PC2 were considered for the analysis (Szymanska-Chargot & Zdunek, 2013).

Biological significance: Change in the chemical structure always changes its biological activity. Antioxidant activity and cytotoxicity of the modified polysaccharides were studied with previous methods.

Results and discussions

Morphological and elemental analysis: Modified polysaccharides changed their morphology as they introduced by other functional groups. Sulfated, carboxymethylated polysaccharide looks like fabric like structure and flakes like structures are observed with acetylated and phosphorylated polysaccharides whereas the native form of ECP is crystalline like structure (Fig. 3.2.1). Change in the elemental composition of the carbon and oxygen of the native form by introducing acetyl and carboxymethyl groups are not much significant. However, the important observation can make from the composition that, decrease in the “C” element from native 51% to 48% and 45% of acetyl and carboxymethyl respectively propose the etherification of the polysaccharides. 8% of the sulfur element and 4% of phosphorous elements confirm the substitution of sulfate groups and phosphate groups on the polysaccharide (Fig. 3.2.2).

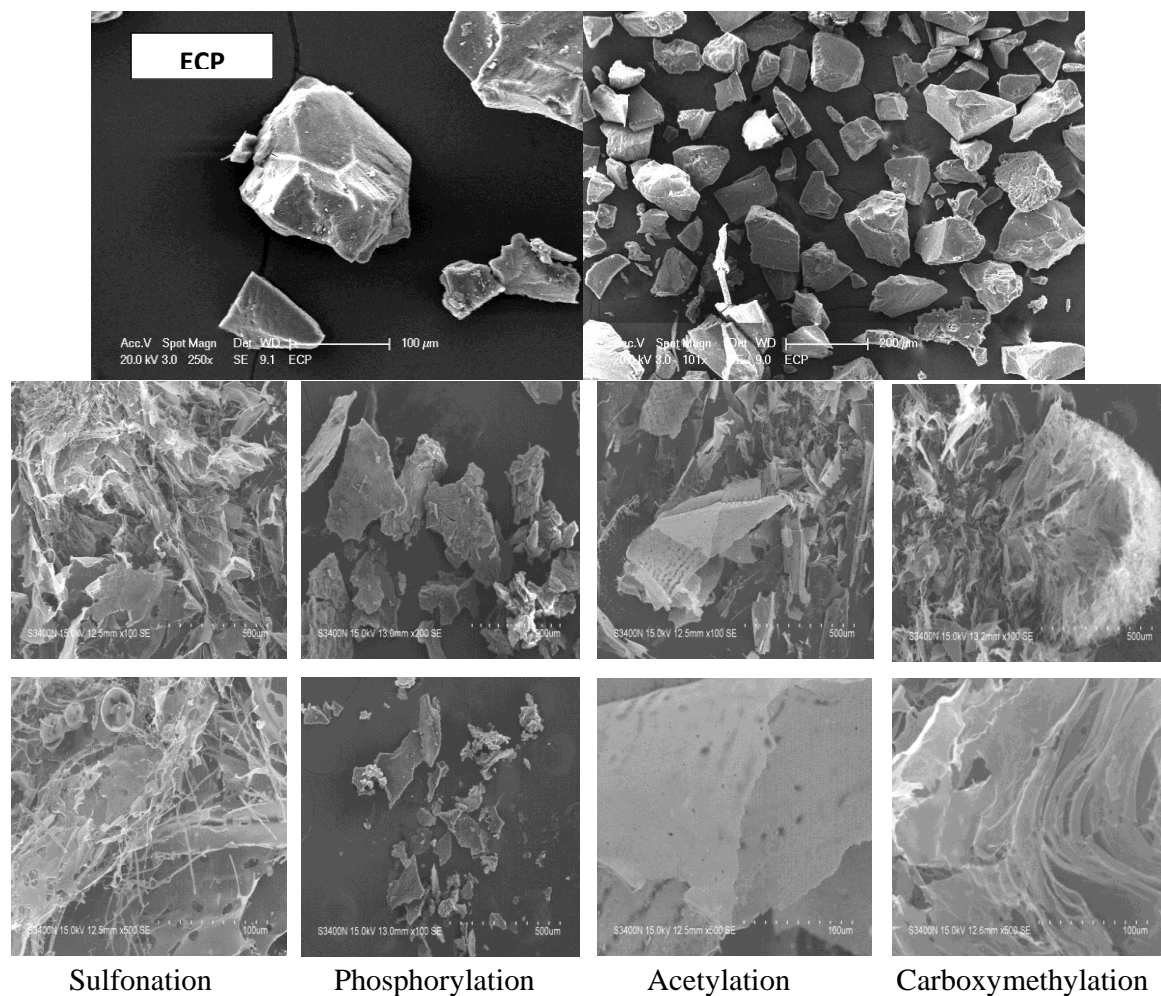


Fig. 3.2.1. Scanning Electron Microscopy images of native and modified polysaccharides

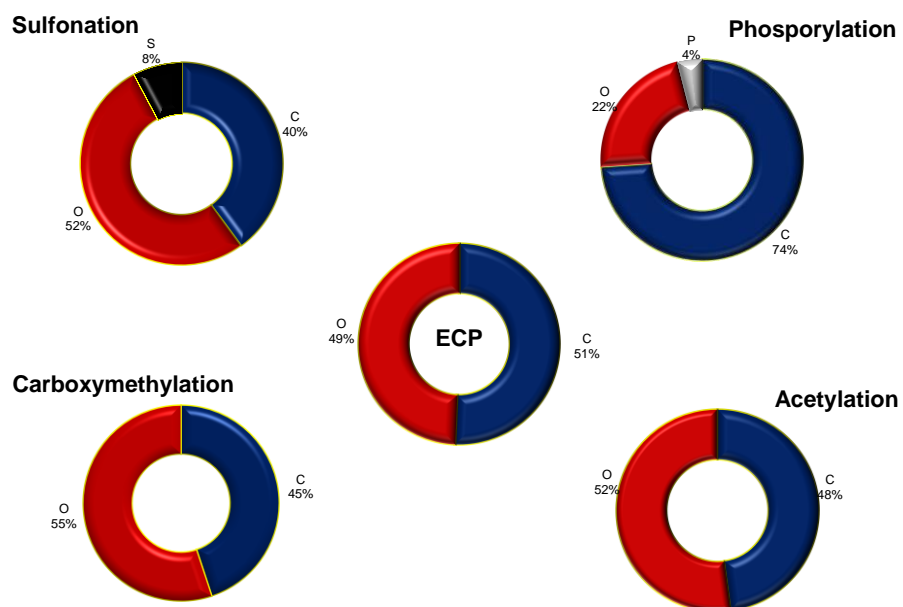


Fig. 3.2.2. Elemental composition of native and modified polysaccharide

Functional group studies and PCA analysis: An IR spectrum of the modified polysaccharides confirms the presence of relative functional groups respective to the modifications. Carrageenan, a natural sulfated polysaccharide obtained from marine algae. Sulfated polysaccharides are familiar with many biological activities like anti-tumour, antithrombin, anti-viral and mostly for anticoagulant activity (Yuan et al., 2005). Heparin know to be a natural anticoagulant, activity depends on the degree of sulphation of heparin (O Neill, 1956). A peak at $1260-1210\text{ cm}^{-1}$ confirms the sulphation of the polysaccharide along with sugar sulfate peaks around 800 cm^{-1} (Gomez-Ordonez & Ruperez, 2011). Phosphate of the polysaccharide gives P=O, P-O stretch vibrations at $1200-1100\text{ cm}^{-1}$. For carboxymethylation and acetylation, CO stretches at $1900-1550\text{ cm}^{-1}$ due to the carbonyl group and stretch at $1800-1740\text{ cm}^{-1}$ assigned to carboxylic groups. Similar stretches were noticed with respectively modified polysaccharide in comparison to native form (Fig. 3.2.3). Change in the intensity of OH stretch of alcoholic hydroxyl groups confirms the derivative forms of esters.

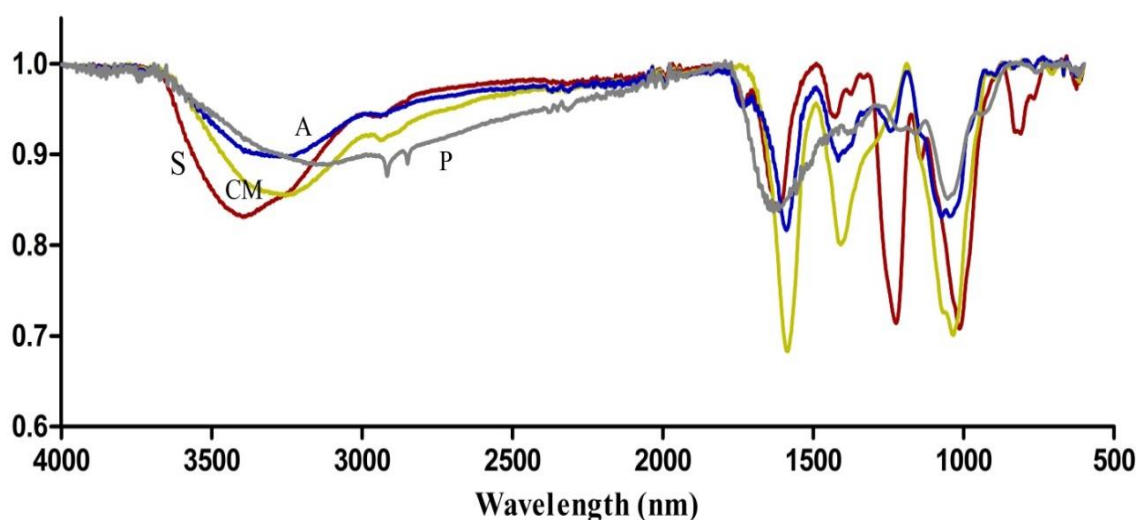


Fig. 3.2.3. ATR spectra of modified polysaccharides; A-Acetylated, CM- Carboxymethylated, S-Sulphated, P-Phosphorylated polysaccharides.

Multiple variation analysis is used to analyze the large set of data to limited variable interpretation like IR spectral data. PCA is one of the methods used to characterize the modified spectra from its native form. Total 75.68% of variation was observed with first 2 principal components PC1 and PC2 and considered to categorize the observations. A negative correlation was noticed with sulfated and phosphate forms whereas togetherness of native form with carboxymethylated and acetylated forms (Fig. 3.2.4A). Biplot shows the influence of variables on observations with particular stretch levels of vibrations at respective wavelengths (Fig.3.2.4B). The combination of the various observations concerning their modifications alters the variation to the 79.94% from 75.68%. (Fig. 3.2.4C).

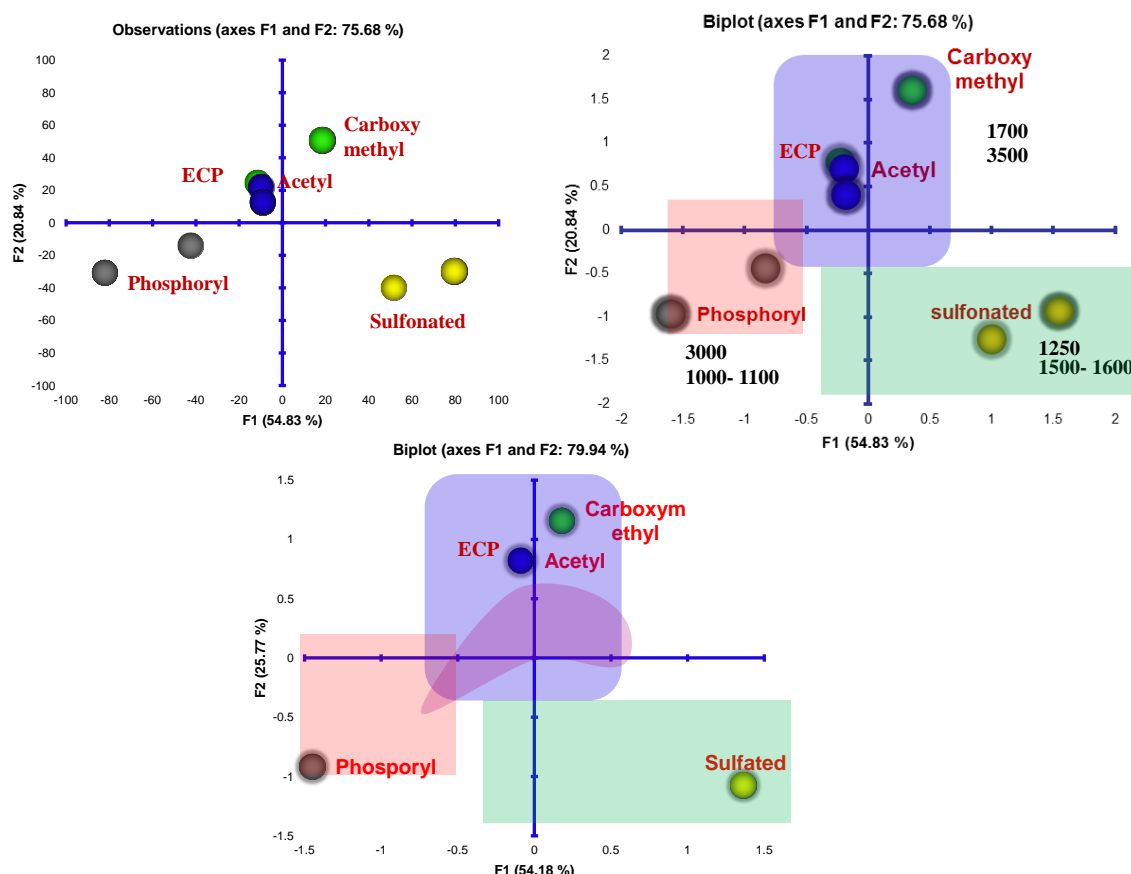


Fig. 3.2.4. Principal Components analysis of modified polysaccharide spectra; A) Plot of PC1 and PC2 of observations, B) Plot of PC1 and PC2 observations with variables, C) Average plot of observations.

Biological Significance: Structural changes always modify the biological activity. Structural changes employed to improve the physicochemical barriers like solubility, flow properties, and others. In addition, Biological significance also greatly mentioned. The addition of sulfate group improves the anticoagulant activity and antioxidant activity (O Neill, 1956). The ability of free radical scavenging activity of increased with the addition of carboxymethyl, sulfate and acetyl groups to the native form of polysaccharide. In contrast, phosphate introduced polysaccharide decreased its ability to scavenge the DPPH free radicals. A polysaccharide isolated from the plant does not show any toxicity towards macrophages. The same results were observed with carboxymethylated and acetylated ECP whereas cell death was noticed with sulfated and phosphorylated ECP (Fig. 3.2.5).

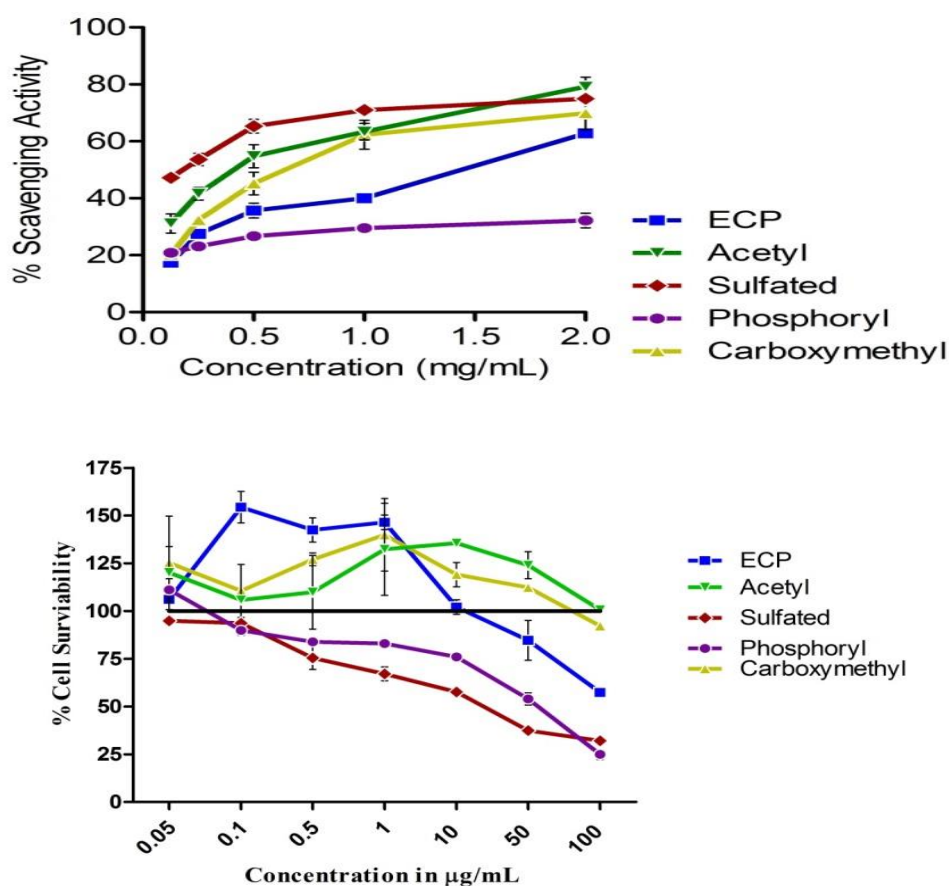


Fig. 3.2.5. The biological significance of Modified polysaccharides:
A) Antioxidant activity, B) Cytotoxic activity.

Conclusions

Change in the structural aspects of the molecules alters their nature towards physicochemical and biological properties. 8% of sulfate and 4% of phosphate groups introduced. Change in the elemental composition of C and O gives the alteration carboxymethyl and acetyl polysaccharides. Principal components analysis of ATR spectra discriminate the respective polysaccharides each other and with native form. Biological significance greatly altered with modified polysaccharides to native form.

CHAPTER IV

ISOLATION

&

CHARACTERIZATION OF LECTIN FROM LATEX OF *E.caducifolia*

Lectins have been used to study the immunology of Red Blood Cells. Seed lectins from Euphorbiaceae were studied by Stillmark (1888) initiates the immunology of lectins. Further elevates the lectin studies on isolation, purification and their biological applications. Lectins have been detected and isolated from different species of all living sources but the majority from angiosperms of the plant. *Azolla caroliana*, a pteridophyte has been reported for the presence of a lectin (Mellor et al., 1981). In gymnosperms, lectin activity was noticed with extracts of leaves and stems of *Taxus baccata* (Jermyn & Yeow, 1975) and *Cycas siamensis* megagametophyte walls extracts (Pettitt, 1977). Seeds of angiosperms are the major contributors of the lectins though several lectins have been also detected and isolated from various parts of the plants. For example, from leaves of *Aloe arborescens* (Suzuki et al., 1979) , *Dolichos biflorus* (Carter & Etzler, 1975), *Griffonia simplicifolia* (Lamb et al., 1983); Barks of *Robinia pseudoacacia* (Horejsi et al., 1978), *Sambucus nigra* (Shibuyas et al., 1987); Sieve saps of *Castanea sativa*, *Salix alba* (Gietl et al., 1979) Bulbs of *Galanthus nivalis* (Van Damme et al., 1987), *Leucojum aestivum* (Van Damme et al., 1988), *Narcissus spp.* (Van Damme & Peumans,1990),

Tulipa gesneriana (Oda & Minmi, 1986); Tubers of *Bryonia dioica* (Peumans et al., 1984), *Trichosanthes kirilowii* (Yeung et al., 1980).

Haemagglutination was also detected in the latex of the plants by Richet in 1909 from *Hura crepitans* and subsequently purified. Although the study of lectins from Euphorbiaceae was done in 1888 by Stillmark, the study was limited to few species later years. Further, the latex biochemistry and isolation of lectin from Euphorbiaceae was reported by Seshagirirao in 2001 (Seshagirirao & Prasad, 2001). In extension to the previous work done by Seshagirirao, this study emphasizes the isolation of high yield lectin, biochemical characterization, and biological activities.

Materials and Methods

Purification of Galactose-binding Protein:

Latex serum: The latex of *Euphorbia caducifolia* was collected and frozen in liquid nitrogen, frozen serum was thawed and centrifuged at 20,000 x g for 1 hour at 4°C. The liquid portion was separated and repeated the freezing, thawing and centrifugation process to get clear serum. The separated clear serum was used for experiments or stored below -20°C until required. The crude serum proteins were precipitated with cold acetone (overnight at -20°C) and centrifuged at 10,000 x g for 10 minutes. The protein pellet was dissolved in phosphate buffer saline and used for experiments or stored at -20°C.

Preparation of Affinity matrix: The Affinity matrices were prepared according to Seshagirirao method (Seshagirirao et al., 2005) In brief, 100 g of seeds of (Guar gum and *Leucaena leucocephala* seed gum) was blended at low speed for few seconds. The seed hulls were removed and kernels (60g) were milled into fine powder. The powder was further sieved through 1mm mesh. Next, the powder was mixed with an emulsion of Epichlorohydrin (99%) in 3 M NaOH. The mixture was incubated for 24 h at 40°C and

later for 6 h at 70°C. The formed matrices were soaked and washed several times with distilled water. Finally, the recovered gel was transferred to 10 mM phosphate buffer saline, pH 7.2 containing 0.02% of sodium azide.

Purification of Galactose-Binding Protein: The purification of galactose binding protein by affinity chromatography was performed at 4°C. 2 mL of latex serum (50µg) was loaded onto matrix column (3 X 7 cm, ca 50 mL), previously equilibrated with phosphate buffer saline pH 7.2 with a flow rate 15 mL/h. The column was washed with it until the effluent gets < 0.02 absorbance at 280 nm. The protein adsorbed was eluted with 0.2M lactose in Phosphate buffer saline. The purity of high protein fractions was verified by SDS-PAGE, pooled and dialysed against phosphate buffer saline.

Protein Estimation: Protein was estimated by Lowry et al., (1951) with minor modifications. Reagent A, was 4% Na₂CO₃ in 0.2 N NaOH, B was 1% CuSO₄, C was 2% Sodium potassium tartarate and D was 1 N Folin's reagent. The working reagent is a mixture of A, B, C in the ratio of 23:1:1 and used within 24 h of preparation. 1 mL of the working reagent was added to 1 mL of the protein sample, mixed well and allowed to stand for 10 min. 0.2 mL of reagent D was added rapidly while vortex the sample. After incubation for 30 min, absorbance was recorded at 750 nm. BSA (Fraction V) was used as standard protein (5-50 µg).

Carbohydrate Estimation and monosaccharide composition: Total carbohydrate was estimated using Dubois method (DuBois et al., 1956). The glycosylated carbohydrates to the protein were studied using GC-MS analysis. Lectin (50 µg) was hydrolysed with 4 M TFA by incubating for 6 h at 80°C. Hydrolysate was dried with methanol under speed vac. The residue was dissolved in pyridine and derivatized with silylating reagent (BSTFA and 1% TMCS) in 1:2 ratio at 40°C for 2h. Monosaccharide derivatives were extracted with

dichloromethane and analyzed for composition using GC-MS (Pitthard & Finch, 2001). Quantitized the monosaccharides using a set of standards.

Haemagglutinating activity: The haemagglutinating activity of the galactose binding protein was determined with minor modifications of (Lis & Sharon, 1972) method.

a) Preparation of Alsevere's solution: The Alsevere's solution was prepared by dissolving 2.05g of glucose, 0.8g of Sodium chloride in 80 mL of distilled water. The pH was adjusted to 6.1 with citric acid and the volume was made to 100 mL of distilled water. The solution was autoclaved, cooled and stored at 4°C.

b) Preparation of Erythrocytes: Venous whole blood was added to the equal volume of Alsever's solution. The blood suspension can be stored as long as two weeks in the Alsever's solution. The erythrocytes were isolated from the stock blood suspension by centrifugation at room temperature using a table top centrifuge (1000 x g for 5 min). The packed erythrocytes at the 1000 x g considered as 100%. The packed cells were washed with cold saline (0.9% NaCl) for 3-4 times (5 mL saline for each ml of packed erythrocytes) and finally, 4% cells were made in saline.

c) Preparation of trypsin treated erythrocytes: The erythrocytes were treated with trypsin on the day of the experiment. The 4% erythrocyte suspension incubated with 0.1% trypsin 1:250 (1000-1500 BAEE units/mg) for one hour at 37°C. The trypsin treated erythrocytes were washed 5-6 times with cold saline to remove the traces of trypsin and 4% cells were made and used for the experiments.

d) Agglutination assay: 100 µl of the protein sample (final conc. 1µg) was serially diluted in a microtitre plate and 100 µl of trypsinized 4% erythrocytes were added to each well. The agglutination was observed visually after incubation of the plate for one hour at

37°C. The highest dilution which showed positive haemagglutination was considered as the Titre. The amount of protein presents in this dilution represents the minimum quantity of protein required for agglutination and is defined as one unit. The specific activity is the number of units per mg of protein.

Sugar Inhibition assay: 50 µl (0.4 M) of serially diluted carbohydrate solution was mixed with 50 µl protein containing 8 haemagglutination units in a microtitre plate and incubated at room temperature for 30 min. 100 µl of 4% trypsinised human O group erythrocyte suspension was added to the incubated solution for 1 h at 37°C. The inhibition concentration of the sugar was recorded as the minimum concentration of sugar required for complete inhibition of 2 haemagglutinating units with 2% of the erythrocytes.

Temperature and pH stability: The lectin (50 µg) was incubated at different temperatures (4°C, 30°C, 40°C, 50°C, 60°C, 70°C, 80°C and 90°C) for 10 minutes and was brought to room temperature. The ability of haemagglutination was tested to determine the temperature. The lectin (50 µg) was incubated at different pH (5, 6, 7, 8, 9 and 10) for 30 minutes and the haemagglutination activity was tested to determine optimum pH.

CD spectroscopy studies: Circular dichroism spectra were recorded to study the secondary confirmations of the lectin using JASCO spectrometer. The lectin was placed in the 2 mm Quartz cuvette with a concentration of 2 HU of protein and recorded spectra. Spectra measured for Far UV region (250-190 nm) and Near UV region (300-250 nm) with 50 nm per minute and accumulated 3 scans. The secondary structure was analyzed using Dichroweb online software. In addition, the effect of the temperature and pH on the unfolding of the lectin was also studied.

Binding Studies: *E. caducifolia* lectin is a galactose specific lectin. The ability of its affinity to the sugar was studied using UV, CD and Fluorescence spectroscopy. 2 mM of lactose was used for the study. Lectin (OD<0.1) was incubated with lactose and studied the binding ability by recording the spectra 300-250 for UV spectrum and 250 -190 for CD spectrum. And also, change in the intrinsic fluorescence of the lectin was studied with the addition of gradient concentrations of lactose (2mM- 6mM) to the lectin and change on the intrinsic fluorescence was recorded using JASCO spectrometer. The lectin was excited at 285 nm with fixed bandwidth 5 nm. Emission was recorded from 300-400 nm with same bandwidth up to 3 scan repetitions. The binding ability of the lectin was calculated using stern-volmer equation.

Biological activities

MTT Assay: RAW 264.7 was used for the experiment. The 1×10^4 cells were inoculated and incubated for 24 h in the CO₂ incubator. The lectin was added to the cells at different concentrations and incubated for different time points. After incubation, MTT, 5 mg/mL was added and incubated in the dark for 4 h. Crystallized formazan was extracted with DMSO and estimated at 590nm of Absorbance. Cell proliferation was calculated with control cells (without lectin) (Khan et al., 2006).

Mouse monocyte macrophages (J774a.1), Human cervical cancer cells (HeLa) and Human breast cancer cells (MCF-7) were used for the experiment. The cells (10,000) were inoculated in 96 well plates and incubated for 24 h in the CO₂ incubator. Then, the cells were treated with different concentrations protein and incubated for different time points. After time points, 3(4,5-dimethyl thiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) was added as 5 mg/mL and incubated further for 4 h. 100 ul of DMSO was added to the wells and absorbance taken at 590nm.

Adherent capacity Assay, Cell Aggregation Assay: The cells (1×10^4 /well) were seeded along with protein (50 $\mu\text{g}/\text{ml}$) in 96 well plates and incubated for 24 h. After incubation, media was removed and tapped on a paper towel to remove loosely attached cells. The cells were stained with 0.1% crystal violet in phosphate buffer saline for 10 min. Further, the plates were washed with water and dried at room temperature. The well images were taken by the microscope and counted the cells aggregations by using the ImageJ software. The adherent capacity was determined by crystal violet quantification at 595 nm.

Cell migration Assay: Monolayer cells of HeLa cells were grown to confluence in a 12 well plate. A scratch was created by using PCR micro tip in the middle of the monolayer well. Detached cells were removed by washing with incomplete media twice. Then the cells were treated with EC lectin with 50, 100 and 200 $\mu\text{g}/\text{ml}$ and untreated was considered as control. The plate was incubated and pictures of the gap area were taken at different time points to observe the migration of the cells towards the gap.

Results and Discussions

Purification of Galactose Binding Lectin: Lectins are glycoproteins with carbohydrate moieties and have the ability to agglutinate the blood. Lectins show specificity towards various sugars. Galactose-binding lectins are specific to the galactose and their derivatives as the name says. Galactose-binding lectins can be purified by affinity chromatography using galactose-specific matrices. The presence of galactose-specific lectins in *Euphorbia caducifolia* and other *Euphorbia* species were reported by Seshagirao and Prasad in 2001. Lectins from *Euphorbia caducifolia* latex was isolated by applying on galactose affinity matrices. Affinity matrices were prepared and evaluated with snake gourd lectin. Applied crude serum was eluted with 0.2 M Lactose and

collected the protein (Fig. 4.1). Two matrices were used to assess the yield of the protein. Purified protein with a yield of 60% was observed with CLGG matrix, and 82% of yield was noted with CLLSG (Table.4.1). CLLSG shows more affinity and gives high purification folds compared to the other matrices like Guar gum, cashew nut and egg matrices (Kottapalli Seshagirirao et al., 2005) and also with Sepharose 4B.

Table 4.1. The yield of galactose-binding protein from *Euphorbia caducifolia* latex on different affinity matrices.

| Purification step | Protein (mg) | Specific Activity | Total Activity (titre X mg) | Yield (%) |
|-------------------|--------------|-------------------|-----------------------------|-----------|
| Crude Protein | 50 | 2560 | 128000 | 100 |
| CLGG | 4.5 | 17066 | 76797 | 60 |
| CLLSG | 6.2 | 17066 | 105809.2 | 82 |

Previously, it has been reported the isolation of ECL using Sepharose 4B column with a yield of (%) which is more as observed with CLGG and less with CLLSG. It shows that the use of CLLSG for isolation or purification of galactose binding lectin is more efficient than another column (). Further purified protein was run over SDS-PAGE to check the lectin purity (Fig. 4.1 (insert)), and lyophilized until the studies its biochemical properties.

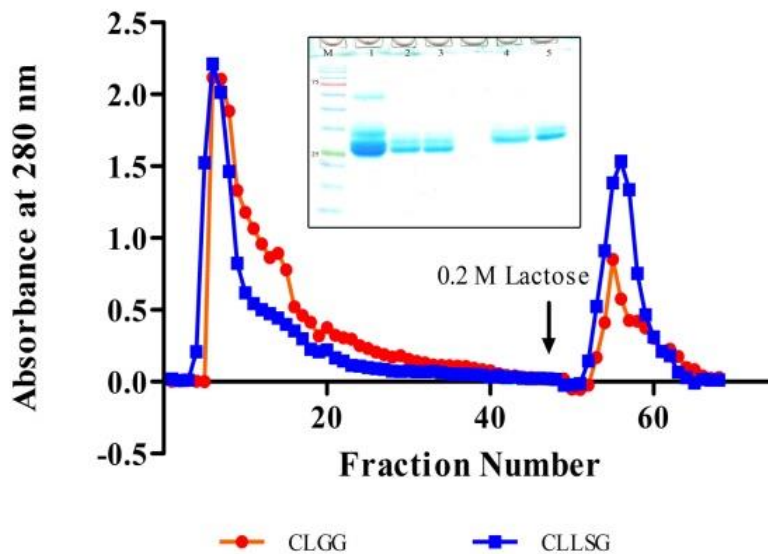


Fig. 4.1. Affinity Chromatography of *Euphorbia caducifolia* Latex serum on CLGG, CLLSG column (3X7 cms) at 4 °C. ; Insert SDS-PAGE of Purified Galactose-binding protein from *Euphorbia caducifolia* latex : M- Protein marker, 1- Crude protein, Purified protein 2-CLGG, 3- CLLSG, and purified protein with reducing agent β -Mercapto14- CLGG, 5- CLLSG.

Haemagglutinating activity and carbohydrate specificity: The haemagglutination activity of ECL was observed more with O compared to other blood groups (Table.4.2). ECL did not show much difference in agglutinating trypsinized red blood cells of Rh positive A, B and AB groups but agglutinated almost 6 folds with Rh O group. It indicates the specificity of the ECL towards the O type as other *Euphorbia* spp. (Lynn & Clevette-Radford, 1986b).

All lectin are carbohydrate specific and inhibition the agglutination of RBC in the presence of their specific sugar where binding sites of the lectin blocked. ECL was observed specificity with different carbohydrates (Table.4.3), galactose and galactose derivatives (Table.4.4). We noticed the delay in agglutination with melibiose and raffinose where the galactose in 1→6 glycosidic linkage but observed no agglutination with lactose where the glycosidic linkage is involved is the 1→4 type. Apart from the specificity of the lectin to the sugar, and also express its specificity with linkages of

specific monomer in oligosaccharides. O-Nitrophenyl galactoside also inhibits the agglutination as galactose majorly binds with lectin sites. Reports say that most of the latex lectin from Euphorbia are specific in galactose moiety either direct or its derivatives (Barbieri et al., 1983; Lynn & Clevette-Radford, 1986a).

Table 4.2. Haemagglutination of Lectin with trypsinized human Rh⁺ Erythrocytes.

| Erythrocyte | Specific Activity (U/mg) | |
|----------------|--------------------------|------------------|
| | Crude Protein | Purified Protein |
| A ⁺ | 400 | 2560 |
| B ⁺ | 400 | 2560 |
| AB | 400 | 2560 |
| O ⁺ | 2560 | 17066 |

Table 4.3. Sugar Inhibition Assay with different sugars and oligosaccharides.

| S.No | Carbohydrate | Agglutination Activity | Galactose present (Yes/No) | Saccharide |
|------|---------------------------------|------------------------|----------------------------|------------|
| 1 | Agar | Present | Yes | Polymer |
| 2 | Agarose | Present | Yes | Polymer |
| 3 | Arabinose | Present | No | Mono |
| 4 | Dulcitol | Present | Yes | Mono |
| 5 | Fructose | Present | No | Mono |
| 6 | Fucose | Present | Deoxy Galactose | Mono |
| 7 | Galactosamine | Absent | Yes | Mono |
| 8 | Galactose | Absent | Yes | Mono |
| 9 | Glucose | Present | No | Mono |
| 10 | Inositol | Present | No | Mono |
| 11 | Mannose | Present | No | Mono |
| 12 | N-Acetyl galactosamine | Absent | Yes | Mono |
| 13 | N-Acetyl glucosamine | Present | No | Mono |
| 14 | Rhamnose | Present | No | Mono |
| 15 | Salicin | Present | No | Mono |
| 16 | Xylose | Present | No | Mono |
| 17 | Lactose | Absent | Yes | Di |
| 18 | Melibiose | Present (Slow) | Yes | Di |
| 19 | Raffinose | Present (Slow) | Yes | Tri |
| 20 | O-Nitrophenyl galactopyranoside | Absent | Yes | Derv. |

Table 4.4. Sugar Inhibition Assay with different Galactose derivatives and their MIC .

| Carbohydrate | Agglutination | Galactose present (Yes/No) | Saccharide | MIC (mM) |
|---------------------------------|----------------|----------------------------|------------|----------|
| Lactose | Absent | Yes | Di | 3.13 |
| Galactose | Absent | Yes | Mono | 25 |
| O-Nitrophenyl galactopyranoside | Absent | Yes | Derv. | 25 |
| Galactosamine | Absent | Yes | Mono | 50 |
| N-Acetyl galactosamine | Absent | Yes | Mono | 50 |
| Melibiose | Present (Slow) | Yes | Di | - |
| Raffinose | Present (Slow) | Yes | Tri | - |

pH and temperature effect: Lectin isolated was observed for the influence of pH and temperature on activity (Fig. 4.2). ECL shows a broad range of activity in pH as well as temperature. Lectin activity notified at pH range 6.0-9.0 but showed maximum activity at pH 8.0. Maximum activity or association constants of the several lectins were found at neutral or slightly alkaline pH (Duk & Lisowska, 1984). Like most of the Euphorbiaceae latex lectins, ECL also inactive above pH 10.0. Temperature plays a good role on activity as well as the stability of a protein apart from pH as physical conditions. *E. caducifolia* lectin is heat stable and shown maximum activity at 60°C and 70°C, and gradually activity ceased above 90°C.

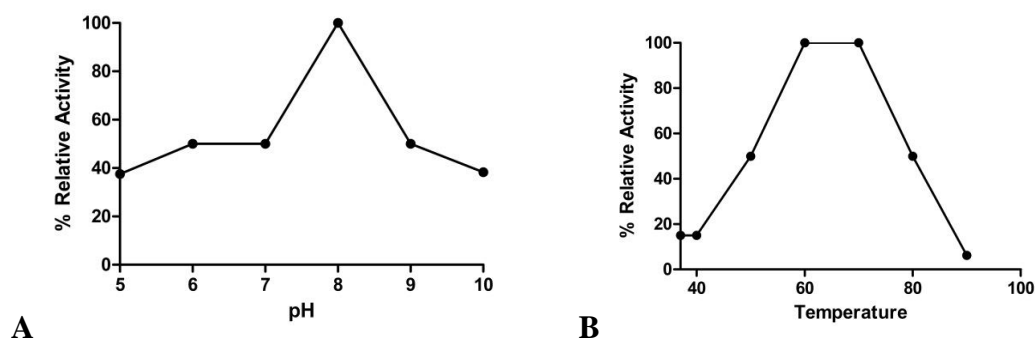


Fig. 4.2. Physicochemical parameters influence on Lectin haemagglutinating activity (A) pH, (B) Temperature

Glycoprotein characterization: Glycoprotein character of the lectin was identified by the Schiff periodic base on PAGE. The carbohydrate portion of the lectin was estimated to be 11.82% and this was almost nearer to the previous reports. Carbohydrate content of the lectin was very similar to the amount present in other species of the genus. Further studies revealed the presence of monosaccharides covalently bound to the protein (Table.4.5). Sugars present in the protein allude to structural conformations of protein as well as its biological importance.

Table 4.5. Sugar composition in the Lectin

| Sugar | Residue/Molecule |
|-----------------|------------------|
| Fuc | 11.3 |
| Gal | 8.3 |
| Glu | 0.6 |
| Man | 8.9 |
| GLcNac | 12.7 |
| Xyl | 1.4 |
| Total sugar (%) | 11.4 |

CD Spectroscopy Studies: Structural confirmation of the lectin was studied by Circular Dichroism spectrometry. Spectra of the native lectin in the near UV and far UV was recorded at 25°C. Far UV spectrum observed with 2 positive peaks at 195 and 230 nm, and 2 negative peaks at 210 and 218 nm. Secondary structure information of the lectin was analyzed by software available at DICHROMEWEB. The secondary structural content obtained are 1% of regular α helix and 6% of disordered α helix (Total α helix of 7%), 21% of regular β sheet and 14% of disordered β sheet (Total β sheet of 35%), 24% β turns and 34% of unordered structures (Fig. 4.3). Overall CD spectrum of the lectin indicates that the predominant of β sheet protein to minor α helical structure of the protein. Influence of the temperature and pH on the secondary structure of ECL also studied using CD spectroscopy. As affecting the RBC agglutination, temperature, as well as pH, alters the structure of the lectin.

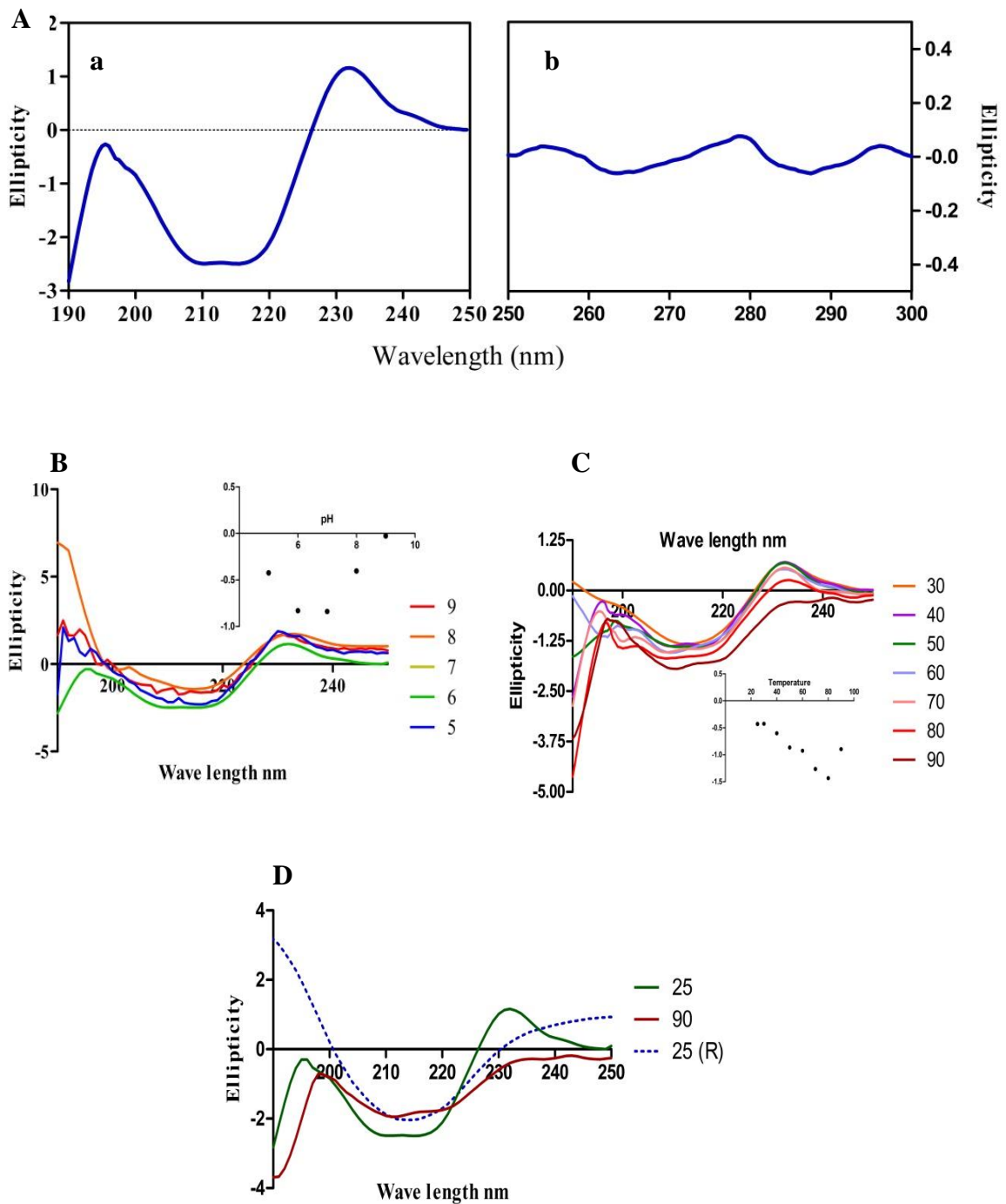


Fig. 4.3. A) Secondary confirmation of the lectin by CD spectrometry, a) Far UV, b) Near UV; Influence of B) pH and C) Temperature on structure, D) Spectrum of cooled lectin after incubated at 90°C.

Binding studies: Galactose specific lectin showed more affinity towards lactose than galactose. Thus, the binding affinity of lectin with lactose was studied with various spectrometers like UV, CD and Fluorescence. The decrease in the absorption of lectin

with gradient addition of lactose shows the binding efficacy of lectin (Fig. 4.4A). Change in the secondary structure of the lectin with the addition of the lactose reveals the influence on binding site of the lectin. Binding of lactose to lectin changed its conformations to more β configuration than native form (Fig. 4.4B). Further, supports the binding efficacy of lectin by fluorescence quenching studies. The native form of tryptophan, phenylalanine, and tyrosine of the protein naturally emits fluorescence as it excited at 285 nm called intrinsic fluorescence. This change in the intrinsic fluorescence was observed with lactose (Fig. 4.4C). Quenching of intrinsic fluorescence was studied using the Stern-Volmer equation and found the Stern-Volmer binding constant to be $1.17 \times 10^3 \text{ M}^{-1}$ with 0.96 regression.

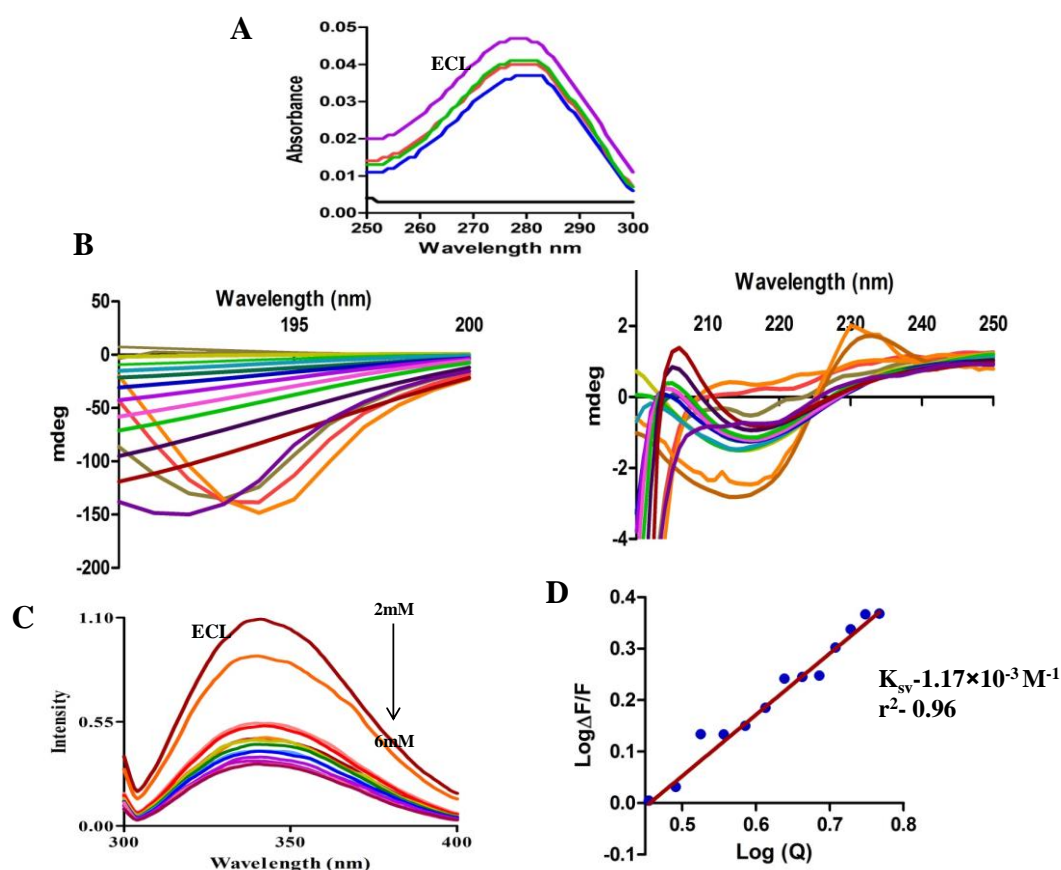
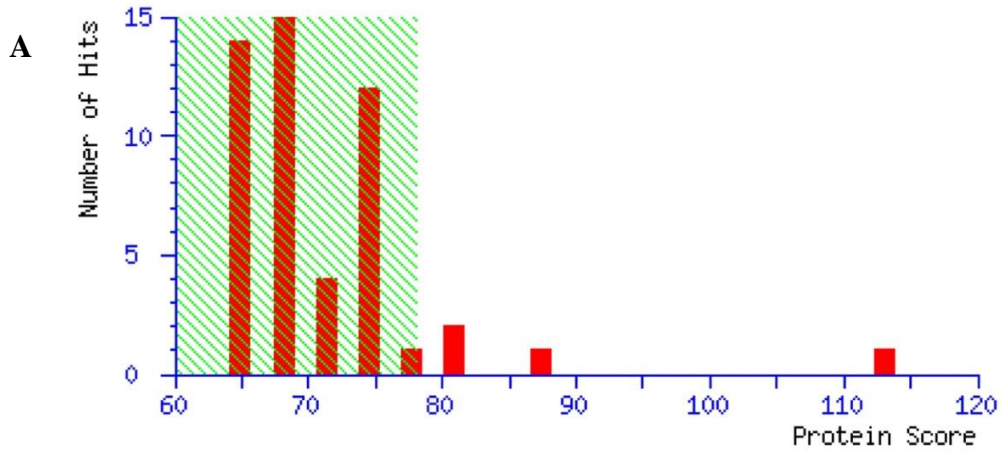


Fig. 4.4. Binding studies, A) UV spectrum, B) CD spectrum studies, C) Fluorescence quenching, D) Stern-volmer plot

Partial peptide sequencing: Purified lectin was subjected to peptide sequence to discover the fragments or domains of the protein. Peptides gained many attentions towards medical and biotechnology aspects, and over 7000 peptides have been identified as they play important roles in human physiology (Fosgerau & Hoffmann, 2015). Much more, peptide research on therapeutic experiencing a commercial renaissance. 3 Major and significant peptides were identified from a total of 8 obtained peptides from *Euphorbia caducifolia* lectin (Fig. 4.5). 3 peptides were sequenced for amino acids by MS/MS analysis and blasted the sequences for identification through NCBI protein database. Not many reports have been found on *Euphorbia* latex or seed lectin protein sequence or peptide sequence. Previously, only peptide sequence reported from *E. characias* and *E. marginata* lectin (Stirpe et al., 1993) and total protein sequence of a mannose-binding lectin from *E. tirucalli* has been reported recently (Kitajima et al., 2016). The peptide sequence of *Euphorbia* lectins was not identical and didn't show any conservative domains of the family (Fig. 4.5).

Hemolytic activity: Aside of agglutination, prolong incubation of RBC with latex serum exhibits the hemolytic activity. *Euphorbia caducifolia* lectin did not show any hemolytic activity with RBC used in haemagglutination assay. Some of the seed lectins like *Sterculia foetida* L and *Clitoria fairchildiana* were reported for their hemolytic activity as most of the lectins in particular latex lectin does not display such activity. Absence of haemolytic activity of the drugs or molecules imparts therapeutic attention as they does not harm the biological system. Lectins are using in various therapeutic application due to non-haemolytic nature as like ECL.

Cell proliferation and Anti-tumour activity: Lectins gained attention because of their proliferative activity on different cell types due to cell surface carbohydrate recognition variabilities (Ashraf & Khan, 2003). *Euphorbia* latex lectins can act as mitogens which



B

E. characias

1 sesytpisgp ngxvdvk

E. marginata

1 aypgshisgp ngfxmdvk

E. tirucalli

1 mhkpiilsls lffsaisfia ealvppsetf qyvnagdfgd yiveyanyr vldpfaqpfq
 61 lcfynttpne ytlalrmgtv rseslmrww eanrakpvre natvtfsedg nlvladadgt
 121 iawqtntsnk gvvgfkllpn gnmvlydsng gfvwqsfdyp tdtllvgqsl klgaatkivs
 181 rasreenvng pyslemeett lslyykpns pnpliyfsfr dllsvsegpl kfvtlgpels
 241 leydrgtlil rkpkynttlt yrleidgnv rlhtyednad wsawevtytl fdrnswetec
 301 qlpqrvcnfg lceddqvac pspgllgws ksceaekvss cgvndfryye lkgvdhfssk
 361 ysdgegpuki ddcgnkcted ckclgyfyht ksstcwivyd lntltkvdns thlafikapi
 421 kvvlsvdi

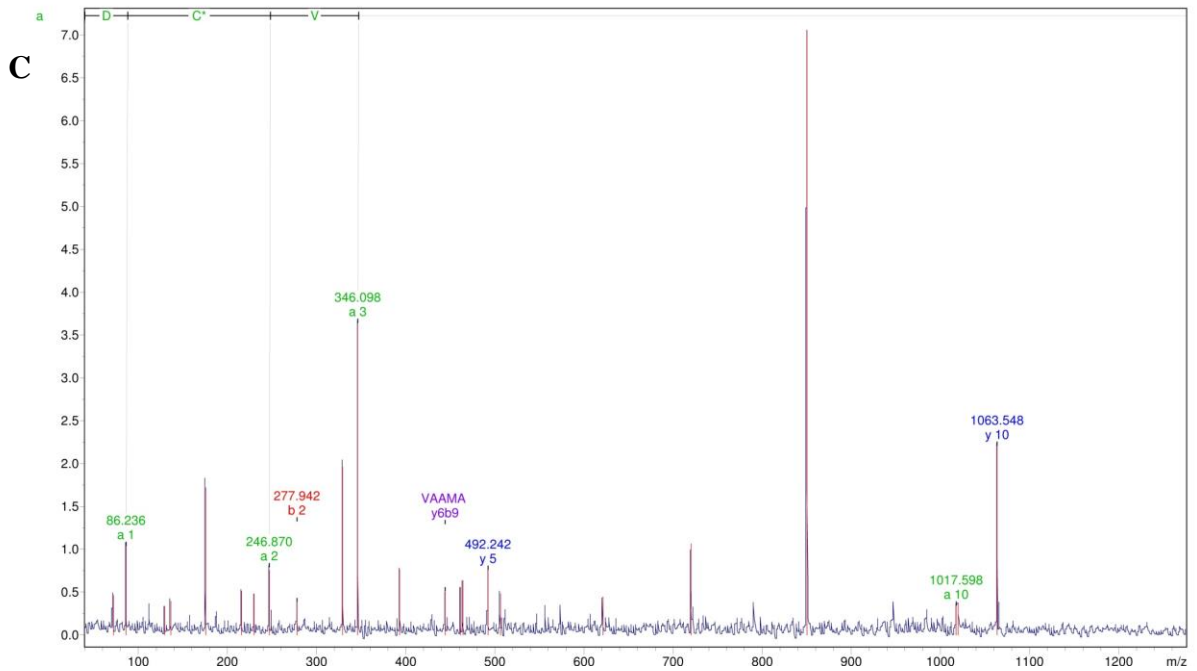
ldcvvvaamak

E. caducifolia

eaptwrldep aghgvnmhr

iiianaldvnm glnvlglrhr

Abs. Int. * 1000



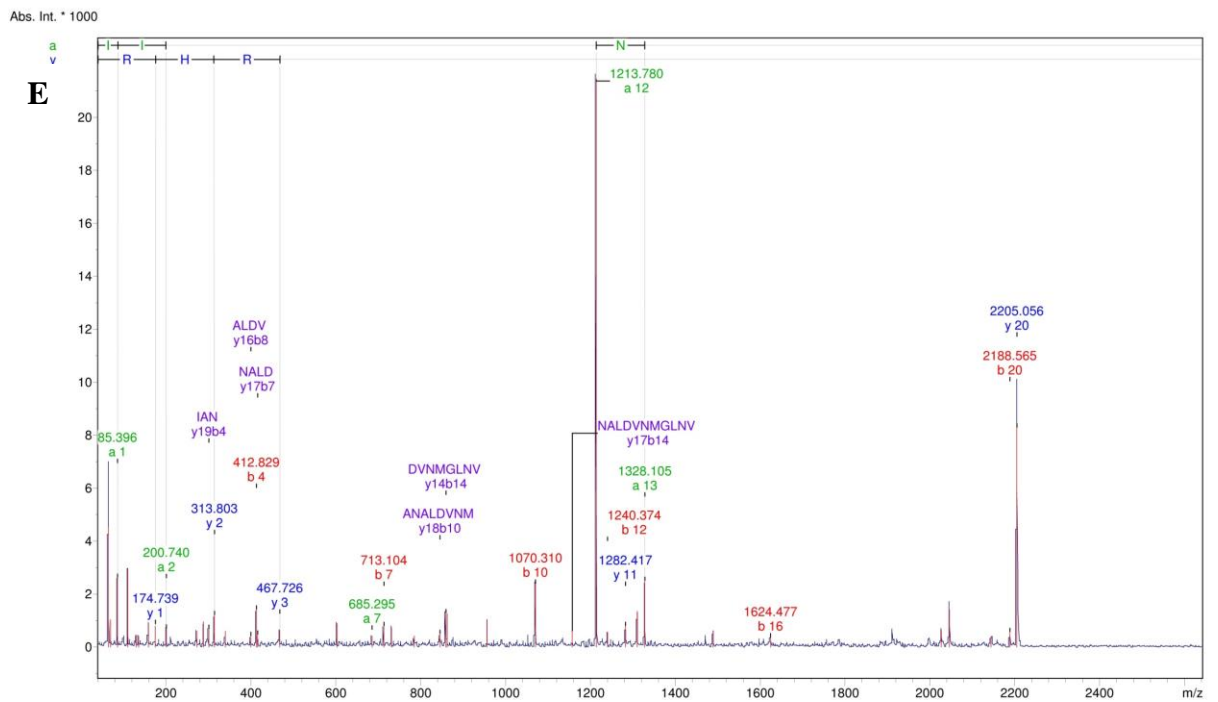
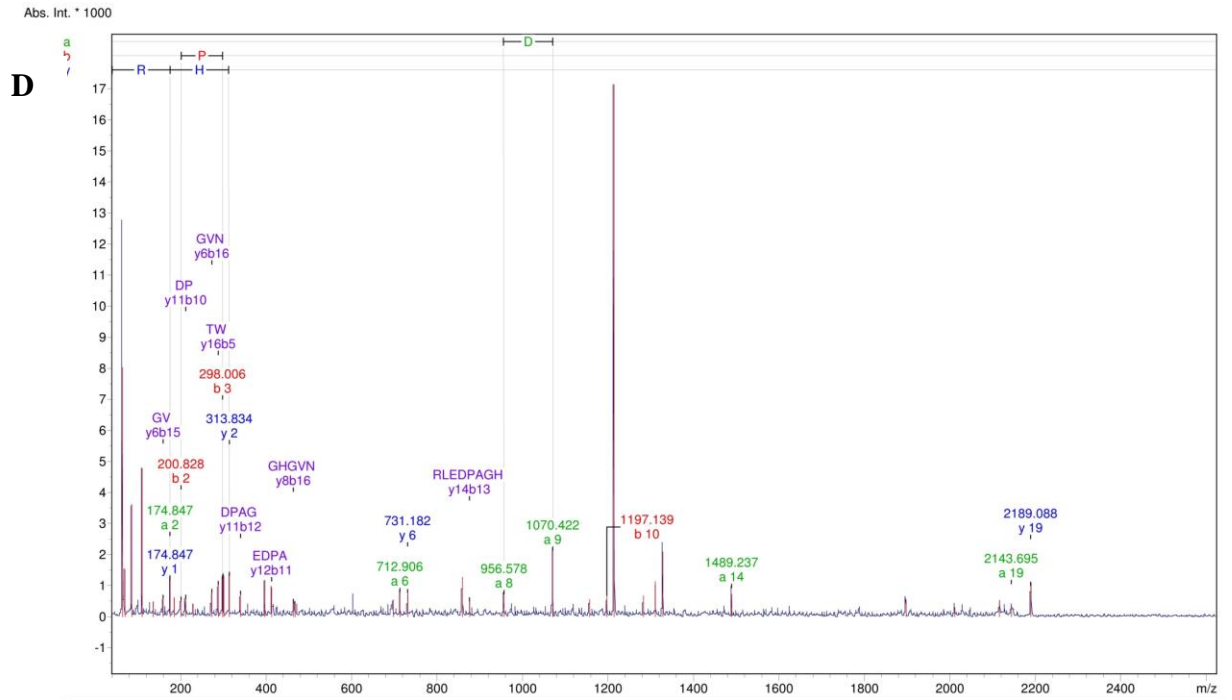


Fig. 4.5. Partial peptide sequencing of Lectin, A) Protein score histogram (78*), B) Euphorbia lectins peptide sequences; Peptide mass spectrum of masses C) 1063, D) 2189, E) 2205.

* Protein score greater than 78 are significant (p<0.5)

induces the mitosis of the cells. Latex lectins from *E. nerifolia* and *E.marginata* were reported for their mitogenic activity (Stirpe et al., 1993). In similar *E. caducifolia* also exhibited a mitogenic activity which was studied by cell proliferation. Mitogens induce the mitotic division of the cell which ultimately leads to the cell proliferation. *E. caducifolia* lectin treated cells were shown an increase in the cell number at all time points and at all concentrations (Fig. 4.6). Cell number increased with increasing concentration of the lectin and this might be due to the mitogen activity of lectin. Maximum activity showed at 3 and 24 h of incubation but it slightly decreased at 100 µg/ml concentration in relative to the other time points after 24 h of treatment. Proliferation ranged in between 0-50% more as compared to the untreated cells (control) and it considered as ECL has the ability to act as a Mitogen. The diverse role of lectins in cell proliferation and differentiation is still undelivered.

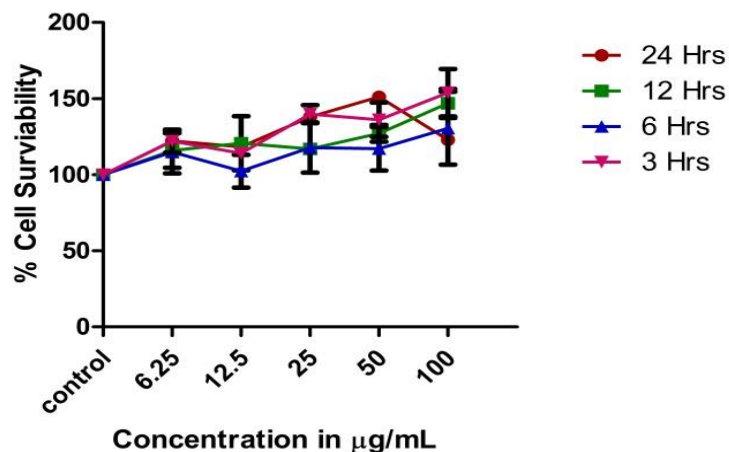


Fig. 4.6. Cell Proliferation assay on RAW 264.7 cell lines at different time points

Various chemicals and metabolites established their role in inhibiting tumour growth and its metastasis. Likewise, many lectins were reported as anti-tumour agents in treating different cancers. They are playing an important role in inhibiting tumour growth and metastasis by involving or influencing the autophagy and apoptosis (Yau et al., 2015).

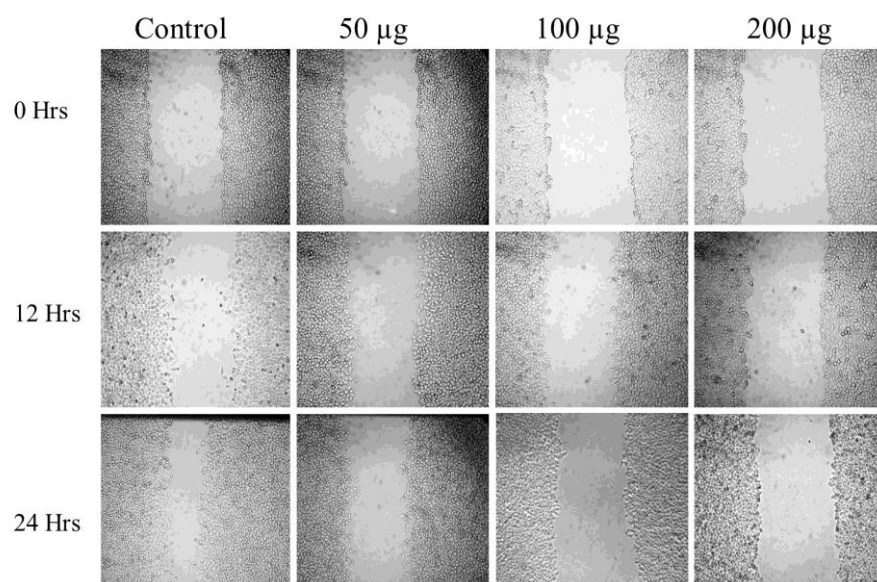
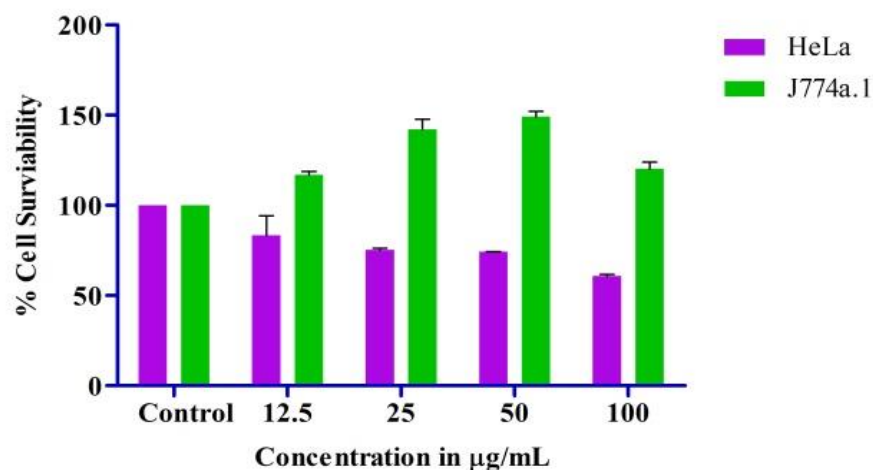


Fig. 4.7. Anti-tumour activity on HeLa cells after 24 h of treatment and Inhibition of HeLa cells migration

Eutirucallin, a latex lectin from *Euphorbia tirucalli* exhibited potential anti-tumour activity against various cancer cell lines (Santiago et al., 2017). ECL also exhibited anti-tumour activity against HeLa cell lines and inhibited cell migration (Fig. 4.7). Cell proliferation was observed with non-tumour cells (J774A.1) whereas cell death was noticed in tumour cells (HeLa). Anti-tumour activity was observed more as increasing the concentration where the inverse effect was shown with non-tumour cells.

The ability of cell migration was also ceased by lectin with all concentrations and exhibited maximum anti-tumour activity at 200 µg/ml. These results promise the anti-tumour activity of the lectin from *E. caducifolia* latex.

Adherent capacity Assay, Cell Aggregation Assay: Lectin shows an affinity towards carbohydrates. Thus agglutinate the RBC by recognizing surface carbohydrates. In the similar ability of lectin aggregation was evaluated with different cell lines (Fig. 4.8). More percentage of aggregation was observed with J774a.1 and MCF-7 cells in comparison to HeLa cells. This might be due to over expression of mucin in J774a.1 and MCF-7. In addition adherent capacity of the cells was also seen similar in both J774a.1 and MCF-7. But it shows more adherent capacity with HeLa cells 2 fold with control.

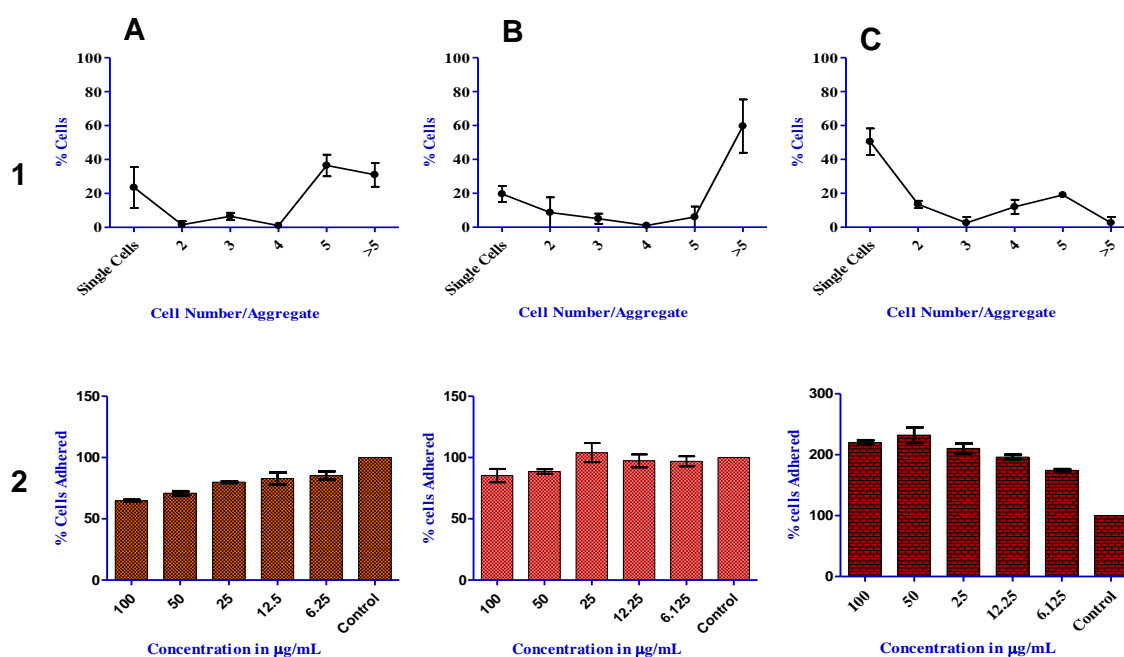


Fig. 4.8. 1) Cell Aggregation Capacity and 2) Adherent capacity of lectin on A) J774a.1, B) MCF 7, C) HeLa cells

Conclusions

E. caducifolia lectin is Galactose specific lectin with mitogenic nature and anti-tumor activity. Lectin is predominant in β conformation and active at pH 6-8 and 60-70°C.

CHAPTER V

ISOLATION

&

CHARACTERIZATION OF STARCH GRANULES FROM LATEX OF *E.caducifolia*

Starch is a major carbohydrate reserve in plants. Starch is stored in tubers and seeds of the plants as granules. It is made up of amylose and branched amylopectin. The granules are water insoluble, dense and quasicrystalline in nature. Physical parameters of the starch depend on the contents of amylose and amylopectin present. Granule structure of the starch under genetic control, therefore the starch granules from different parts of the same plant may differ in their structure. Czaja has reported structure and classification of the starch granules in vascular plants in 1969 and 1978 (Czaja, 1969, 1978). Characteristic shapes of Euphorbiaceae latex starch granules were reported by Rafn (1798) , Groom (1889), Molisch (1901), Solereeder (1908), P. Mahlberg (1973 & 1975), Seshagirirao and Prasad (1988).

Materials and Methods

Isolation and Purification: The latex of *Euphorbia caducifolia* was collected and frozen in liquid nitrogen, frozen latex were thawed and centrifuged at 20,000 x g for 1 hour at 4°C. The supernatant was removed and sediments were dissolved in hexane to remove the fatty acids, oils and etc. The sediments remained were dried at room temperature. The

dried powder was washed with citrate buffer pH 6.0 for 2-3 times and suspended in water. From the suspension, the starch grains were isolated by centrifugation in 2M sucrose at 2000 X g for 30 min. the sediment starch grains were washed with 0.2 M phosphate buffer pH 7.2 and distilled water for 3 times (K. Seshagirirao & Prasad, 1988). The prepared starch grains were used for microscopic studies and Physiochemical properties.

Microscopic studies: Latex starch grains were studied with a light microscope. Tincture iodine was used as a stain. Scanning Electron Microscope studies were done by transferring the grains to metallic stubs, dried in a desiccator. The starch grains were coated with Palladium-gold (Pd/Au) and observed under SEM at 15-20 kV (P. Mahlberg, 1973) . The confocal laser scanning microscope was used to analyze structural architecture. A Leica 60x/1.4NA/oil immersion was used to study the grain structure. Ar⁺ laser was used as the excitation source and by using a set of PMTs, photoluminescence images were recorded (Manca et al., 2015). 2% Starch grains were prepared in distilled water containing 0.02% sodium azide as a preservative. Further, observed thinly sectioned starch granules under Transmission Electronic Microscope (Mamoru, Keiji, & Dexter, 1979). For sectioning, Starch granules were prefixed in fixative composed with glutaraldehyde (3%) along with sucrose (0.1 M) in 50 mM sodium cacodylate buffer at pH 7.5 and incubated in osmium tetroxide (2.5%) for 2.5 h. After the post fixation, granules were dehydrated with successive percentages of ethanol (50%, 70%, 80% and finally 100%). Dehydrated specimens were washed with acetone 3 times and embedded in resin and sectioned using the ultratome. Sectioned specimens were observed under TEM.

Composition

Blue value (BV) and amylose content: The blue value of the complex (Iodine-starch) was measured at 680 nm. In brief, starch was dissolved in acetate buffer pH 4.8 and

iodine and potassium iodide added to it and measured the blue value (BV) (Takeda, Takeda, & Hizukuri, 1986).

Infra Red spectroscopy analysis: The starch granules for analyzed for the functional groups referred to the standard potato starch. FT-IR analysis was performed by using Platinum –Alpha, a single reflection Attenuated Total Reflection(ATR) module. The spectrum of the starch granules was measured from 4000 cm^{-1} to 500 cm^{-1} .

X-ray diffraction Analysis: Starch granules are semi-crystalline in nature. The crystalline behavior of the granules was studied using X-ray powder diffractometer. The diffractogram of the starch was measured in 2Θ between 5 and 40° with 1° /min scan rate. The crystallinity of the starch was calculated as follows,

$$\text{DC (\%)} = A_c / (A_c + A_a)$$

DC- Degree of crystallinity, A_c - Total area under crystalline peaks, A_a - Area under amorphous peaks.

Thermal and retrogradation properties: Differential scanning calorimeter (DSC) used to study the thermal characteristics of the starch. Gelatinization of the starch was investigated under the natural condition and different water proportions. The pan was sealed after addition of water and equilibrated at room temperature for 2 hrs. The experiment was performed from 30 to 120°C with heating rate $5^{\circ}\text{C}/\text{min}$ and the empty sealed pan was used as a reference. After cooled down, the pan was kept at 4°C and -20°C for 1 week and rescanned for retrogradation properties. The retrogradation percentage was calculated by:

$$\text{R\%} = \Delta H_r / \Delta H_g \times 100$$

Solubility, Swelling Index and Turbidity: Starch granules were analyzed for its solubility in water along with swelling index according to a method of Li & Yeh (2001). The starch granules (db) was added to the water I a screw-capped tube and vortexed for 10 s. Incubated for 30 min at temperatures 50, 60, 70, 80 and 90°C with additional mixing at 2 min interval. Then, cooled to room temperature and centrifuged for 30 min at 2000 g. The cloudy layer and sediments were separated and dried to constant weight. The weight of supernatant (W_L) and sediments (W_S) was used to calculate the water solubility Index (WSI) and Swelling Index (SI) as follows (Kong et al., 2016):

$$WSI = W_L / 0.1 \times 100\%$$

$$SI = W_S / [0.1 \times (100\% - WSI)] \text{ (gg}^{-1}\text{)}$$

Turbidity was measured as mentioned by Lan et al. (2008). The water suspension of starch was boiled for 1 hrs with constant stirring and cooled to 30°C. The turbidity of the solution was determined with Chemito UV2100 spectrophotometer at 620 nm against blank (distilled water) (Reis, Cristiana, & Beirão-da-costa, 2012)

Lintnerization: The starch granules were subjected to mild acid hydrolysis with 2.5 N HCl for 5 days at 30°C and aliquots of the treated samples at different time points was diluted and centrifuged. The supernatants were analyzed for sugars released at various temperatures and time points (Bertoft, 2004).

Results and Discussions

Microscopic analysis: Starch granules from the Euphorbia latex is diverge in shape and structure in comparison to other starch grains like rice, maize, corn and other. The characteristic starch grains were isolated from latex and analyzed under various microscopes. Dumb bell shaped starch was noticed with a light microscope and

confirmed in iodine staining (Fig. 5.1). Scanning electron microscope revealed the structure with the size of 40-60 μm where maximum size of starch grains 110 μm with potato (Fig. 5.2). Similar shape starches have been reported from *Euphorbia cattimandoo*, *E. milli*, *E. nerifolia*, *E. tirucalli* and also in its family members like *Pedilanthus tithymaloides* and *Synadenium grantii* (K. Seshagirirao & Prasad, 1988). Apposition growth was reported in starch granules. At early stages, starch granules are round or ovoid but on mature they stretch to characteristic shapes like discoid, rod or dumb bell (French, 1984) (Fig. 5.3). Molecular orientation of the starch architecture gives the birefringent pattern with polarized light. Dark Maltese cross was reported with avocado starch in 1987 by Kahn (Kahn, 1987) and in potato starch (French, 1984).

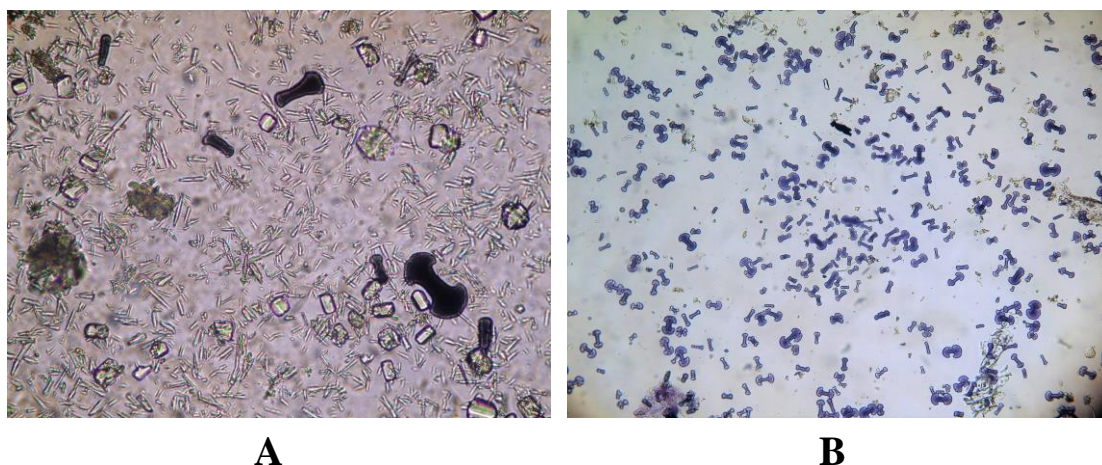


Fig. 5.1 Iodine starch granules from *Euphorbia caducifolia* latex
A) Starch granules in the latex along with calcium oxalate crystals, B) Purified starch

Confocal laser microscopy divulges the internal structure of the starch granules. Generally, starch granules are made up of amylose and amylopectin from the midpoint of the granule termed as “Hilum”. At lower levels of the structure, the alternate arrangement of amorphous and crystalline regions are made due to the higher level arrangement by clustering of side chains branching of radially arranged amylopectin. This appearance of the periodicity of starch granules is a universal feature and irrespective of botanical

sources(Daniel J Gallant, Bouchet, & Baldwin, 1997). The almost identical molecular architecture was noticed with *E. caducifolia* starch (Fig. 5.4). Alternative arrangements or growth rings formed were noted and the same was confirmed by TEM studies. TEM studies spotted the growth rings with 100-400 nm thick and smooth edges of starch (Fig. 5.5). It can observe central cavity of the starch granule which is universal.

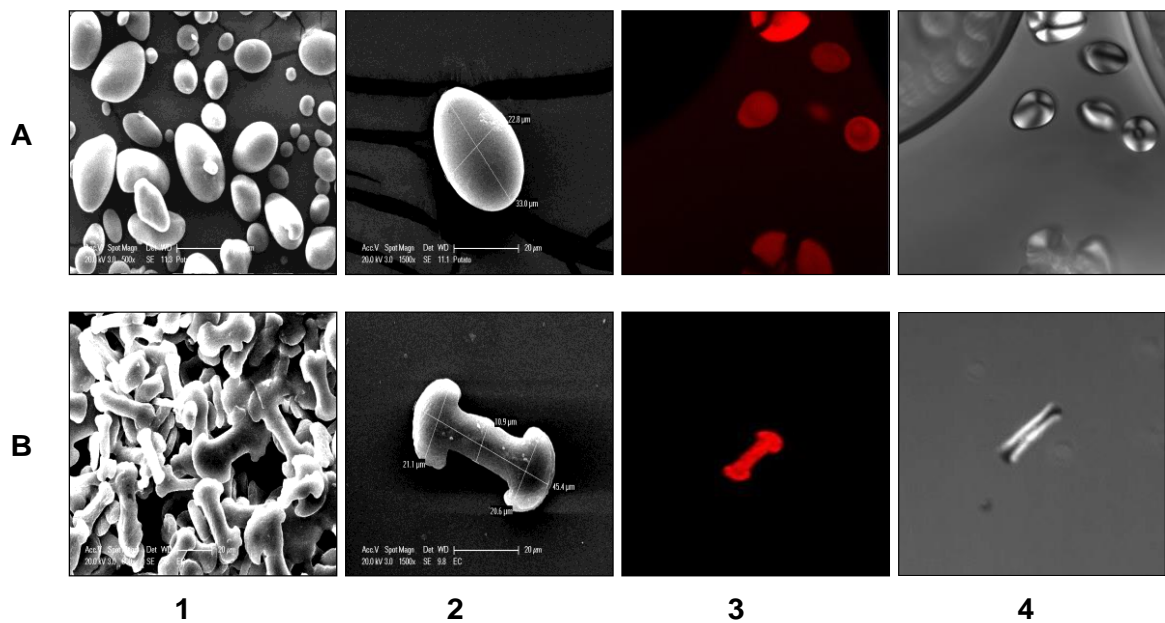


Fig. 5.2 Microscopic view of Starch granules.

A) Potato Starch, B) *Euphorbia caducifolia* Starch;
1,2-Scanning Electron microscope; 3, Confocal Laser microscopy, 4, Polarized light

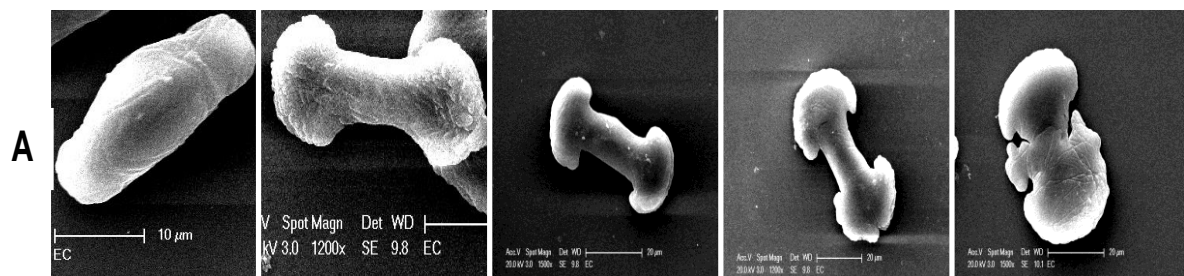


Fig. 5.3 Structural View of starch granules.

A) Growth process of starch granules (Left to Right) young to mature stages

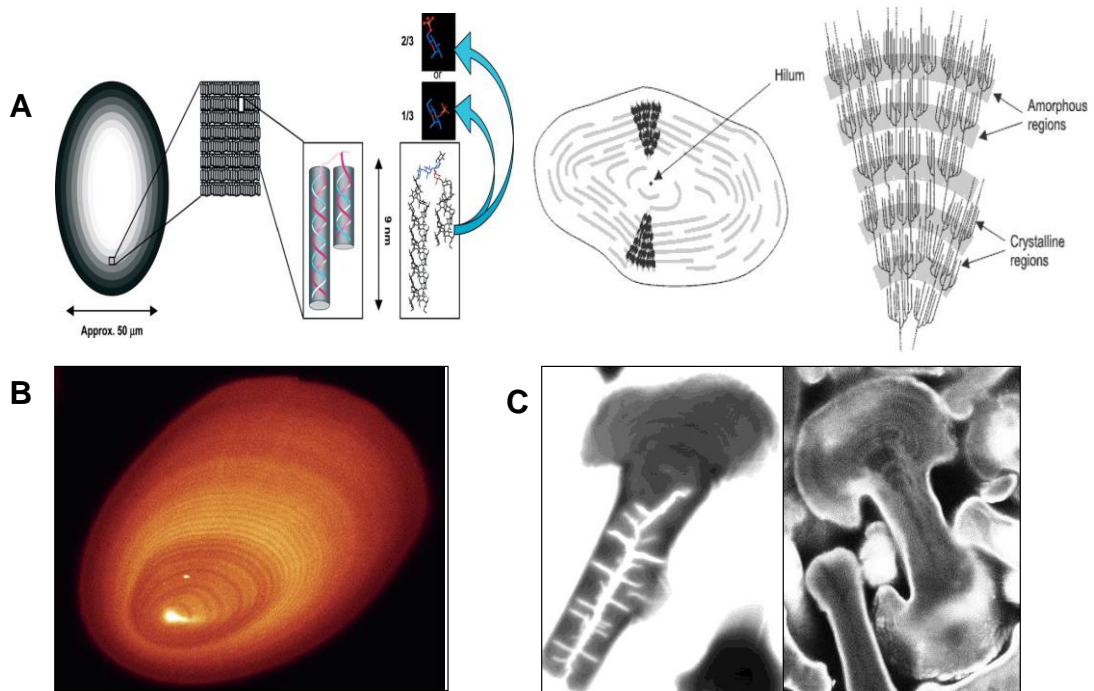


Fig. 5.4 Starch granule Architecture

A) Starch granule architecture model; B) Potato starch; C) *Euphorbia caducifolia* starch

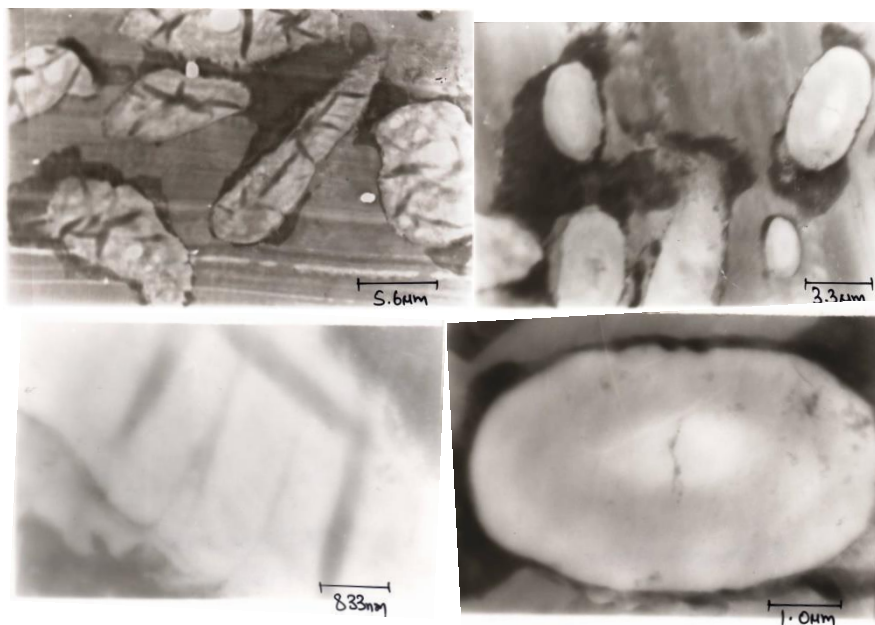


Fig. 5.5 Transmission Electron Microscopy of sectioned starch granule

Fragmentation: Fragmentation of the starch granules reveals the sub microscopic structure of the starch granule. Starch granules were fragmented with high impact pressure and High-speed trituration and observed its structure under scanning electron microscope. It observed that the radial arrangement of rod like micro fibrillary structure embedded in an amorphous matrix of starch granules. The appearance of fragmented starch granules is characteristic of the structure (Fig. 5.6). These observations were supported by previous studies of corn starch where similar arrangements of microfibrils in the amorphous region (Sterling & Spit, 1958).

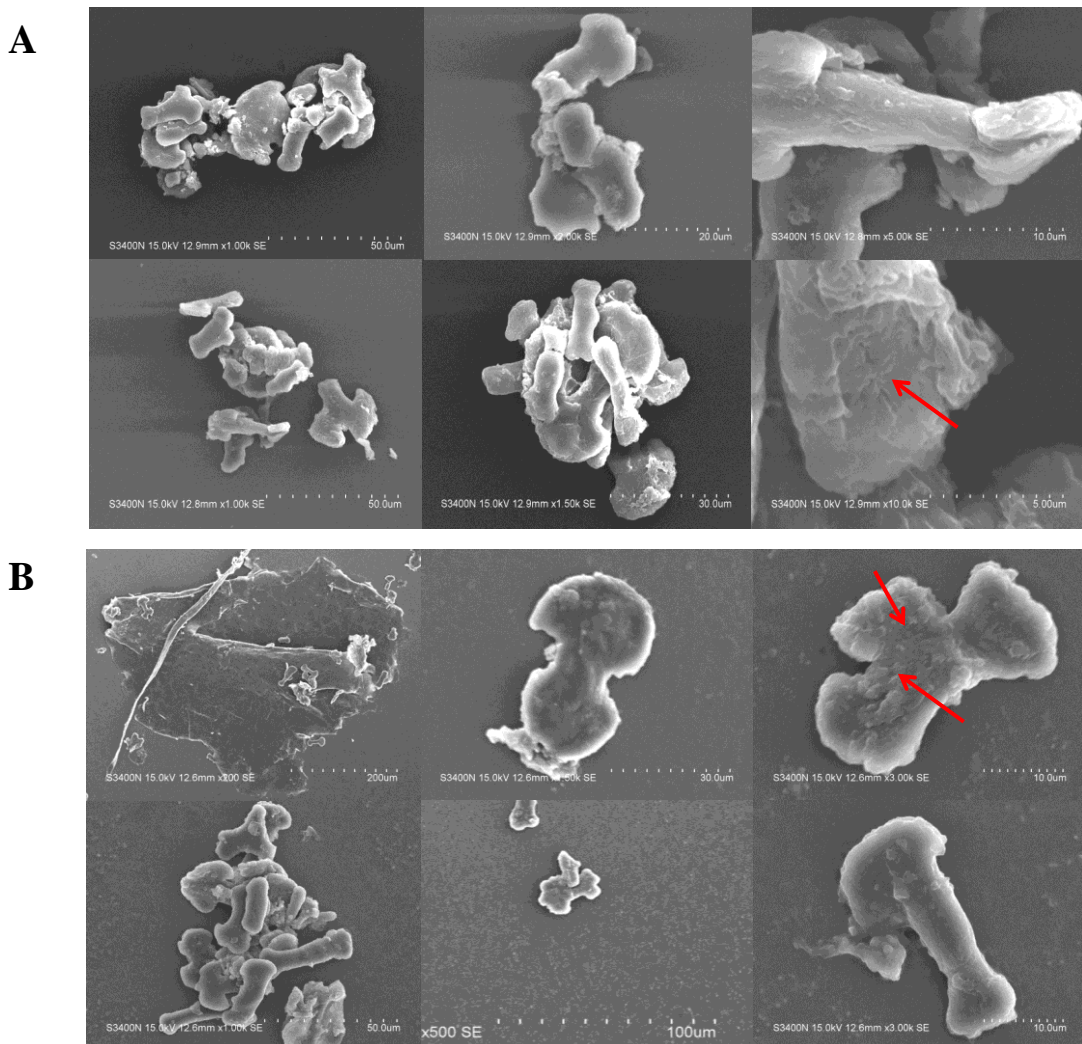
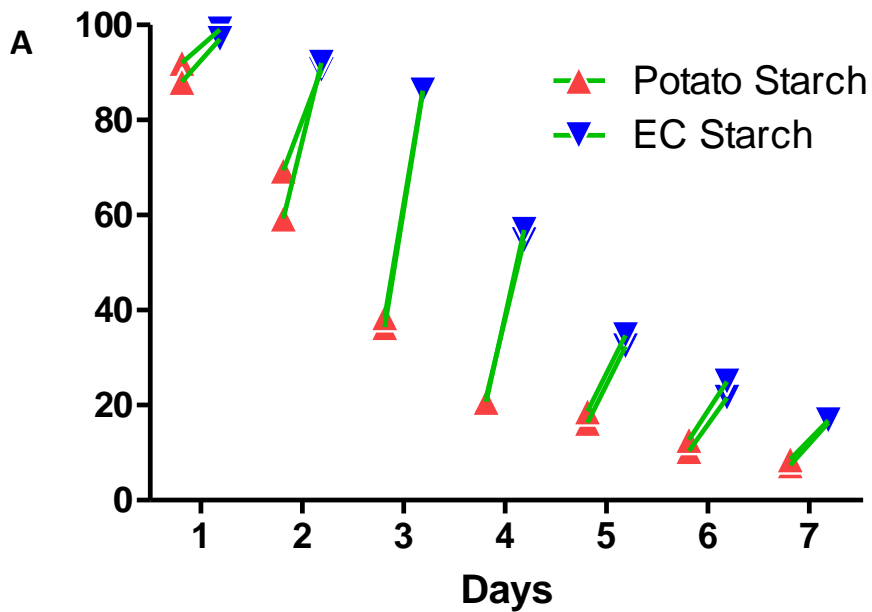


Fig. 5.6. Fragmentation of the starch granules under SEM

A) Fragmented by Rotation, B) Fragmented by High pressure

Lintnerization: Lintnerization is the process of acid treating starch. Acid catalyzes the starch was reported by Nageli in 1874 (Nageli, 1874). Acid digests the amorphous regions of the starch granules and left crystalline regions are called lintners. This has been first reported by Lintner in 1886 (Lintner, 1886). Lintnerization reduces the molecular weight of the starch. Lintnerization was observed with potato as well as with EC starch. The rate of lintnerization is high for the EC starch in comparison to potato starch. EC starch lintners are constant after 10 days with 20% of lintners where as potato starch shown after 7 days with 5%. Later, the lintners at every stage were noted under Scanning Electron Microscope (Fig. 5.7). Effect of lintnerization on various starch granules shows the complexity in the starches (Jayakody & Hoover, 2002).



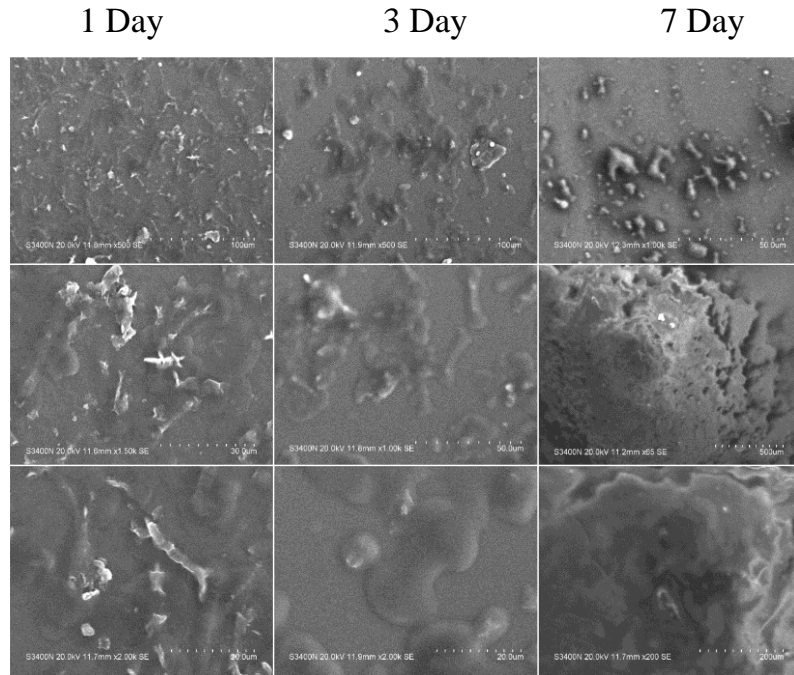


Fig .5.7 Lintnerization of starch

A) Percentage of lintners, B) Lintner images on SEM

Physicochemical properties and Amylose contents: Amylose contents were found to be 45% where normal starch grains from rice, wheat and maize were found to be 20-33%. This implies the high content of the amylose in EC starch. This high content of the amylose was supported by high blue values in comparison to potato (Fig. 5.8). Overall indicates the presence of high amylose content to the potato. Iodine forms complex with amylose of the starch granules give a blue colour to the molecule it can also form red to purple colour complex with amylopectin (Colin & Claubry, 1814; Foster et al., 1943; Stromeyer, 1815). These results suggest the structures possessing intermediate between amylopectin and amylose.

Physicochemical properties like solubility, swelling index and turbidity were analyzed in comparison to reference potato starch (Fig.5.9). The swelling index decreased and not completely solubilized even after 100°C with EC starch whereas potato starch

completely solubilized above 80°C. This results in the complexity of the molecular structure of EC starch granule. The starch granules with high amylose contents swell to a low degree since starch granules are more rigid and resistant to digestibility (Sandhya Rani & Bhattacharaya, 1989). EC starch shows lower turbidity value than potato starch. Lower turbidity due to reluctant nature settled starch granules (Lan et al., 2008).

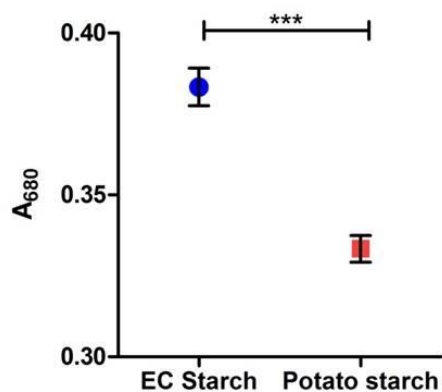


Fig. 5.8. Blue value of Iodine starch complex

Crystallinity and functional group analysis: Starch granules isolated were analyzed for crystallinity and variations in the functional group present. Crystal structure released the presence of the intermediate composition of Amylose and amylopectin. EC starch shows the C-type crystallinity whereas potato with A-type. Starch have 15-45% of crystallinity varies botanical sources. Thus, starch granules displays “Maltose cross” under polarized light (D J Gallant et al., 1992; Zobel, 1988). It has been known that starch granules from the EC and other Euphorbiaceae are devoid of “Maltose cross” (Spilatro & Mehlberg, 1985). Therefore, the crystallinity of the starch granules exhibits intermediate i.e C-type crystallinity. XRD patterns suggest the C-type crystallinity of the EC starch (Fig. 5.10) with peaks at 13°, 15°, 23° and 25°. Infra-Red spectra of EC starch and potato starch are look similar which suggest the absence of impurities (Fig. 5.10).

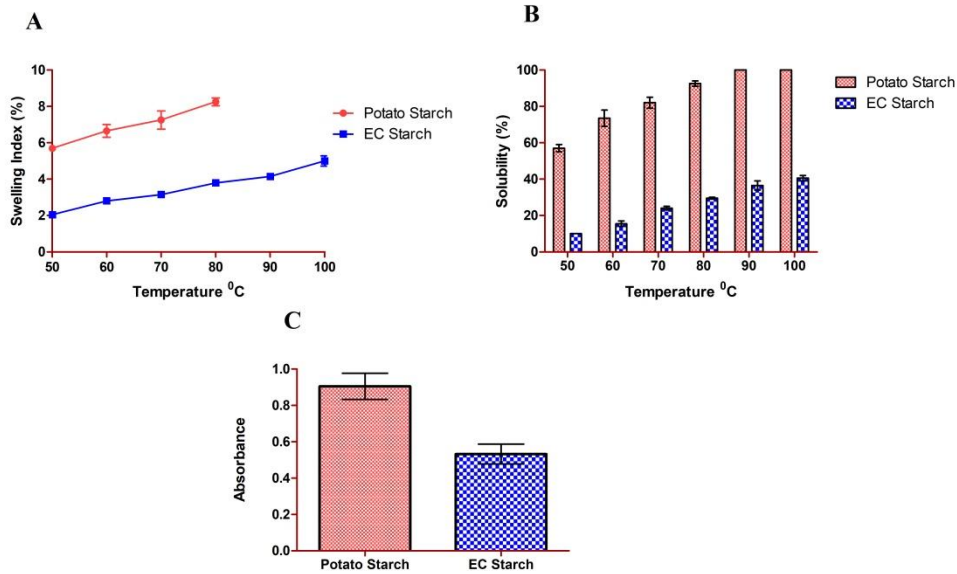


Fig. 5.9 Physicochemical properties of starches
A) Swelling index, B) Solubility, C) Turbidity

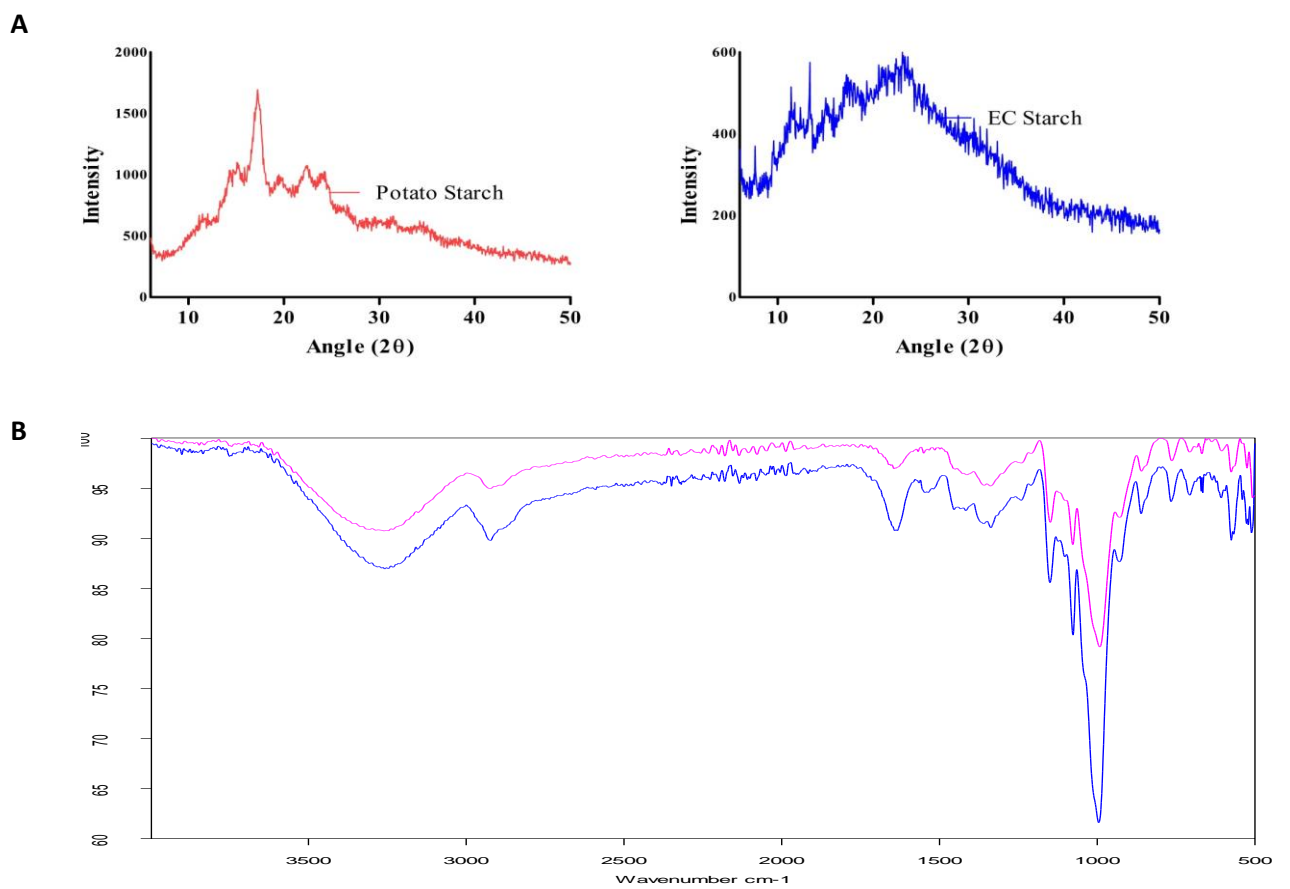


Fig. 5.10. A) X-ray diffractions of potato and EC starch;
B) IR spectrum of the starches potato (pink), EC starch (blue)

Thermal and retrogradation properties: Differential Scanning Calorimetry was used to study the gelatinization properties of starch granules (Table 5.1). When a molecule heated, thus results in the loss of weight and absorption or release of the energy occurs as a result of phase transition. Starch granules with 5% water and 5% NaCl result in the endothermic peaks at 58.89°C and 61.22°C respectively. Change in the enthalpy was 24 and 21 J. Gelatinization temperatures and change in the enthalpy closely resembles the corn starch (Jane et al., 1999). The apparent amylose content of starch correlated to onset and end set of temperatures. High amylose content lowers the onset temperature. EC starch granules do not exhibit any retrogradation properties with 5% water. Realignment of heat disrupted amylose and amylopectin called retrogradation (Wang et al., 2015).

Table 5.1 Thermal properties of EC starch

| Sample | Thermal | | | ΔH |
|----------------------|---------|---------|-------|------------|
| | Onset | End set | Peak | |
| EC Starch | 0 | 0 | 0 | 0 |
| EC Starch + 5% Water | 50.47 | 69.41 | 58.89 | 24.176 |
| EC Starch + 5% NaCl | 52.77 | 70.54 | 61.22 | 21.593 |

Conclusions

Starch granules from *E. caducifolia* latex are flat dumb bell shape. Laser Confocal microscope revealed the birefractive nature and growth rings of the starch granules as like potato starch. Granules grow from round or ovoid (young) to full mature dumb bell or discoid shapes. Blue values of the starch disclose the high contents (45%) of amylose with C-type crystallinity. High amylose presence decreased the solubility, swelling ability, turbidity, and lintnerization. Typical starch spectrum was identified with IR spectra with devoid of retrogradation properties.

Summary

Euphorbia caducifolia Haines, a folk medicinal plant was selected to evaluate potentiality of the plant scientifically. Established pharmacognosy of the plant with phytochemical studies including traditional morphological characters. *Euphorbia caducifolia* is an arid semi-arid plant with paracytic stomata, vein islets- 5-7, vein termination-7-10. Phytochemical screening reveals the presence of acidic polysaccharides, terpenoids, phenolics, and flavonoids. Phylloclade extract can be exhibited strong free radical scavenging activity and latex of the plant cause irritation due to terpenoids and Haemolytic in nature.

An acidic polysaccharide was isolated and found to be brown yellow colour. ECP exhibited the free radical-scavenging activity (DPPH), Nitric oxide scavenging activity and Reducing power as antioxidant agent and also act as Emulsifying agent for pharmaceutical purpose. *InVitro* studies of the Polysaccharide on cell line RAW 267.4, showed the effective production of inflammatory cytokines IL-6 and TNF- α , and cytotoxicity assay showed the significant Cell proliferation. Production of Reactive Oxygen Species by ECP may be responsible for the cell proliferation by inducing the inflammatory cytokines (IL-6 & TNF- α).

Euphorbia caducifolia latex contains Galactose binding lectin which was isolated using two matrices, where CLLSG matrix recovered more lectin than common CLGG. Lectin is thermally stable and active at pH 6-8, and predominant in β structure. *Euphorbia caducifolia* lectins are mitogenic and show Anti-tumour activity.

Euphorbia caducifolia latex contains Flat dumbbell shaped starch granules. Starch granules showed Birefringence with polarized light as a characteristic feature of the starch. EC starches are intermediate of amylose and amylopectin contents. Thus results in C-type crystallinity in XRD patterns. *Euphorbia caducifolia* starch granules contain rodlike microfibrils arranged in radial orientation and do not exhibit any retrogradation properties.

This study suggests that *Euphorbia caducifolia* is proven to be a medicinally potential plant with strong antioxidant extracts, immunomodulatory polysaccharides, mitogenic and antitumor lectin, and novel structured starch granules for the preparation of biocomposites.

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Isolation and Characterization of Certain Biomolecules from *Euphorbia caducifolia* Haines

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Phytochemical screening and Antioxidant activity of *Euphorbia caducifolia* extracts

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Abstract

Purpose: This study was conducted to screen the phytochemical compounds present in the phylloclade of *Euphorbia caducifolia* Haines and to measure the ability of free radical scavenging activity as an antioxidant. **Methods:** The collected plant phylloclades were properly dried, powdered and extracted with hexane, methanol, and water. Extracts were tested for phytochemicals, total phenolic and flavonoid content, and antioxidant activities. **Results:** Carbohydrates, phenols, flavonoids, tannins, glycosides were present in water extract and terpenoids along with phenols and flavonoids in methanol. Hexane contains a maximum of terpenoids and reducing sugars. Water extract showed maximum DPPH[•] free radical (86±1.5%) and nitric oxide scavenging activities (39.23±4.7%) followed by methanol and hexane due to the presence of phenolic (68.740 ± 0.025 mg of GAE/g) and flavonoid contents (50.04±0.125 mg of RE/g). **Conclusion:** This study suggests that *Euphorbia caducifolia* is a promising plant for the therapeutic purposes due to its antioxidant activity of water extract and high contents of terpenoids in methanol and water extracts. A further study of isolation and characterization of active compounds from the plant is necessary to treat various diseases and for multiple applications.

Keywords: *Euphorbia caducifolia*, Free radical scavenging activity, Phylloclade, Phenolic content, Flavonoids.

Introduction

Plants are a rich source of useful drugs since ancient times. In India, plants have been used for many therapeutic applications by our traditional system of medicines such as Ayurveda, Unani, Homeopathy, and Siddha. Traditional medicine system gave birth to many active compounds in early years of natural chemistry, but reports of natural products from different sources has been decreased rapidly in last decade and looking for new active compounds [1][2]. According to National Medicinal Plant Board (NMPB), a maximum number of plants using in the folk medicine compared to other documented medicine systems [3]. Now it needs to explore new natural compounds from potential folk medicinal plants and ethnomedicinal plants for better therapeutic applications.

All living organisms contain an antioxidant defensive mechanism to counter the free radicals, reactive oxygen species (ROS), reactive nitrogen species (RNS) and other oxidants produced as byproducts of the metabolism. Most of the diseases caused by overproduction and reactive mechanism of free radicals [4][5]. Many medicinal plants, spices, aromatic plants using as food supplements and medicines are natural antioxidants [6]. It has been an upsurge of interest for potential antioxidants from plants sources.

Euphorbia caducifolia Haines (leafless milk hedge), a folk medicinal plant belongs to Euphorbiaceae (Castor family). It is a latex-producing plant and found in rocky areas of the tropical region [7]. The plant latex is used for wound healing, leucoderma, and skin eruptions [8]. Latex and root of the plant used to treat cancer [9]. Root also used for snakebite in Maharashtra [10]. Leaf extracts of the plant were reported for its antimicrobial activity [11]. Apart, there are not many reports on phylloclade of the plant which are a rich source of biofuel and other compounds [12]. Therefore, the present study was conducted to reveal the phytochemical compounds exist in the phylloclade and to explore the ability of free radical scavenging activity as an antioxidant.

Materials and Methods

Collection of Plant material

The aboveground parts of *Euphorbia caducifolia* plant were collected from Bhongiri rocky areas, Yadadri district of Telangana (Fig. 1) and shade dried until the material obtains a constant weight.

Preparation of Extracts

The dried plant material was made into powder and used for extraction. The powder was extracted with hexane, methanol, and water in successive order. The extracts were concentrated at temperature 40°C under vacuum and used for further analysis.

Phytochemical screening

Extracts were analyzed for alkaloids, carbohydrates, glycosides, flavonoids, phenolic compounds, quinones, tannins, saponins, and terpenoids [13].

Estimation of Total Phenolic Content (TPC)

The TPC of the extracts was determined using the Folin-Ciocalteu colorimetric method. 1 ml of extract was mixed with 1 ml of 1 N Folin's reagents and well shaken. After 5 min, 1 ml of 10% sodium carbonate was added and incubated for 1 h at room temperature. The absorbance was taken after incubation at 760 nm against blank. Total phenolic content was given as Gallic acid equivalents (GAE) [14].

Estimation of Flavonoids

0.1 ml of extract was added to 0.3 ml 5% sodium nitrite and mixed with 3 ml of 1% aluminum chloride. After 5 min incubation, 2 ml of 1 M NaOH was added and made up the volume to 10 ml with water. The solution was mixed well and measured absorbance at 510 nm against blank. The flavonoid content was calculated using rutin as standard [14].

Antioxidant Activity**DPPH Scavenging Assay**

DPPH (2,2-diphenyl 1-picrylhydrazyl) free radical used to determine the scavenging activity of extracts [15]. DPPH (1 ml, 0.1mM) added to various concentrations of extracts and BHT (3 ml) and 0.5 h incubated in the dark and measured absorbance at 517 nm against the blank (n=3). The activity was calculated by the following equation [16]:

$$\text{Percentage inhibition} = \frac{(\text{Abs control} - \text{Abs sample})}{\text{Abs control}} \times 100$$

Nitric oxide radical scavenging Activity

Nitric oxide (NO) produced by Sodium Nitroprusside at pH 7.2-7.4 was determined by Griess reagent (Griess diazotization reaction). The reaction mixture (3 ml) was incubated at 25°C for 2 hr containing Sodium Nitroprusside (10 mM) and extracts with different concentrations. An Aliquot (0.5 ml) was added to Griess reagent (0.5 ml) and measured at 546nm (n=3). The activity was calculated by comparing control and test sample. Ascorbic acid used as reference compound [17].

Results and Discussion**Phytochemical analysis**

Phytochemical studies of extracts showed the presence of various classes of metabolites. *E. caducifolia* showed the high quantity of terpenoids in methanol and hexane extracts as *Euphorbia* species are predominant with cyclic and acyclic terpenoids [18]. Test for the presence of alkaloids shown no significant colouration. Water extract was rich in carbohydrates, glycosides, tannins, phenols, flavonoids and saponins. Further experiments revealed that the presence of acidic polysaccharides and saponin glycosides. Methanol extract contained reducing sugars, glycosides and some phenolic compounds whereas in hexane extract showed positive for reducing sugars and glycosides along with terpenoids (Table 1).

Estimation of total phenolic and flavonoid content

Phenolic compounds such as flavonoids and tannins are excellent free radical scavengers and protect the plant from different types of oxidative damages [19]. Water extract showed the high phenolic and flavonoid contents whereas no significant amounts observed in hexane extract. Methanol extract showed good amounts of flavonoids compared to phenolics (Table 2). Antioxidant activity of the extracts might be due to the presence of phenols and flavonoids present [20][21].

Antioxidant activity

Free radicals are the most reactive molecules and damage the cells. Aerobic organisms protect themselves from free radical by the defensive antioxidant mechanism, and it is necessary to provide external antioxidants in failure defensive system [22]. The extracts were analyzed for their antioxidant capacity to use as a free radical scavenger. Most of the members of the castor family are a good source of natural antioxidants such as *Euphorbia hirta* [14], *E. heterophylla* [23]. *E. caducifolia* extracts also exhibited antioxidant activity. The potentiality of the extract to scavenge the free radical produced was considered for antioxidant activity (Fig. 2). DPPH and NO are the two radicals used for the experiments. All extracts were showed activity on concentration dependency. Water extract exhibited more scavenging activity towards DPPH (86±1.5%) as well NO (39.23±4.7%) to hexane extract. More free scavenging activity of water extract might be due to the presence of the high phenolic and

flavonoid contents. Methanol extract also exhibited similar (DPPH-77.81±1.3% and NOSA-36.03±2.71%) to water but dropped activity frequency at high concentration of the extract. Higher antioxidant activity was observed at higher concentrations (Fig. 3). Extracts showed less ability to scavenge the NO free radical as compared to DPPH.

Conclusion

Euphorbia caducifolia is a promising plant for the various therapeutic applications with rich contents of terpenoids, polysaccharides, tannins, flavonoids and phenolic compounds. The presence of significant amount of total phenolics and flavonoids contents in water extract might be responsible for strong antioxidant activity as compared with methanol and hexane extract. Hexane and methanol extract can be used for the terpenoid sources and water extract for phenolics and flavonoids. High contents of terpenoids in methanol and hexane extracts are useful in various applications. A further study of isolation and characterization of bioactive molecules from the plant is required for therapeutic and multiple applications.

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Table 1: Phytochemical Screening of *E. caducifolia* extracts

| Fraction | Water | Methanol | Hexane |
|---------------|-------|----------|--------|
| Alkaloids | -- | -- | -- |
| Carbohydrates | ++ | ++ | ++ |
| Glycosides | ++ | ++ | ++ |
| Flavanoids | ++ | + | -- |
| Phenolics | ++ | + | -- |
| Tannins | ++ | -- | -- |
| Saponins | ++ | -- | -- |
| Quinones | -- | ++ | ++ |
| Terpenoids | -- | ++ | ++ |

Table 2: Total Phenolic and Flavonoid contents in *E. caducifolia* plant extracts

| Fraction | Phenolics | Flavonoids |
|----------|------------------------|-----------------------|
| | mg of GAE/g of Extract | mg of RE/g of Extract |
| Water | 68.740 ± 0.025 | 50.04 ± 0.125 |
| Methanol | 00.926 ± 0.06 | 13.534 ± 0.754 |
| Hexane | 00.080 ± 0.014 | 00.004 ± 0.0002 |

(n=3, Mean ± SD)

Figure 1: Distribution of *Euphorbia caducifolia* in the wild.

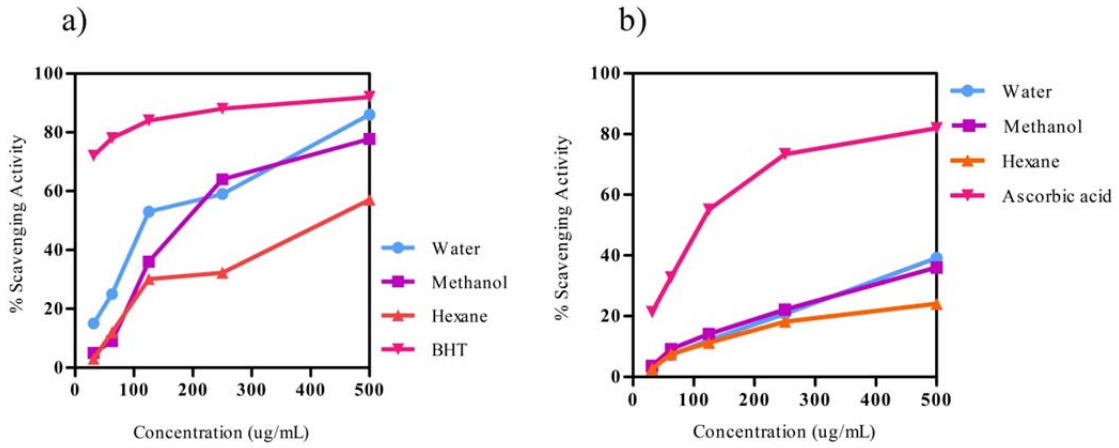


Figure 2: Antioxidant activity of *E. caducifolia* plant extracts a) DPPH Scavenging activity, b) Nitric Oxide Scavenging activity

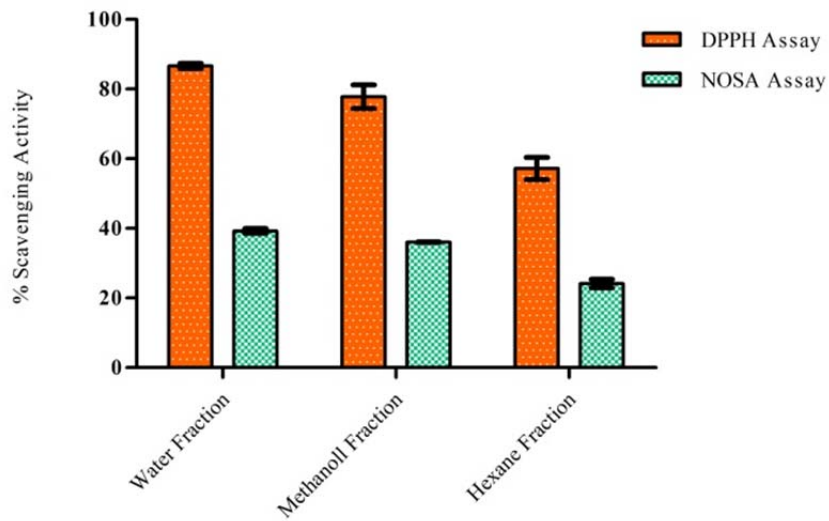


Figure 3: Antioxidant activity at maximum concentration (500 ug/ml) of *E. caducifolia* plant extracts



PREPARATION OF HERBARIUM SPECIMEN FOR PLANT IDENTIFICATION AND VOUCHER NUMBER

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Abstract: A herbarium is a storehouse of plant specimens which are collected, dried and mounted on handmade paper sheets. They will be arranged in plant families recognized system of classification and kept in pigeon holes of steel or wooden cup boards and maintained carefully for current and future studies. It is a reference material for naming, identification and classification of the plants. The herbaria are indexed with unique codes in the “*Index Herbariorum*” presently assigned and maintained by New York Botanical Garden herbarium. Also the non-vascular plant specimens (algae, certain fungi, lichens, bryophytes and certain pteridophytes) can be fixed in FAA fixative and bring to the herbarium for identification and voucher specimen number. The article also informed the merits of the “Plant Specimen Preparation Kit” from Nihon Vogue-Sha, Japan for use of specimens’ preparation both vascular and non-vascular plants where there is no involvement of chemicals and harsh environmental conditions. Since the identification and voucher specimen number is essential in these days for research as well as for publications, here we are presenting useful information.

Key words: Herbarium; Plant identification; Voucher specimen number; Voucher material number; Plant Specimen Preparation Kit - Nihon Vogue-Sha

Introduction

The “Herbarium” has two meanings; one is a repository or storehouse of collected plant specimens and second is a plant specimen according to accepted international standards. If the term “Herbarium” followed by a code assigned by of “*Index Herbariorum* (IH)” authorities in the parenthesis (Thiers, 2016), it shall be consider as repository and the plural is “herbaria”. If the term “Herbarium” is not followed by the code in parenthesis we shall be consider it as a “plant specimen” and plural is “herbarium specimens”. It is believed that the first established in 1570 in Bologna, Italy, by Luca Ghini. It is now estimated that there are nearly 350,000,000 specimens that are documented from the world’s vegetation. These herbarium specimens are available at approximately 3,000 herbaria in the world, with approximately 12,000 associated curators and plant specialists (Thiers, 2016). It shall be noted that those collections that are permanent scientific repositories are included in IH and the new registrants must demonstrate that their collection is accessible to scientists, and is actively managed. Each institution is assigned a permanent unique identifier in the form of a one to eight letter code, a practice that dates from the founding of IH in 1935. We shall note that the first six editions of IH were print copies published by the International Association for Plant Taxonomy in the Netherlands (1952–1974), subsequently Dr. Patricia Holmgren, then Director of the New York Botanical Garden, served as co-editor of edition 6, and subsequently became the senior editor of IH. She oversaw the compilation of print copies volumes 7 and 8, and Dr. Noel Holmgren, scientist on the NYBG staff, oversaw the development of the IH database, which became available from 1997 in on-line. From September 2008 on words Dr. Barbara M. Thiers, Director of the NYBG Herbarium, became the editor of IH.

IH is a detailed directory of public herbaria of the world and the staff and plant specialists associated with them. They are authorized to identifying the plant and provide “Voucher Specimen

Number”. Each herbarium in the Index is assigned an official code that is used as a standard reference for citation. It consists of the herbarium code, institution, city, state, staff member, correspondent, and research specialty. The data is maintained and available at New York Botanical Garden Herbarium at <http://sweetgum.nybg.org/science/ih/> (Thiers, 2016)

With the advent of Information Technology, new techniques are adopted now for the herbaria include the micro-herbaria in the form of photographs and the electronic herbaria in the form of digitized databases. The Voucher herbarium specimen is pressed plant sample deposited for future reference and it will be verify the identity of the specific plant used in a study. It should be deposited in a recognized herbarium indexed in “*Index Herbariorum*”. The herbaria also keep the voucher materials such as seeds, wood sections, pollen, micro slides, fluid preserved flowers, fruits etc. Herbaria keep all categories of plants specified by International Code of Nomenclature of algae, fungi and plants (ICN) (McNeill et al, 2012; Anonymous, 1996). Although the herbarium and specimen preparations are mostly confines to seed plants, there are specific techniques practicing to algae, fungi, and plants (Anonymous, 1996). These specimens are in liquid preservation for non-vascular plants, bryophytes etc. (Anonymous, 1996). As we may aware that now we are extracting bioactive compounds from these organisms and several of the international journals and periodicals insisting for the voucher specimen number for the specimens.

In the recent days it is practicing that a voucher specimen must be deposited in a recognized herbarium committed to long-term maintenance. It was started from knowledge exchange through ethnic groups by oral tradition and then in the documented form. Herbaria conserving identified and authenticated plants for future correspondence play an important role in this regard and it declared different auxiliary aspects of herbaria for phytomedicinal research (Funk, 2003; Neisbitt, 2014; Ahmed and Hasan, 2016). Also the vouchers are crucial in authenticating the taxonomy of an organism, as a tool for identifying localities of the taxon, and for additional taxonomic, genetic, ecological, and/or environmental research (Eisenman et al., 2012). Although collaboration between botanists and chemists is extremely important for the initial collection and identification of plant materials, a need also exists for documentation at later stages of research, such preparation for publications. Botanists collaborating with chemists need to inform and educate colleagues about proper vouchering methods and providing detailed voucher information in all publications. Making vouchers is extremely important, but sharing the information about where the vouchers are stored is also critical. Voucher information is being increasingly required by many journals publishing articles on plant-based chemistry research.

Since the identification and voucher specimen number is essential in these days for research as well as for publications such as plant natural products, pharmaceutical sciences, medicinal botany and allied subjects (WHO, 1998; Hildreth et al, 2007), the information on the herbarium specimens preparation for the plant identification and voucher specimen number we would be presenting useful information on these aspects.

Preparation of herbarium specimens:

The preparation of the specimens was followed by methods stated in Lawrence (1967), Jones and Luchinger (1987), Anonymous (1996) and Manilal and Kumar (1998)

i) Collection: The specimen material (plants) which you are interested in should be collected as whole (if they are herbs) or part of plant along with flowers and fruits/carpels (Figures 1 to 6). Before putting your specimens in the collection bag you should carefully remove all the insects, spider-webs and foreign bodies attached to your specimens. Then the specimens mounted in 42 cm x 29 cm (16 ½” x 11 ½”) size blotting paper. If the plant specimen larger than the specimen blotting paper they can mounted in V or N or M shape to accommodate the entire plant material in above size. Also the leaves mounted in dorsiventral position. The half-size of the regular newspaper can fit this size and economical! Put the mounted specimens in between cardboards/wooden-frames (Figures 7 to 9).

A well-designed field notebook has numbered sheets in printed form with standard pro-forma for entering field data. Usually a field note book has 100 leaves and the pages are serially numbered (called field numbers) to be suffixed to the collector's name, when cited (Figure 10).

ii) Poisoning and drying the specimen: Poisoning kills the plants and prevents the formation of abscission layer and thereby the leaves, flowers and fruits will be intact with the specimen (twig) will not be getting detached from the plant. The poisoning is generally done by dipping the whole plant in a saturated solution of mercuric chloride in ethyl alcohol (usually 20 gm in a litre of alcohol). The plant is again put between the blotters in the presser till it gets completely dried. Mercuric chloride is corrosive for metals, and hence enamel trays and disposable gloves are used. Lauryl Pentachloro-phenate (LPCP) is also used (3.75% in white spirit) for poisoning the specimens. It is safer than mercuric chloride and leaves the plant features more intact. The solution can also be applied to mounted specimens by spraying. Then, the specimens are spread out for pressing and drying. It is important that the plants are put under sufficient pressure; otherwise more time will be required to achieve a good desiccation, besides they could be damaged by dampness and moulds. Every specimen in the press must be linked with the data in the field note book (Figure 10). Detailed notes should be entered in the field note book at the time of collection in the field itself. The best one can do is to use a tag for each specimen. Bulky plant parts can directly be placed in contact with corrugated material to speed up drying. Instead of blotting paper we can use newspapers, which are cheap and readily available. Once a specimen has become dry and stiff, it is ready for mounting.

Cacti, succulent and tuberous/bulbose plants can be cut their stems, and leaves into transverse and longitudinal sections to observe their morphology. These plants should be poisoned longer-time to completely kill the specimens. Also they can also the softening of the cactus can be aided by the immersion in boiling water for half a minute, taking care of avoiding the immersion of the flowers. Instead of boiling, one can employ dilute acetic acid or strong alcohol or formalin (1.5 parts formalin, 1 part water). Some plants have tubers or bulbs and they must be treated before drying.

iii) Mounting and labelling: The dried plant specimens are now ready for mounting on herbarium sheets. Fixing the processed plant specimen on herbarium sheet is called *mounting*. A standard herbarium sheet is 28 cm (breadth) x 42 cm (length) and usually made up of heavy long-lasting white handmade paper or thick sheet (Figure 11). The sheet is usually stiff and flexible so as to prevent damage during the handling of mounted specimens. The common technique is pasting specimens to sheet with natural glue (usually Gum Arabic). Small quantity of copper sulphate or thymol crystals or may be added to the glue as insect repellent. It is advised to have a paper bag/pouch attach to the herbarium sheet to keep any seed/fragments detached from the specimens (Figure 11). Now the herbarium specimen sheet pasted with a label usually at the right side bottom corner.

The herbarium specimen sheet is now ready for deposition/reposition in the one of the indexed herbaria for the plant identification as well as to acquire a "Voucher Specimen" number.

The non-vascular plant specimens (algae, certain fungi, lichens, bryophytes and certain pteridophytes) either dried or can be fixed in FAA fixative [Formalin-Acetic Acid-Alcohol in the ratio of 50% ~ 70% Ethanol (90 ml), Glacial Acetic Acid (99.6%) 5 ml and Formaldehyde (38%) 5 ml] and bring to the herbarium (Lawrence, 1967; Anonymous, 1996). Further if you have any inquiry can contact the Curator or Prof-in-charge of the herbarium.



Figure 1*



Figure 2



Figure 3



Figure 4



Figure 5



Figure 6*: Insert the plant between the newspaper sheets

Figures 1-5: Collect the whole plant with the roots (if possible) or branches with flower & fruits



Figure 7*: Write the information about the plant (e.g., location, date, collector, habitat, etc.)

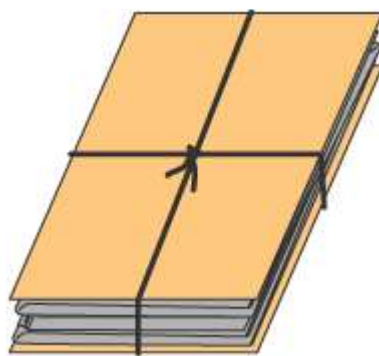


Figure 8*



Figure 9

Figures 8* & 9: Bundle the plants (in newspapers) with the field boards/wooden-frame

UNIVERSITY OF HYDERABAD
Hyderabad-500 134.

7
FLORA OF University of Hyderabad Campus

6 COLL. NO. DATE 20-8-85

5 FAMILY : Euphorbiaceae

7 GENUS & SPECIES : Acalypha indica L.
Fl. of Madras II 930

6 LOCALITY : Life Sciences Experimental Garden

HABITAT : Macrophyte Wet areas/shady places

5 ♂ Br. Ebel ⊕ ♂ P₄ A₀ G₀

♀ Br. Ebel ⊕ ♀ P₄ A₀ G₁₃

4 DISTRIBUTION : Throughout campus

3 DESCRIPTION : Herb. with mosaic leaf arrangement.
Fl. of both ♂ ♀ in axillary spikes. ♂ upper most. Ls long
petioled, ovate or rhombic acute upto 7.5 cm

2 VERN. NAME & USE : Munipinda (Tel.)

1 PHOTO DATA :

0 COLLECTOR : K. Seshagirinao

CM

| | |
|---|---|
| E | F |
| 7 | 7 |
| ○ | ○ |

Figure 10: A page from field note book



Figure 11: Mounted herbarium specimen with number tag and seed/fragment pouch

Recently the Nihon Vogue-sha, Japanese company is offering a new approach/technique to botanical plant preservation for herbarium specimens. The "Plant Specimen Preparation Kit" utilizes a simple and reliable vacuum pump and a unique air press to ensure that plant specimens are beautifully preserved with their natural colors retained (Anonymous, 2016).

This method also offers an improvement in preserving plant specimens by using special "stock" bags that create an airtight condition for long periods. These stock bags are the optimal way to display natural looking plant specimens and are ideal to protect specimens loaned to other institutions or individual researchers. These bags are essential for long lasting specimen preservation (Figures 12 to 14). You watch the drying process in a video at <https://www.oshibana.com/herbarium/en/index.php/making-a-specimen/drying-the-plant> . The Plant Specimen Preparation Kit makes it easy to solve the problems that have traditionally been part of creating conventional pressed flowers and plants.

- It's easy to do and takes less time
- The beautiful plant colors are saved/preserved
- A special bag prevents damage so you'll feel a great satisfaction with the beautiful results of your efforts.



Figure 12*: Nihon Vogue-sha, Japanese company's "Plant Specimen Preparation Kit"



Figure 13*: Peanut plant herbarium specimen

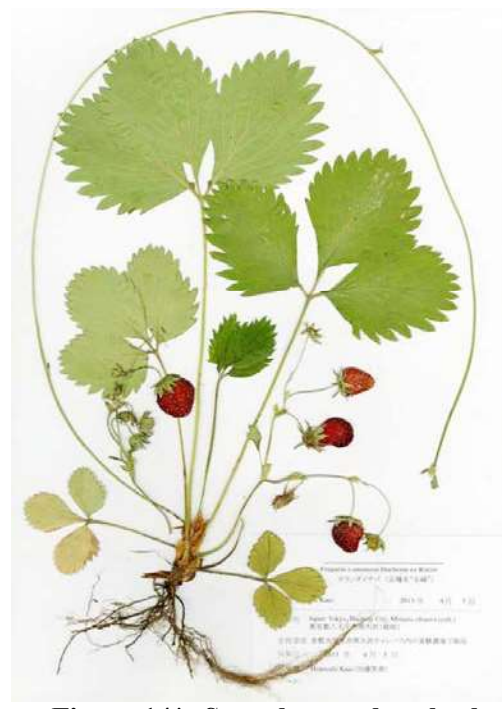


Figure 14*: Strawberry plant herbarium specimen

*The figures & information are courtesy of Nihon-Vogue-sha, Tokyo, Japan (<https://www.oshibana.com/herbarium/en/index.php>)

Conclusions

Voucher specimens provide a permanent, physical record and form the foundation on which all natural product research stands (Funk, 2003; Culley, 2013). The annotations and recommendations in this regard are noteworthy (Perkins, 2016). An additional voucher specimen is advisable, and can serve back-ups in case damage to or loss of the main voucher specimen. The preparation and proper storage of vouchers can also include provide plant tissue for DNA analysis. In the recent the research journals now require that DNA and amino acid sequences intended for publication be submitted to a sequence database, such as GenBank before being published, but many journals do not yet require that vouchers be made for the plants from which these sequences were isolated (Pleijel et al., 2008). It is now indeed very much critical for dioecious plants where the male and female plants analysis followed by separate voucher specimen number for each sex plant. The additional conclusion on the usage of the “Plant Specimen Preparation Kit” from Nihon Vogue-Sha, Japan would be more promising for specimens’ preparation both vascular and non-vascular plants where there is no involvement of chemicals and harsh environmental conditions. These specimens can be recommending using in the analysis of chemical, biochemical and molecular biology.

Competing interests

The authors declare that they have no competing interests.

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